

Sample Question Paper - 35
Science (086)
Class- X, Session: 2021-22
TERM II

Time allowed : 2 hours

Maximum marks : 40

General Instructions :

- (i) All questions are compulsory.
- (ii) The question paper has three sections and 15 questions. All questions are compulsory.
- (iii) Section–A has 7 questions of 2 marks each; Section–B has 6 questions of 3 marks each; and Section–C has 2 case based questions of 4 marks each.
- (iv) Internal choices have been provided in some questions. A student has to attempt only one of the alternatives in such questions.

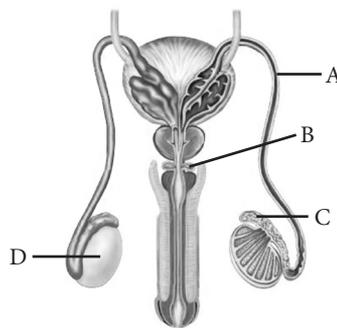
SECTION - A

- 1. What is meant by periodicity of properties of elements? Why are the properties of elements placed in the same group of the periodic table similar?
- 2. Rewrite the following statements after correction, giving reasons.
 - (i) Elements in the same period have equal valency.
 - (ii) The metallic character of elements in a period increases gradually on moving from left to right.

OR

In the periodic table, how does the tendency of atoms to lose electrons change on going from

- (i) left to right across a period?
 - (ii) top to bottom in a group?
3. Identify labelled part A- D and write their functions.



- 4. How did Mendel interpret his result to show that traits may be dominant or recessive? Describe briefly.
- 5. What will happen if there is no greenhouse effect?

OR

Why the flow of energy in food chain is unidirectional?

- 6. (i) LPG (liquefied petroleum gas) is used as a fuel for cooking in homes. Why is LPG stored as a liquid?
- (ii) Which among the following has shortest bond length : C_2H_4 , C_2H_6 , C_2H_2

OR

- (ii) Why carbon does not form the ionic compounds?
(ii) Draw the electron dot structure of ethane.
7. A copper wire has diameter 0.5 mm and resistivity $1.6 \times 10^{-8} \Omega \text{ m}$. Calculate the length of this wire to make it resistance 100 Ω . How much does the resistance change if the diameter is doubled without changing its length?

SECTION - B

8. How reproduction takes place in the following organisms?
(a) *Hydra*
(b) *Planaria*
(c) *Bryophyllum*

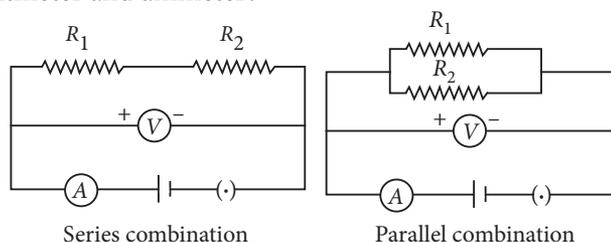
OR

What is the significance of sexual mode of reproduction?

9. Write a note on vegetative propagation.
10. Two circular coils A and B are kept close to each other, of which coil A carries a current. What will you observe in the galvanometer connected across the coil B
(a) if current in the coil A is changed?
(b) if both the coils are moved in the same direction with the same speed?
Give reason to justify your answer in each case.
11. Write Joule's law of heating.

OR

Two resistors are connected first in series and then in parallel across a battery as shown. What effect will it have on the readings of voltmeter and ammeter?



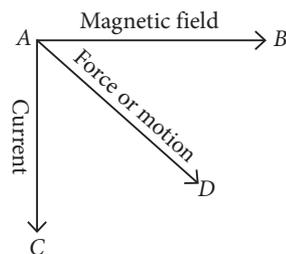
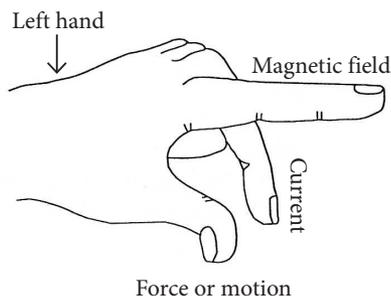
12. In one of his experiments with pea plants Mendel observed that when a pure tall pea plant is crossed with a pure dwarf pea plant, in the first generation, F₁ only tall plants appear.
(a) What happens to the traits of the dwarf plants in this case?
(b) When the F₁ generation plants were self-fertilised, he observed that in the plants of second generation (F₂), both tall plants and dwarf plants were present. Why it happened? Explain briefly.
13. Explain the phenomenon of ozone depletion. What are the factors responsible for it. What are its consequences?

SECTION - C

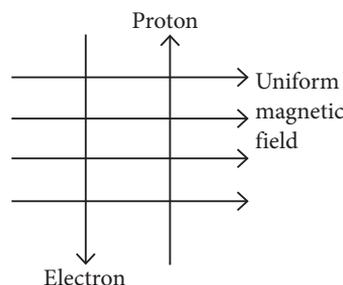
This section has 02 case-based questions (14 and 15). Each case is followed by 03 sub-questions (a, b and c). Parts a and b are compulsory. However, an internal choice has been provided in part c.

14. Andre Marie Ampere suggested that a magnet must exert an equal and opposite force on a current carrying conductor, which was experimentally found to be true. But we know that current is due to charges in motion.

Thus, it is clear that a charge moving in a magnetic field experience a force, except when it is moving in a direction parallel to it. If the direction of motion is not parallel to the direction of magnetic field, the magnitude of force experienced depends on the charge, velocity (v), strength of magnetic field (B), and sine of the angle between v and B . Direction of magnetic force is given by Fleming's left hand rule.



- (a) If an electron is travelling horizontally towards east. A magnetic field in vertically downward direction exerts a force on the electron along which direction?
- (b) If a charged particle is moving along a magnetic field line. Find the magnetic force on the particle.
- (c) A uniform magnetic field exists in the plane of paper pointing from left to right as shown in figure. In the field an electron and a proton are moving as shown. Find the direction of force on electron and proton.



OR

Why does a current-carrying wire move when placed in magnetic field?

15. As neutral atom carbon has electronic configuration $K L$, to gain inert gas configuration it can either 2, 4

donate 4 valence electrons (helium gas configuration) or gain 4 electrons (neon gas configuration), but it cannot do so. To acquire inert gas configuration carbon, can only share its 4 valence electrons with other atoms forming covalent bonds. A covalent bond can be defined as a chemical bond formed between two atoms by mutual sharing of valence electrons so that each atom acquires the stable electronic configuration of the nearest noble gas. The concept of covalent bonds was given by Langmuir and Lewis to explain bonding in non-ionic compounds. The covalent bonds are of three types. If each atom contributes one electron, the covalent bond formed is called a single covalent bond and is represented by a single line ($-$) and if each atom contributes two electrons, the covalent bond formed is called a double bond and is represented by a double line ($=$) and if each atom contributes three electrons, the covalent bond formed is called a triple bond and is represented by a triple line (\equiv).

- (a) Out of the compounds, SO_2 , NH_3 , HCl and O_2 , which compounds do not contain a double bond?
- (b) Given an example of diatomic molecule containing a triple bond.
- (c) Define covalency and give an example of molecule in which all its atoms joined together by double covalent bonds?

OR

Explain briefly the electron dot structure of Cl_2 molecule.

Solution

SCIENCE - 086

Class 10 - Science

1. When elements are arranged in increasing order of their atomic numbers, elements with similar chemical properties are repeated at definite intervals. This is known as periodicity of properties of elements.

Elements placed in the same group of the periodic table have similar properties because they have same number of outermost electrons and hence, show same valency. Thus, they all will form similar type of compounds.

2. (i) Elements in the same group have equal valency because the elements of the same group have same number of valence electrons.

(ii) The metallic character of elements in a period decreases gradually on moving from left to right because the tendency to lose electrons (electropositivity) decreases on moving from left to right in a period.

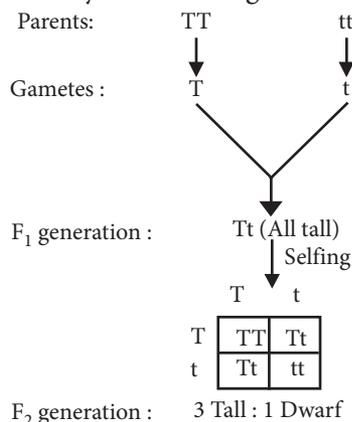
OR

(i) Tendency of atoms to lose electrons decreases from left to right in a period.

(ii) Tendency of atoms to lose electrons increases down the group.

3. In the given figure, part 'A' is vas deferens that carries sperms. Part 'B' is Cowper's glands that secrete fluid which helps in the lubrication of the penis during copulation. Part 'C' is epididymis that stores sperms and part 'D' is testis that produces sperms and testosterone.

4. Mendel crossed the pea plant for two contrasting characters under consideration. The trait that expressed itself in F_1 generation was dominant and the one not expressed in F_1 generation was recessive. He later selfed the plants of F_1 generation and recovered, both parental traits in a definite proportion in F_2 generation. Mendel interpreted his results as, the trait that expressed itself in F_1 was dominant and the one that reappeared in F_2 generation was recessive. It can be demonstrated by the following cross :



5. Greenhouse gases like CO_2 , CH_4 and CFCs are essential for keeping the earth warm and hospitable. They prevent a substantial part of long wave radiations emitted by earth to escape into space, but these gases radiate a part of this energy back to the earth. This phenomenon is called greenhouse effect. Because of greenhouse effect, the mean annual temperature of earth is $15^\circ C$. In its absence, it will fall to $-18^\circ C$.

OR

A food chain consists of series of populations which are related by eating and being eaten. There is a unidirectional flow of energy from sun to producers and subsequently to series of different types of consumers, i.e.,

Solar radiations \rightarrow Producers \rightarrow Herbivores \rightarrow Carnivores

It cannot pass in reverse direction. There is always a decrease in the flow of energy and content with rise in trophic level. As some energy is lost at each step as heat and is also used up in various metabolic activities.

6. (i) LPG is stored as a liquid so that more fuel can be kept in a container. Liquids contain more particles per unit volume than gases. It is also easier to transport liquids than gases.

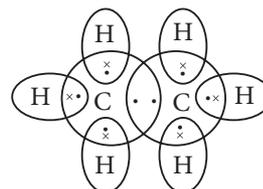
(ii) $HC \equiv CH(C_2H_2)$ has shortest bond length.

OR

(i) For formation of ionic compounds, carbon should either gain four electrons to form C^{4-} ions or should lose four electrons to form C^{4+} ions but formation of both these ions is not possible due to high energy considerations.

(ii) Molecular formula of ethane is C_2H_6 .

Its electron dot structure is :



7. Given; resistivity of copper wire = $1.6 \times 10^{-8} \Omega m$, diameter of wire, $d = 0.5 mm$ and resistance of wire, $R = 100 \Omega$

$$\text{Radius of wire, } r = \frac{d}{2} = \frac{0.5}{2} \text{ mm}$$

$$= 0.25 \text{ mm} = 2.5 \times 10^{-4} \text{ m}$$

Area of cross-section of wire, $A = \pi r^2$

$$\therefore A = 3.14 \times (2.5 \times 10^{-4})^2$$

$$\approx 1.9625 \times 10^{-7} \text{ m}^2$$

$$\approx 1.9 \times 10^{-7} \text{ m}^2$$

$$\text{As, } R = \rho \frac{l}{A}$$

$$\therefore 100 \Omega = \frac{1.6 \times 10^{-8} \Omega \text{ m} \times l}{1.9 \times 10^{-7} \text{ m}^2}$$

$$l \approx 1200 \text{ m}$$

If diameter is doubled ($d' = 2d$), then the area of cross-section of wire will become

$$A' = \pi(r')^2 = \pi\left(\frac{d'}{2}\right)^2 = \pi\left(\frac{2d}{2}\right)^2 = 4A$$

Now $R \propto \frac{l}{A}$, so the resistance will decrease by four times or new resistance will be

$$R' = \frac{R}{4} = \frac{100}{4} = 25 \Omega$$

8. (a) In *Hydra*, a bud develops as an outgrowth due to repeated cell division at one specific site which develops into tiny individuals and detach from the parent body on maturation and become new independent individuals.

(b) *Planaria* can be cut into any number of pieces and each piece grows into a complete organism. This process is known as regeneration.

(c) Buds, produced in the notches along the leaf margin of *Bryophyllum*, fall on the soil and develop into new plants.

OR

Sexual reproduction may be defined as the production of offsprings (new individuals) by the fusion of two gametes (usually one from male parent and the other from female parent) to form a diploid zygote which develops into a mature organism. Gamete formation involves meiosis or reduction division. The gamete mother cell is diploid ($2n$), *i.e.*, it has two sets of chromosomes. This single diploid cell divides by meiosis to form 4 haploid (n) daughter cells. Each daughter cell becomes a gamete, either male or female. Each gamete possesses single set of chromosomes. Thus, this division involves copying of the DNA as well as the cellular apparatus. There is a stage in such nuclear division where crossing over of chromosomes take place. This is very important step which results in a slight different composition of chromosomes in gametes. Fusion of these gametes results in the formation of a slightly different individuals which show variations. The variations which lead to the appearance of such characters which fit to the changing environment result in the survival of the species. Chances of variation, therefore, are much more in sexual mode of reproduction as compared to asexual reproduction. Moreover, chances of the

production of compatible generations are also more in sexual reproduction.

9. In vegetative propagation, a new plant can be obtained from vegetative parts such as stem, root and leaves. Vegetative propagation property can be used in growing more plants by different methods like:

- (i) Layering, *e.g.*, jasmine plant
- (ii) Cutting method, *e.g.*, sugarcane, rose, grapevine
- (iii) Grafting method, *e.g.*, roses, mango, citrus fruit plants, etc.

Vegetative propagation is advantageous as-

- (i) it is a fast, convenient and economical method to produce many plants of commercial importance.
- (ii) plants which do not have capacity to produce seeds can be produced through it.
- (iii) it produces plants identical to its parent plant with respect to various traits.

10. (a) When the amount of current in the coil *A* is changed, an induced current will induce in the coil *B* due to change in magnetic field lines *i.e.*, magnetic flux.

(b) If both the coils are moved in the same direction with the same speed, then there is no net change in magnetic flux. Hence there will be no deflection in the galvanometer.

11. The Joule's law of heating implies that heat produced in a resistor is

- (i) directly proportional to the square of current for a given resistance,
- (ii) directly proportional to resistance for a given current, and
- (iii) directly proportional to the time for which the current flows through the resistor.

$$\text{i.e., } H = I^2 R t$$

OR

There will be no change in the reading of voltmeter because applied voltage remains the same in both the cases but ammeter reading will be less in case of series combination as compared to parallel combination of the same resistors because $R_s > R_p$ and $I \propto \frac{1}{R}$ (for constant applied voltage) as shown in the following circuit diagrams.

