SURFACE TENSION

SYNOPSIS

- The attractive forces between the molecules of a substance are called cohesive forces
- The attractive forces between the molecules of different substances are called adhesive forces.
- The maximum distance upto which the cohesive force between two molecules exists is called the molecular range and of the order 10⁻⁹m.
- An imaginary sphere drawn around a molecule with a radius of molecular range is called the sphere of influence of that molecule.
- The force per unit length across any imaginary line drawn on the surface of a liquid and acts perpendicular to it is known as surface Tension

$$\left(T = \frac{F}{l}\right)$$
 Unit - N/m or dyne /cm.

- Surface tension is due to cohesion between the molecules of a liquid.
- Surace tension is a molecular phenomenon.
- Surface tension is independent of surface area.
- Factors which influence surface tension of a liquid

• It depends upon the medium in contact with its surface

- It decreases with increase of temperature.
- It decreases when a substance dissolves partially.

• It increases when a substance dissolves completely.

APPLICATIONS OF SURACE TENSION:

- Rain drops and small droplets of mercury are spherical in shape due to surface tension.
- A new needle can be made to float on the surface of water due to surface tension.
- Mosquitos, spiders etc., can easily walk on the surface of water due to surface tension.
- The property of surface tension is used in the manufacture of lead shots.

• Surface tension of liquid metals is very very high **WORK:**

 Work done per unit area in increasing the surface area of a film is known as surface energy.
 S = W/A. Units - J/m² (or) erg/cm².

- Work done in forming a liquid drop is, w = Surface Area x Surface tension = $AT = 4 \pi r^2 T$.
- Work done in increasing the size of liquid drop from radius r_1 to r_2 is $W = 4 - T(r_2^2 - r_2^2)$
 - $W = 4 \pi T (r_2^2 r_1^2)$
- Work done in blowing a soap bubble is $w = 8 \pi r^2 T.$
- Work done in increasing the size of a soap bubble from radius r_1 to r_2 is, $W = 8 \pi T (r_2^2 - r_1^2)$
- Work done in forming a circular liquid film of radius 'r' is, $w = 2 \pi r^2 T$.
- Work done in increasing the area of circular soap film from radius r_1 to r_2 is $W = 2 \pi T (r_2^2 - r_1^2)$.
- When a liquid drop of radius 'R' splits into 'n' identical droplets, then the total surface area of n droplets will be greater than that of big drop. So energy is absorbed.
- Work done to split a liquid drop of radius 'R' into 'n' no. of identical droplets is $W = 4 \pi R^2 T (n^{1/3} - 1).$
- when n identical droplets are combined to form a big drop, then the energy will be released.
- Work done to form a big drop from 'n' identical droplets each of radius 'r' is $W = 4 \pi r^2 T (n n^{2/3})$.
- A thin disc of radius R has a hole of radius r. The energy required to pull it from the surface of water

of surface tension T is(W= $T\pi (R^2 - r^2)$)

• If a number of identical smaller drops merge to form a single big drop of radius R, the rise in

temperature of the drop is $\frac{3T}{dJS}\left[\frac{1}{r} - \frac{1}{R}\right]$

Where J is mechanical equivalent of heat, S is specific heat and d is density

A large number of identical smaller drops each of radius r merge to form a single drop of radius R. The energy released in the process is converted into KE of bigger drop formed. Then speed of the

bigger drop formed is $\sqrt{\frac{6T}{d}} \left[\frac{1}{r} - \frac{1}{R} \right]$

ANGLE OF CONTACT:

- Angle of contact of a liquid with respect to a solid is the angle between the tangent drawn to the liquid surface at the point of contact and the surface of the solid, measured inside the liquid.
 - The angle of contact depends on solid-liquid pair, temperature and impurities.

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- The angle of contact may assume any value between 0° to 180°.
- If the angle of contact is less than 90° then the liquid wets the solid and if the angle of contact is greater than 90° then the liquid does not wet the solid.
- The angle of contact between water and glass free from grease and oil is nearly equal to zero.
- For mercury and glass pair, angle of contact is 140°.
- The angle of contact is not changed by the inclination of solid object in the liquid.
- Angle of contact increases with increase in temperature.
- Water proofing agents increase the angle of contact.
- Wetting agents decrease the angle of contact. CAPILLARITY:
- The rise or fall of a liquid column in a capillary tube dipped in a liquid is known as capillarity.
- Capillarity is due to surface tension and it is due to cohesive and adhesive forces.
- If the angle of contact is greater than 90°, then
 The liquid does not spread on the surface.
 - Liquid does not wet the surface.
 - Liquid exhibits convex surface.
 - There will be capillary depression.
 - •Cohesive forces will be greater than adhesive forces.
- If the angle of contact is less than 90° then a. The liquid spreads on the surface.
 - b. Liquid wets the surface.
 - c. Liquid exhibits concave surface.
 - d. There will be capillary rise.
 - e. Adhesive forces will be greater than cohesive forces.
- If the angle of contact is equal to 90°, then a. The level of the surface of the liquid will be horizontal
 - b. The level does not rise or fall.
 - c. Cohesive forces will be equal to adhesive forces.
- If the angle of contact is zero then the tangent drawn to the curvature will coincide with the wall of the immersed tube.
- The weight of the liquid column in the capillary tube is balanced by the force due to surface Tension.

 $2\pi r T \cos \theta = Mg$

 $2\pi r T \cos\theta = \pi r^2 h dg$ [$M = dV = d\pi r^2 h$]

Surface Tension by capillary tube method

$$T = \frac{r(h+r/3)dg}{2\cos\theta}$$

If r is very small compared to h, then

$$T = \frac{hrdg}{2\cos\theta}$$

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where r = radius of the capillary tube

- h = height of liquid column
- d = density of the liquid
- g = acceleration due to gravity
- θ = angle of contact
- When a capillary tube of insufficient length is dipped in the liquid, the liquid will not overflow but stays at the top with adjustable meniscus.
- When diameter of capillary tube increases twice, the height of liquid column falls down to half. $(r_1h_1 = r_2h_2)$.
- Since $h \propto \frac{1}{r}$, the graph between h and r is a

rectangular hyperbola.

When capillary tube is dipped vertically in a liquid then the rise in the liquid is h. If the tube is tilted by making the inclination ' θ ' with vertical then the slant height of the liquid is $l = h/\cos \theta$.

EFFECT OF TEMPERATURE ON SURFACE TENSION

Over small ranges of temperature, the surface tension of a liquid decreases linearly with the rise of temperature according to the relation S_t

 $= \mathbf{S}_{0} \left(1 - \alpha \Delta t \right)$

where a = temperature coefficient of surface tension

- $s_t = surface tension at t^0C$
- $s_0^{'}$ = surface tension at 0°C
- Δt = Change in temperature.
- α = temperature coefficient of surface tension
- In the case of molten copper and molten cadmium, surface tension increases with increase of temperature.
- The surface tension of any liquid at its critical temperature is zero.

EXCESS PRESSURE INSIDE A LIQUID DROP & SOAP BUBBLE

- Excess pressure inside a liquid drop of radius r is given by P = 2T/r.
 - Excess pressure in a soap bubble of radius r is given by P = 4T/r.

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• When two soap bubbles of radii a and b in vaccum coalesce under isothermal condition, then radius of the resultant bubble is,

$$R = \sqrt{a^2 + b^2}$$

• When a soap bubble of radius r₁ and another of radius r₂ are brought together, then the radius of curvature 'R' on the common interface is

$$R = \frac{r_1 r_2}{r_2 - r_1} \left(If \ r_2 > r_1 \right)$$

FORCE AGAINST SURFACETENSION:

- Force required to pull a wire of length 'l' from the surface of water of surface tension T is F=21T
- Force required to pull a circular ring of radius r from the surface of water of surface tension T is $F=4\pi rT$
- Force required to pull a rectangular plate of length 'l' and breadth 'b' from the surface of water of suface tension T is

F = 2(l+b)T

• Force required to pull a rectangular plate of lentgh 'l' and thickness 't' from the surface of water of surface tension T is

$$F = 2(l+t)T$$

- Force required to pull a circular disc of radius R from the surface of water of surface tension T is $F = 2\pi RT$
- Force required to pull a circular disc of radius R with hole of radius r from the surface of water of surface tension T is

 $F = 2\pi (R + r)T$

- Between two glass plates a water drop is squeezed to form a thin film of thickness (d) and surface area (A). The force required to seperate the two plates is, F = 2TA/d.
- If the radii of the two limbs of a U-tube are r_1 and r_2 , the difference between the levels of a liquid
 - poured in it is

$$h = \frac{2T}{dg} \left[\frac{1}{r_1} - \frac{1}{r_2} \right] \text{ here } r_1 > r_2$$

Where d is density and T surface tension of the liquid.

• When a cylindrical glass tube is closed at one end and it is made heavy such that it floats in the vertical position. If the depth of the heavy end is h below liquid surface, then $h = \frac{2\pi rT + mg}{\pi r^2 dg}$

Drops of liquid of density d_1 are floating half immersed in a liquid of density d_2 . If T is the surface tension of the liquid, then the radius of the drop is

$$r = \sqrt{\frac{3T}{\left(2d_1 - d_2\right)g}}$$

A capillary tube is vertically dipped in a liquid. The height of the liquid in the tube is 'h' and the total set up is kept in a lift.

If the lift is moving up with an acceleration 'a' then the height of the liquid in the tube is given by

$$h' = h \left[\frac{g}{g+a} \right]$$

If the lift is moving down with an acceleration 'a' then the height of the liquid in the tube is given by

$$h' = h \left[\frac{g}{g - a} \right]$$

CONCEPTUAL QUESTIONS

Two needles are floating on the surface of water. 01. A hot needle when touches water surface between the needles, then they move 1. Closer 2. Away 3. Out of the liquid 4. Into the liquid 02. When there are no external forces, shape of the liquid is determined by 1. Density of liquid 2. Temperature only 3. Surface tension 4. Viscosity 03. When a soap bubble is given a positive charge it expands. If it is given a negative charge, then it 1. Expands 2. Contracts 3. Remains Same 4. Does not hold negative charge. 04. The rise of liquid into capillary tube is h_1 . If the apparatus is taken in a lift moving up with acceleration, the height is h_2 , then $1. h_1 = h_2$ 2. $h_1 > h_2$ $3.h_{2} > h_{1}$ 4. $h_2 = 0$ 05. In a gravity free space, shape of a large drop of liquid is 2. Cylindrical 1. Spherical 3. Neither Spherical nor cylindrical 4. Nearly spherical

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06. A bubble of radius r, is inside the other of radius 14. Surface tension of liquid metals when compared to common liquids is $r_2 (r_2 > r_1)$. The radius of a bubble which has same pressure difference as the pressure 1. Very high 2. Very low difference of inside the smaller and outside the 3. Same order 4. Depends on temperature larger bubble is 15. A water proofing agent changes the angle of contact from 1. $\frac{r_1r_2}{r_2+r_1}$ 2. $\frac{r_1-r_2}{r_1r_2}$ 3. $\frac{r_1r_2}{r_2-r_1}$ 4. $\frac{2(r_1r_2)}{r_2-r_2}$ 1. Acute to $\pi/2$ 2. $\pi/2$ to obtuse 3. Acute to obtuse value 07. An air bubble of radius r is formed at a depth h 4. Obtuse to acute value below the surface of water. The pressure inside 16. A drop of water of surface area A and radius R the bubble is : $[T = surface tension, P_0 =$ at a temperature T breaks into two drops of atmospheric pressure, d = density of water) area A_1 and A_2 , radii r_1 and r_2 and having temperatures T_1 and T_2 . Then the correct 1. $P_0 + \frac{2T}{r}$ 2. $\frac{4T}{r} + \frac{h}{r}$ statement is 1. $A = A_1 + A_2$ 3. $R = r_1 + r_2$ 2. $T = T_1 + T_2$ 4. $A < (A_1 + A_2)$ 3. $P_0 + hdg + \frac{4T}{r}$ 4. $P_0 + hdg + \frac{2T}{r}$ Surface tension of water is T_1 . When oil spreads 17. 08. Neglecting gravity, the potential energy of a on water surface tension becomes T_2 , then molecule of a liquid on the surface of the liquid when compared to the potential energy of a 1. $T_1 > T_2$ 2. $T_1 = T_2$ 3. $T_1 < T_2$ 4. $T_2 = \frac{T_1}{2}$ molecule inside the liquid is 1. Greater 2. Less 3. equal If 'L' is the capillary rise or dip and 'A' the cross 18. 4. depending on the liquid, sometimes less and sectional area of the tube, other conditions being sometimes more the same, then 09. The nature of r-h graph ('r' is radius of capillary 1. LA = Constant 2. $L\sqrt{A}$ = Constant tube and 'h' is capillary rise) is 1. Straight Line 3. L/A = Constant 4. L/\sqrt{A} = Constant 2. Parabola 3. Ellipse 4.Rectangular hyperbola A disc of paper of radius r is floating on the 19. 10. The factor on which the rise of liquid in a capillary surface of water of surface tension T. Then force tube does not depend is of surface tension on the disc is 1. Density of Liquid 2. Atmospheric pressure 1. T. π r 2. T. 2 π r 3. T. 4 π r4. $\frac{T}{2\pi r}$ 3. Radius of Capillary 4. Angle of Contact 11. It is difficult to fill a capillary tube with mercury than with water since 20. A disc of paper of radius R has a hole of radius 1. Angle of contact between glass & mercury is r. It is floating on the surface of a liquid of more than 90° and the angle of contact between surface tension 'T'. Then force of surface tension glass and water is less than 90° . on the disc is 2. Angle of contact is between glass and mercury 1. T.2 π R 2. T.2 π (R-r) 3. T.2 π (R+r) is less than 90° and the angle of contact between 4. T.4 π (R+r) glass and water is more than 90° . 21. A disc of paper of radius R has a hole of radius 3. Angle of contact is same for both water and r. It is floating on the surface of a liquid of surface tension 'T'. The energy needed to pull it mercury. 4. Mercury is denser than water. out carefully is 12. The surface tension of a liquid at its boiling point is 1. T.2 π (R+r) 2. T. π (R²-r²) 4. T. πr^2 1. Maximum 2. Zero 3. T. π (R²+r²) 3. Same as at room temperature 4. Minimum but more than zero KEY 13. The addition of soap changes the surface tension 03)1 01)2 02) 3 (04) 2of water to T₁ and that of sugar solution changes 05) 1 06) 1 (07)408)1 to T₂. Then 11)1 12)2 (09) 410) 2 1. $T_1 = T_2$ 2. $T_1 > T_2$ 3. $T_1 < T_2$ 4. T_2 / T_1 13)3 14) 1 16)415)318) 2 19)2 20)3 17) 3 21)2 **JR. PHYSICS** 313 SURFACE TENSION

NUMERICAL QUESTIONS LEVEL-1

WORK:

- 01. The work done in forming a soap bubble of radius 10 cm from soap water of surface tension 35 dyne cm⁻¹ is Joule 1.8800 2.8.8 x 10⁻³ 3.88 4.14000
- 02. The work done in incressing the raduis of a soap bubble from 4 cm to 5 cm is Joule(given surface tension of soap water to be 25 x 10⁻³ N/m)
 - 1. 0.5657×10^{-3} 2. 5.657×10^{-3}
 - 3. 56.5×10^{-3} 4. 565×10^{-3}

CAPILLARYTUBE

- 03. Surface tension of water is 70×10^{-3} Nm⁻¹ and g = 10 ms⁻². When a capillary tube of radius 0.5 mm is dipped in water, the length of the liquid column due to rise of liquid in the capillary is cm 1.0.28 2.0.028 3.2.8 4.2.8 x 10⁻³
- 04. When two capillary tubes A and B are immersed in water, the heights of water columns are found to be in the ratio 2 : 3. The ratio of the radii of tubes A and B is
 - 1. 2:3 2. 4:9 3. 9:4 4. 3:2
- 05. A capillary tube of radius 0.25 mm is dipped vertically in a liquid of density 800 kg m⁻³ and of surface tension $3x 10^{-2}$ Nm⁻². The angle of contact of liquid-glass is given by Cos q = 0.3. If g = 10ms⁻² the rise of liquid in the capillary tube is Cm 1. 9 2. 0.9 3.9 x 10⁻³ 4. 0.09
- 06. Water rises to a height of 6 cm in a capillary tube of radius 'r'. If the radius of the capillary tube is 3r, the height to which water will rise iscm.
 1. 18 2. 9 3. 2 4. 3
- 07. When a capillary tube is immersed in ethyl alcohol whose surface tension is 20 dyne cm⁻¹, the liquid rises to a height of 10 cm. Density of the liquid is 0.8 gmcm^{-3} . If $g = 10 \text{ ms}^{-2}$, the radius of the capillary tube is mm. (Angle of contact of ethyl alchol w.r.t. glass is 60°). 1. 0.0025 2. 0.025 3. 0.25 4. 2.5
- 08. The capillary rise of water in a tube of diameter 0.70 mm is 4 cm. If $g = 10 \text{ ms}^{-2}$, the surface tension of water is Nm⁻¹. 1. 70 x 10⁻³ 2.70 3. 0.35 x 10⁻² 4.7.0 x 10⁻³
- 09. When a clean lengthy capillary tube is dipped vertically in a beaker containing water, the water rises to a height of 8 cm. What will happen if another capillary tube of length 4 cm and same radius is dipped vertically in the same beaker containing water. Angle of contact of water is 0^{0} .
- 1. Water will flow out like a fountain. 2. Water will rise to a height of 4 cm only and the angle of contact will be zero. 3. Water will rise to a height of 4 cm only and the angle of contact will be 60° . 4. Water will not rise at all 10. Two capillary tubes 'A' and 'B' are dipped in water. The cross section of 'B' is four times that of 'A'. The capillary rise in 'B' is times that of 'A'. 1.2 2.0.5 3.4 4.0.25 11. Capillary tubes of diameters 1, 1.5, 2 mm are dipped vertically in the same liquid. The capillary ascents of the liquid in the tube are in the ratio 1.2:3:4 2.6:4:3 3.3:4:6 4.4:3:2 **EXCESS PRESSURE INSIDE A LIOUID DROP AND SOAP BUBBLE** 12. The surface tension of soap solution is 0.05 N.m⁻ ¹. If the diameter of the soap bubble is 4 cm. The excess pressure inside the soap bubble over that of outside is pascal 1.10 2.1.03.0.1 4.0.25 13. Two liquid drops have their diameters as 1 mm and 2 mm. The ratio of excess pressures in them is 1.1:22.2:13.4:14.1:414. The excess pressure inside a small air bubble of radius 0.05 mm in water of surface tension 70 dyne cm⁻¹ is pascal 1.28 2. 2.8×10^2 3.2800 4.280 If the diameter of a soap bubble is 20 mm and if 15. the surface tension of soap water is 0.04 N.m⁻¹, the excess pressure inside the bubble is .. Nm⁻² 2.1.6 1.0.16 3.0.016 4.16 16. The surface tension of a soap solution is 25 x 10^{-3} N/m. The excess pressure inside a soap bubble of diameter 1 cm is 1.5 pa 2. 10 pa 3. 20 pa 4. 40 pa **RADIUS OF THE RESULTANT BUBBLE** If a soap bubble of radius 3 cm coalesce with 17. another soap bubble of radius 4 cm under isothermal conditions, the radius of the resultant bubble formed is cm 3.5 1.7 2.1 4.12 18. A soap bubble of radius 6 cm and another bubble of 8 cm coalesce under isothermal conditions in vacuum. The radius of the new bubble is 1.3 cm 3.10 cm 4.7 cm 2.4 cm KEY 02)1 03)3 (04)401) 2 05)2 06) 3 07) 2 08)1 09) 3 10) 2 11)2 12)1 13)2 14) 3 15)4 16) 3

17) 3

18) 3

| WORK:01. A sap bubble of radius 2 cm explodes. If the surface tension of soap water is 40 dyne cm², the energy released is Joule 1.1280 m x 10² 2.1280water vises to a height of 3.5 cm. The specific gravity of mercury is 13.6 and angle of contact of water x 1.0² 2.128002. Surface tension of mercury is 1.6 as 1.2 1 : 6.8 : 1.2 1 : 6.8 : 1.2 : 6.8 : 1.2 : 6.8 : 1.2 : 6.8 : 1.2 : 6.8 : 1.2 : 6.8 : 1.2 : 6.8 : 1.2 : 6.8 : 1.5 4.1 : 1603. The work done in blowing a soap bubble slowly from a radius of 5 cm to a radius of 10 cm is | LEVEL-2 | 09. When a capillary tube is dipped vertically in |
|--|---|--|
| 02.Surface tension of mercury is $35 \times 10^{-3} \text{ Mm}^{-1}$. The energy spent in spraying a drop of mercury of radius 1 cm into 1 million drops of equal size is | WORK:01. A soap bubble of radius 2 cm explodes. If the surface tension of soap water is 40 dyne cm ⁻¹ , the energy released is Joule $1.1280 \pi \times 10^{-7}$ 2.1280 3.12800 4.128 | water, water rises upto a height of 10 cm. When the same capillary tube is dipped in mercury it falls through a height of 3.5 cm. The specific gravity of mercury is 13.6 and angle of contact of water w.r.t. glass is 0° . Angle of contact of mercury w r.t. glass is 135° Then the ratio of |
| 03. The work done in blowing a soap bubble slowly from a radius of 5 cm to a radius of 10 cm is | 02. Surface tension of mercury is $35 \times 10^{-3} Nm^{-1}$. The energy spent in spraying a drop of mercury of radius 1 cm into 1 million drops of equal size is micro joule | surface tensions of water and mercury is 1. 6.8 : 1 2. 1 : 6.8 3.1 : 5 4. 1 : 1 10. The surfacetension of a liquid is0.245Nm⁻¹. Its angle of contact is 60°. Its density is 1000 kg/ |
| $3.4.4 \times 10^3$ $4.8.8 \times 10^3$ 10^3 $4.8.8 \times 10^3$ $04.$ The energy required to break a mercury drop of 1 cm radius into 1000 small drops of equal size is | 1. 4356 2.4.356 3.43.56 4.0.4356 03. The work done in blowing a soap bubble slowly from a radius of 5 cm to a radius of 10 cm is | m³. When a capillary tube of diameter 1 mm is dipped vertically in it, the capillary rise is |
| 105. The amount of energy dissipated when 8 water drops of 0.6 mm radius coalesce to form one big drop is | 04. The energy required to break a mercury drop of 1 cm radius into 1000 small drops of equal size is Joule (given the surface tension of mercury is 0.4 N.m^{-1}) $1.144 \pi \times 10^{-2}$ $2.1.44 \pi$ $3.14.4 \pi \times 10^{-4}$ 4.144π | of cross section is used, then the height of the liquid column will be |
| CAPILLARY TUBE06. A capillary tube of diameter 0.4 mm is dipped(a capillary tube of diameter 0.4 mm is dipped(b capillary tube of diameter 0.4 mm is dipped(c capillary tube of contact of mercury w.r.t. glass is(c capillary tube) is(c capillary tube) is(c capillary tube) of the meniscus in the capillary tube is(c capillary tube) of radius 0.2 mm is(c capillary tube) of a capillary tube if(c capillary tube) of contact of water glass is 0^0 .(c capillary tube) the vater rises is | 05. The amount of energy dissipated when 8 water drops of 0.6 mm radius coalesce to form one big drop isJoule (given surface tension of water is 7.2×10^{-2} Nm ⁻¹) $1.1.3 \times 10^{-7}$ 2.13 x 10^{-6} 3.13×10^{-7} 4.0.13 x 10^{-8} | 1.0.2 cm 2.0.02 cm 3.2 cm 4.4 cm 13. A capillary tube is dipped in water vertically. Water rises to a height of 10mm. The tube is now tilted and makes an angle 60° with vertical. Now water rises to a height of 1.10 mm 2.5 mm 3.20 mm 4.40 mm |
| AND SOAP BUBBLE AND SOAP BUBBLE AND SOAP BUBBLE AND SOAP BUBBLE 10.24 2.24 3.2.4 4.0.024 11.0.24 2.24 3.2.4 4.0.024 12.1.2 2.24 3.2.4 4.0.024 13.1.1.2 2.2.1 3.1.1 4.1.4 14.1.2 2.2.1 3.1.1 4.1.4 15. The radius of a soap bubble is 4 cm. The radius of the drop formed by soap solution is 2 cm. The ratio of excess pressures in the two is 1.1.2 2.2.1 3.1.1 4.1.4 16. A small air bubble of 0.1 mm diameter is formed just below the surface of water. If surface tension of water is 0.072 Nm ⁻¹ , the pressure inside the air bubble in kilo pascal is (Atmospheric pressure = 1.01 x 10 ⁵ pa) 1.0.24 mm 2.2.4 mm 3.2 cm 4.1.2 cm 10.12 2.12 1 3.1.1 4.1.14 10.12 2.2.1 3.1.1 4.1.14 11.1.1.1.1.1.1.1.1.1.1.1.1.1.1.1.1.1.1 | CAPILLARY TUBE 06. A capillary tube of diameter 0.4 mm is dipped in a beaker containing mercury of density 13.6 x 10³ kg m⁻³ and surface tension 0.49 N m⁻¹. The angle of contact of mercury w.r.t. glass is 130°. [Cos 130° = -0.64280. The depression | of the moon as compared to that on the earth is 1. 1/6 times 2. 6 times 3. 4 times 4. 1/4 times |
| a dipped vertically in a beaker containing water of surface tension 7.0 x 10⁻² Nm⁻¹. The height to which water rises is | of the meniscus in the capillary tube is | AND SOAP BUBBLE 15. The radius of a soap bubble is 4 cm. The radius of the drop formed by soap solution is 2 cm. The ratio of excess pressures in the two is 11:2 22:1 31:1 41:4 |
| surface tension of water is /.2 x 10 ⁻² Nm ⁻¹ 1.0.24 mm 2.2.4 mm 3.2 cm 4.1.2 cm | dipped vertically in a beaker containing water of surface tension 7.0 x 10^{-2} Nm ⁻¹ . The height to which water rises is | 16. A small air bubble of 0.1 mm diameter is formed just below the surface of water. If surface tension of water is 0.072 Nm⁻¹, the pressure inside the air bubble in kilo pascal is (Atmospheric pressure = 1.01 x 10⁵ pa) 1.28.9 2.0.289 3.0.0289 4.103.88 |
| JR. PHYSICS SURFACE LENSION | surface tension of water is /.2 x 10 ⁻² Nm ⁻¹ 1.0.24 mm 2.2.4 mm 3.2 cm 4.1.2 cm JR. PHYSICS | 315 SURFACE TENSION |

RADIUS OF CURVATURE OF INTERFACE

17. Two air bubbles of radii 2 mm and 4 mm, formed in the same liquid come together to form a big bubble. If the surface tension of the liquid is 0.70 Nm⁻¹, the radius of curvature of common interface to both bubbles will be

1.6 mm with concave surface towards smaller bubble.

2.2 mm with concave surface towards bigger bubble.

3.4 mm with concave surface towards smaller bubble.

4.4 mm with concave surface towards bigger bubble.

FORCE AGAINST SURFACE TENSION

A thin film of water of thickness 80 mm is 18. sandwiched between the glass plates and forms a circular patch of radius 12 cm. The normal force required to separate the plates is (surface tension of water is 0.072 N/m)

1.8.14 N 2.81.4 N 3.40.7 N 4.0.814 N

19. Calculate the force required to separate the glass plates of area 10⁻² m² with a film of water 0.05 mm thickness between them (surface tension of water = $70 \times 10^{-3} \text{ N/m}$)

- 20. A vessel has a small hole at its bottom. If water can be poured into it up to a height of 7 cm without leakage ($g=10 \text{ ms}^{-2}$), the radius of the hole is (surface tension of water is 0.07 Nm^{-1}). 1.2 mm 2.0.2 mm 3.0.1 mm 4.0.4 mm
- 21. A square wire frame of side 'L' is dipped in a liquid. On taking out, a membrane is formed. If the surface tension of liquid is T, the force acting on the frame due to the membrane will be 1.2 TL 2.4 TL 3.8 TL 4.16 TL
- If the force required to pull out a glass plate of 22. length 9.8 cm and thickness 2 mm from a liquid is 0.6 gmwt. The surface tension of water is ... Nm⁻¹ 1.2.94 x 10⁻³ 2.29.4 x 10³ 4.29.4 x 10⁻³ 3.29.4 x 10⁻²
- The length of a rubber cord floating on water is 23. 5 cm. The force needed to pull the cord out of water is N (surface tension of water is $7.2 \ge 10^{-4} \text{ Nm}^{-1}$).

4)
$$w = 4\pi R^2 T \left[n^{\frac{1}{3}} - 1 \right]$$

5)
$$w = 4\pi r^2 T \left[n - n^{\frac{2}{3}} \right]$$

6)
$$h = \frac{2\cos\theta T}{rdg}$$

7)
$$h = \frac{2T}{rdg}$$

8)
$$r = \frac{2T}{hdg}$$

9)
$$T \infty \frac{hd}{\cos\theta}$$

11)
$$\frac{h_1}{h_2} = \sqrt{\frac{A_2}{A_1}} = \frac{r_2}{r_1}$$

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13)
$$h^1 = \frac{h}{\cos\theta}$$

11)

14)
$$g^{1} = \frac{g}{6}$$

15) $\frac{P_{1}}{P_{1}} = \frac{4T/r_{1}}{1} = \frac{4r_{2}}{1}$

15)
$$\frac{1}{P_2} = \frac{1}{2T/r_2} = \frac{n_2}{2r_1}$$

16) Total pressure inside air bubble= excess pressure inside soap bubble+ Atmospheric pressure

$$P = \frac{2T}{r} + P_0$$
18)
$$F = \frac{2T}{d}A = \frac{2T}{d}\pi r^2$$

$$20) \quad r = \frac{2T}{d \lg}$$

For each wire force is 2 Tl so that for four wires 21) of frame 8T1

22)
$$T = \frac{F}{2(l+t)}$$

23)
$$T = \frac{F}{2l}$$

| | LEVEL 3 1 90.5 \times 10 ⁻⁶ 2 90.5 \times 10 ⁶ | | | | | | |
|-----|---|-----|--|--|--|--|--|
| | | | $\begin{array}{cccccccccccccccccccccccccccccccccccc$ | | | | |
| | A film of water is formed between two straight | 00 | When a 'II' shaped slider of negligible mass is | | | | |
| 01. | narallel wires each of 10 cm long and at | 09. | dinned in a soan solution and lifted a thin film of | | | | |
| | separation of 0.5 cm. The work to be done to | | apped in a soap solution and inted, a tim minor | | | | |
| | increase the distance between the wires by 1 | | soap is formed in the frame. It supports a weight $af2.0 \times 10^{-2}$ N. If the length of the alideric 40 are | | | | |
| | mm is micro joule (given surface | | of 2.0×10^{-1} N. If the length of the slider is 40 cm, | | | | |
| | tension of water is $72 \times 10^{-3} \text{ Nm}^{-1}$ | | the surface tension of the film of soap is | | | | |
| | $1 144 	 2 144 	 3 144 	 10^{-6} 	 4 144$ | | 1. 25 Nm^{-1} 2. 2.5 Nm^{-1} | | | | |
| 02 | Let W be the work done when a bubble of volume | 10 | 3. $2.5 \times 10^{-2} \text{ Nm}^{-1}$ 4. $2.5 \times 10^{-3} \text{ Nm}^{-1}$ | | | | |
| 02. | V is formed from a given solution. Work is required | 10. | A vessel whose bottom has round holes with | | | | |
| | to be done to form a bubble of volume 2V is | | diameter 0.1 mm is filled with water. The maximum | | | | |
| | $1.W$ $2.2W$ $3.2^{1/3}W$ $4.4^{1/3}W$ | | height up to which water can be filled without | | | | |
| | CAPILLARYTUBE | | leakage is cm (given surface tension = | | | | |
| 03. | A long capillary tube of radius 1 mm, open at | | $7.5 \times 10^{-2} \text{ Nm}^{-1} \text{ and } g = 10 \text{ ms}^{-2}$ | | | | |
| 001 | both ends is filled with water and placed | | 1.100 cm 2.75 cm 3.50 cm 4.30 cm | | | | |
| | vertically. What will be the height of water | 11. | A glass plate of length 20 cm and breadth | | | | |
| | column left in the capillary? (Surface tension of | | 2 cm and thickness 4.0 mm weighs 18 gm wt. in | | | | |
| | water is $73.5 \times 10^{-3} \text{ N.m}^{-1}$ | | air. It is held vertically with longside horizontally | | | | |
| | 1.03cm 2.3cm 3.6cm 4.0.03 cm | | and the lower half under water. If the surface tension | | | | |
| 04. | When an experiment is done to find the surface | | of water is 7 x 10^{-2} Nm ⁻¹ and g = 10 ms ⁻² , the | | | | |
| | tension of a liquid on earth by capillary rise method, | | apparent weight of the plate isN. | | | | |
| | the height of the liquid column is 4 cm. When the | | 1.0.1 2.0.01 3.0.0018 4. 0.12056 | | | | |
| | same experiment is done on another planet whose | 12. | A needle of length 5 cm is floating on water of | | | | |
| | mass is 4 times and radius is twice that of the earth, | | surface tension 0.07 Jm ⁻² and is not wetted by | | | | |
| | then the height of liquid column iscm | | water. Then the weight of the needle will be | | | | |
| | 1.4 2.2 3.1 4.8 | | $1.7 \ge 10^3 \text{ N}$ $2.7 \ge 10^{-3} \text{ N}$ | | | | |
| 05. | A 20cm long capillary tube is dipped in water. The | | 3.5.2 x 10 ⁻² N 4.4 x 10 ⁻² N | | | | |
| | water rises upto 8 cm. If the entire arrangement is | 13. | A small hole is made at the bottom of a hollow | | | | |
| | put in a freely falling elevator, the length of the water | | sphere. The water enters into it when it is taken to | | | | |
| | column in the capillary tube will be | | a depth of 40 cm under water. If the surface tension | | | | |
| 0.0 | 1.Zero 2.8 cm 3.4 cm 4.20 cm | | of water is 0.07 Nm ⁻¹ , diameter of the hole is | | | | |
| 06. | In a surface tension experiment, with a capillary tube | | 1.1/7 mm 2.1/14 mm 3.2/15 mm 4.3/14 mm | | | | |
| | water rised up to 0.1m. If the same experiment is repeated | 14. | The length of one edge of a glass plate of | | | | |
| | in an artificial satellite, which is revolving around the earth | | thickness 0.2 cm is 9.8 cm. If this edge of the | | | | |
| | and at an altitude of 3R from the surface of the earth, | | glass plate touches the surface of a liquid of | | | | |
| | water with the in the capital y tube up to a neight of $1.0.1 \text{ m}$ | | surface tension 60 dyne/cm, then it is pulled | | | | |
| | 3.0.4 m A Full length of the tube | | down with a force of (Assume that angle of | | | | |
| F | VCFSS PDFSSUDF INSIDE A LIOUID | | contact to be zero). | | | | |
| | DROP AND SOAP BUBBLE | | 1.1100 dyne 2.2400 dyne | | | | |
| 07. | The excess pressure due to surface tension inside | | 3.1000 dyne 4.1200 dyne | | | | |
| | a spherical drop is 6 units. If eight such drops | 15. | The material of a wire has specific gravity 8. If | | | | |
| | coalesce, the excess pressure inside the new | | it is not wetted by water. what is the maximum | | | | |
| | drop is | | diameter of the wire that will float on the surface | | | | |
| | 1.3 units 2.6 units | | of water? $(T = 70 \text{ dyne} / \text{cm})$ | | | | |
| | 3.12 units 4.48 units | | 1.0.75 cm 2.1.5 mm | | | | |
| F | ORCE AGAINST SURFACE TENSION | | 3.0.75 cm 4.1.5 cm | | | | |
| 08. | When a capillary tube of radius 0.2 mm is dipped | | KEY | | | | |
| | vertically in water of surface tension 0.072 Nm ⁻¹ and | | 01) 2 02) 4 03) 2 04) 1 | | | | |
| | of angle of contact is 0°, the weight of water supported | | 05) 4 06) 4 07) 1 08) 1 | | | | |
| | by surface tension in the capillary tube is N | | 09) 3 10) 4 11) 4 12) 2 | | | | |
| | | | 13) 3 14) 4 15) 2 | | | | |
| | HYSICS | 317 | SURFACE TENSION | | | | |

| | HINTS | B. As the aeroplane mo | ves fast on the runway the | | |
|---|---|--|--------------------------------|--|--|
| 1. | $w = 2T\Lambda A$ | pressure is more on th | ne upper surface of its wings | | |
| | | and less on the bottor | n surface of the wings. | | |
| 2 | r_{2} | 1. both A and B are tr | ue 2. A is true but B is false | | |
| | 2 | 3. A is false but B is true 4. both A and B are f | | | |
| | $w \propto v^{\frac{2}{3}}$ | 3. Consider the following two statements A and B a | | | |
| | | identify the correct answer. | | | |
| 3. | $\pi r^2 h dg = 2 \times 2\pi r T \cos\theta$ | A) The work done in | blowing a bubble of volume | | |
| | $\sigma m r^2$ | V is W, then the wor | rk done in blowing a soap | | |
| 4. | $\frac{S_1}{S} = \frac{m_1}{m} \times \frac{r_2}{r^2}$ | bubble of volume 2V | will be $2^{2/3}$ W. | | |
| | $g_2 m_2 r_1$ | B) The excess press | ure inside a soap bubble of | | |
| | $\frac{1}{3}$ = 1; (1 1; 1 | | 85 | | |
| 7. | $R = h^{2}r^{2}$ R= radius of the big drop | diameter D and surfa | ce tension S is $\frac{1}{D}$ | | |
| | r– radius of the small drop | 1) A&B are false | 2) A is false but B is true | | |
| | $r = R / n^{1/3}$ | 3) A is true but B is f | alse 4) A & B are true. | | |
| | -2T | 4. A drop of water breaks | into two droplets of equal | | |
| | P = | size. In this proces | s which of the following | | |
| 8 | $E = 2\pi \pi T \cos \theta$ | statements is correct | ? | | |
| 0. | $\Gamma = 2\pi T \Gamma \cos \theta$ | a) the sum of temperature | of the two droplets together | | |
| 0 | $T - \frac{F}{F}$ | is equal to the origina | l temperature of the drop. | | |
| 9. | l = 2l | b) the sum of masses of | the two droplets is equal to | | |
| | 2T | the original mass of the | ne drop. | | |
| 10. | $h = \frac{21}{dra}$ | c) the sum of the radii of | the two droplets is equal to | | |
| 11 | | the radius of the origi | nal drop. | | |
| 11. | Apparant weight of the plate in water = weight of the plate in water = $(Correspondence)$ | d) the sum of the surface | areas of the two droplets is | | |
| 12 | the plate in air - (Surface tension+ $Optimust$) E= 21 T | equal to the surface a | rea of the original drop. | | |
| 12. | $\Gamma = 2LI$ | 1. a is correct | 2) b is correct | | |
| 12 | $d = \frac{4T}{2}$ | 3) c is correct | 4) d is correct | | |
| 15. | Plg | <u></u> | <u>CY</u> | | |
| 14. | F = 2(l+t)T | 1)1 2)2 | 3) 4 4) 2 | | |
| | | MATCHING TYPE QUE | ISTIONS | | |
| 15 | $2lT = \frac{d_g}{dt} \times \pi r^2 \lg$ | 1. Match the following: | T : | | |
| 101 | d_r | List -1 | LISI - II | | |
| | NEW MODEL QUESTIONS: | a) Surface tension | epergy | | |
| | TRUE OR FALSE TYPE QUESTIONS: | h) Canillary rise | f) Domination of adhesive | | |
| 1. Co | insider the following two statements A and B and | o) cupillary fise | force | | |
| | Identify the correct answer. | c) Spherical shape of | | | |
| A) | work done in blowing a soap bubble of radius r | rain dropos | g) Increase in temperature | | |
| | and surface tension 1 is $8\pi r^2 T$. | d) Tiny droplets of | | | |
| B)A | n impurity which is highly soluble in liquid increases | water act as ball | | | |
| 1 | the surface tension of the liquid. | bearings. | h) Excess of pressure | | |
| 1, A & B are true 2) A is false but B is true | | The correct match is | | | |
| $\begin{vmatrix} 3 \\ 2 \\ 0 \end{vmatrix}$ | A is true but B is faise.4) A & B are faise | 1) a - e; b - f; c - g; d - h | | | |
| 2.00 | identify the correct choice in the given answers | 2) a h; b g; c f; d | d e | | |
| Δт | the excess of pressure inside a small drop is more | 3) a - f; b e; c g; d | h | | |
| | than that of a big dron | | | | |
| | and and of a off arop. | | | | |
| | | | | | |
| | | | | | |

4) a - g ; b - f ; c - e ; d - hList-II 2. Match List - I with List II e) $\sqrt{R_1^2 + R_2^2}$ Angle of contact a) 0^{0} e) liquid will rise f) $\frac{R_1R_2}{R_2 - R_1}$ b) 90° f) water silver c) = 90° g) liquid will not wet d) > 90⁰ h) pure water and glass g) $\frac{2T}{r}$ The correct match is 1) a--h; b -- g; c-- f; d-- e h) $\frac{T}{r}$ 2) a -- e; b-- f; c-g;d-- h 3) a--e; b--g; c--f; d--h 4) a--h; b-- f; c-- g; d-- e The correct match is 3. Match List I with List - II List - I List-II **ANSWERS:** 1) 4 2) 1 a) Meniscus of water in a glass e) convex 3) 1 Capillary tube b) Meniscus of water in silver f) curved 1. capillary tube c) Meniscus of mercury is glass g) flat capillary tube d) Meniscus of water in glass h) concave capillary tube of insufficient length The correct match is 1) a -- h; b-- g; c-- e; d-- f 2) a -- e; b-- f;c-- g; d -- h 2. the following cases. 3) a -- f; b-- e; c--g; d-- h 4) a-- h; b-- g; c-- f; d-- e 4. Study the following: List-II List - I a) Excess pressure e) Independent of temperature b) Surface energy f) Varies directly with (radius)² c) Capillary rise g) Varies directly proportional to temperature d) Angle of contact h) Varies inversely with $(Area)^{12}$ 3. i) Varies inversely with (Volume)^{1/3} The correct match is to their weights. d -- f 1) a--i; b--h; c -- e; 2) a--h; b--i; d -- e c -- g; liquid surface. 3 a -- h; b-- f; c -- i; d -- e 4) a--i; b--f; c -- h; d -- g 5. Match List-I with List-II liquid surface. List-I a) Radius of the common surface when two soap bubbles 1. of radii R_1 and R_2 coalesce ANSWERS b) Radius of the large drop when two liquid drops of 1) 1 2) 1 radii R₁ and R₂ combine at constant temperature c) Excess of pressure in cylindrical drop d) Excess of pressure in the air bubble in water

1) a -- e; b-- f; c-- g; d-- h 2) a--f; b-- e; c-- h; d-- g 3) a-- f; b-- e; c-- g; d-- h 4) a--e; b-- f; c-- h; d-- g 5)2 4) 4 **ORDER ARRANGING TYPE OUESTIONS:** Four capillary tubes a, b, c and d having diameters 2 mm, 1mm, 0.8mm, and 0.6mm are dipped in four beakers containing water with 'a', tube vertically 'b' tube 30° , 'c' tube 45° and 'd' tube 60° inclination with the vertical. Arrange the lengths of water column in the tubes in descending order. 1. d, c, b, a 2) d, a, b, c 3) a, c, d, b 4) a,b,c,d Identify the ascending order of the works done in a) Work done in increasing the radius of soap bubble from 2 cm to 4 cm (T = 0.04 Nm⁻¹). b) Work done in increasing the radius of a soap bubble from 2 cm to 4 cm (T = 0.04 Nm⁻¹). c) Work done in breaking a liquid drop of radius 1 cm into 10⁶ droplets of same size $(T=0.08 \text{ Nm}^{-1}).$ 1) b, a & c 2) c, a & b3) a, b & c 4) c, b & a Arrange the increasing order of forces required to pull the body from the surface of liquid in addition a) A circular ring of radius R with its plane on the b) A circular ring of radius R with its plane in the c) A wire loop of square shape having the diagonal

equal to 2R with its plane on the liquid surface.

3) 2

| Directions 1) (2) (1 3) (4) (1 5) 3 (6) (7) 2 (8) 1 MULTICORRECT TYPE QUESTIONS: A scattine and reason. While answering these questions you are required to choose any of the following four responses. A. If both Assertion and Reason are true, and the Reason is correct explanation of the Assertion. B. If hoth Assertion is false, but the reason is true. D. If Assertion is false, but the reason is true. A. Attrictal temperature intermolecular forces for R: Atthead temperature intermolecular forces for R: Atthead tensist encodend and gases become equal. Liquid can expand without any restriction. A. Attriation and pression of a liquid does not overflow. R: The mentiscus of liquid at the top of the tube becomes flat. A. At the tange in surface tension of the liquid does not overflow. R: The mentiscus of liquid teresastens on heating. A. At the impurities always decreases on heating. A. At the tange in surface tension of the liquid depends upon the degree of contamination of the flaguid does not overflow. R: The adhesive force between molecules. A. At The angle of contact is small herefore detergent max the clothes dust in less time the clothes dust in less time the A 2) B 3) C 4) D A. At The angle of contact is small herefore detergent max the liquid resist deforming forces. A. At the angle of contact is small herefore detergent max the liquid tesset dust in less time the higher document is dusting tension of the close wither the solid. A. A close splation of the liquid the point of the dises is an event of the solid of the adhes is ease of the context is small herefore detergent max the close there may is a dare correct to the solid of the same correct to a liquid the same correct to a liquid th | ASSERTION AND REASON TYPE QUESTIONS: | | ANSWERS | | | |
|---|--------------------------------------|--|--|--|--|--|
| These questions consist of two statements as Assertion and reason. While answering these questions you are required to choose any of the following four responses. A. If both Assertion and Reason are true and the Reason is correct explanation of the Assertion. C. If Assertion is flate, but the reason is false. D. If Assertion is false, but the reason is false. D. If Assertion is false, but the reason is false. D. If Assertion is false, but the reason is false. D. If Assertion is false, but the reason is false. D. If Assertion is false, but the reason is false. D. If Assertion is false, but the reason is false. D. If Assertion is false, but the reason is false. D. If Assertion is false, but the reason is false. D. If Assertion is false, but the reason is false. D. If Assertion is false, but the reason is false. D. If Assertion is false is statma calculated height of liquid in the tube, the liquid at the top of the tube becomes flat. D. A : Hot soap solution is preferred for washing dirty clothes. The change in surface tension of the liquid depends upon the degree of contacting. There change in surface tension of the liquid depends upon the degree of contacting. There change in surface tension of the liquid depends upon the degree of contacting in surface tension of the myniv. D (A) A (2) B (3) C (4) D A. A: The angle of contact is small therefore detergents rinse the clothes dust in less time. D (A) A (2) B (3) C (4) D A. A: The angle of contact for washing ting the adhesive force between molecules of water and glass is greater than cohesive forces between molecules of water and glass is greater than cohesive forces between molecules of water and glass is greater than cohesive forces between water molecules. D (A) A (2) B (3) C (4) D A. A: The angle of contact tween molecules of water and glass is greater than cohesive forces between nolecules of the solid (1) the solid of the solid. L. A (A) are correct (2) a, & c are correct 3) a & d are correct (2) a, b & d are correct 3) a & d are correct 1) a, b, & d are co | Dire | ctions | 1)12)13)14)15)36)17)28)1 | | | |
| Assertion and reason. While answering these questions you are required to choose any of the following four responses. A. (Fhoth Assertion and Reason are true and the Reason is correct explanation of the Assertion. B. If both Assertion and Reason are true, but the Reason is false. D. If Assertion is rote, but the reason is false. C. Oli is sprayed on stepanetical empenature, surface tension of aliquid becomes zero. R. Attribute temperature intermolecular forces for liquid and passe become equal. Liquid can expand without any restriction. H. A. Attribute temperature intermolecular forces for liquid in the tube, the liquid does not overflow. R. The meniscus of liquid at the top of the tube becomes flat. A. (Hot so any solution is preferred for washing dirty clothes. A. (Hot soan solution is preferred for washing dirty clothes. A. (Hot soan solution is preferred for washing dirty clothes. A. (Hot soan solution is preferred for washing dirty clothes. A. (Hot soan solution is preferred for washing dirty clothes. A. (Hot soan solution decreases on heating. A. (A the soan of sloup solution decreases on heating. A. (A the soan solution decreases on heating. A. (A the soan solution decreases on the clothes dust in less time 1) (A 2) (B 3) (C 4) (D A. (The angle of contact is small therefore detergents rims the clothes dust in less time 1) (A 2) (B 3) (C 4) (D A. (The angle of contact is small therefore detergents rims the clothes were molecules. A. (A angle so of the dide expands while a small by the correct 2) a k are correct 1) a k are correct 1) a k a care correct 1) a k are c | | These questions consist of two statements as | MULTICORRECT TYPE OUESTIONS: | | | |
| guestions you are required to choose any of the following four responses. A. If both Assertion and Reason are true, but Reason is correct explanation of the Assertion. B. If both Assertion and Reason are true, but Reason is not correct explanation of the Assertion. C. If Assertion is false, but the reason is false. D. If Assertion is false, but the reason is false. D. If Assertion is false, but the reason is false. A. A troitical temperature, surface tension of a liquid depends unbe the liquid does not overflue without any restriction. D(A 2) B 3) C 4) D A. A: Hot soap solution is preferred for washing dirty clothes. R: The ensiesus of liquid at the top of the tube becomes flat. D. A 2) B 3) C 4) D A. A: The impurities always decreases the surface tension of a liquid. R: The change in surface tension of the liquid depends upon the degree of contantination of the impurity. D(A 2) B 3) C 4) D A. A: The impurities always decreases the surface tension of a liquid. R: The change in surface tension of the liquid depends upon the degree of contantination of the impurity. D(A 2) B 3) C 4) D A. A: The angle of contact is small therefore detergents (DA 2) B 3) C 4) D A. A: The angle of contact is small therefore detergents (DA 2) B 3) C 4) D A. A: The angle of contact is small therefore detergents (DA 2) B 3) C 4) D A. A: The angle of contact is small therefore detergents (DA 2) B 3) C 4) D A. A: The angle of contact is small therefore detergents (DA 2) B 3) C 4) D A. A: The angle of contact is small therefore detergents (DA 2) B 3) C 4) D A. A: The angle of contact of the solid. C: A: The angle of contact is small therefore detergents (DA 2) B 3) C 4) D A. A: The angle of contact is small therefore detergents (DA 2) B 3) C 4) D A. A: The angle of contact is small therefore detergents (DA 2) B 3) C 4) D A. A: The angle of contact is small therefore detergents (DA 2) B 3) C 4) D A. A: The angle of contact is small therefore detergents (DA 2) B 3) C 4) D A. A: The angle of contact is small therefore detergents (DA 2) B 3) | | Assertion and reason. While answering these | 1. Identify the correct statements from the following. | | | |
| following four responses. A. If both Assertion and Reason are true and the Reason is correct explanation of the Assertion. B. If both Assertion and Reason are true, but the Reason is true. I. A startical temperature, surface tension of aliquid becomes zero. R. At this temperature intermolecular forces for liquid and gases become equal. Liquid cam expand without any restriction. I. A 2. JB 3. C 4. JD A. The thange of contact is small therefore detergration of a liquid depends upon the degree of contamination of the linguid depends upon the degree of contamination of the linguid depends upon the degree of contamination of the impurity. I. A. 2. JB 3. C 4. JD A. The change in surface tension of surge of contact is small therefore detergration of a liquid tersist deforming forces water molecules. I. A 2. JB 3. C 4. JD A. The change in surface tension of the liquid and assolid is a graemate of a liquid and assolid is a spherical. B. A. 2. JB 3. C 4. JD A. The change of contact is small therefore detergration of a liquid the surface tension is due a spherical of the adm and there of liquid by the length of the correct 4. Ja, a dare correct 4. Ja, a dare correct 4. Ja, a dare correct 4. Ja a dare correct 1. Ja dare corr | | questions you are required to choose any of the | a. Every liquid surface acquires a minimum surface | | | |
| A. If both Assertion and Reason are true and the Reason is correct explanation of the Assertion. B. If both Assertion and Reason are true, but Reason is correct explanation of the Assertion. C. If Assertion is true, but the Reason is false, D. If Assertion is false, but the reason is fue. D. If Assertion is the, but the reason is fue. A. At retrical temperature, surface tension of a liquid and gases become equal. Liquid can expand without any restriction. A. When height of attabe is less than calculated height of liquid in the tube, the liquid dest net use of a liquid. A. When height of attabe is less than calculated height of liquid in the tube, the liquid dest net use of a liquid. A. Hot soap solution is preferred for washing dirty clothes. R. Surface tension of soap solution decreases on heating. A. The impurities and explement and edge to a liquid is staren in tube. A. The impurities and explement and edge to a liquid is staren in a tube. A. The change in surface tension of the liquid depends supon the degree of contact is small therefore detergents in loss time. A. A. The rangle of contact of pue waterwing dass is greater than cohesive force between molecules. A. A. The adhesive force between molecules of water and glass is greater than cohesive force between molecules. A. A. The rangle of contact is small therefore detergents in less time. A. A. The rangle of contact is small therefore detergents in less time. A. A. The rangle of contact of pue waterwing dass is acces. A. The rangle of contact of pue waterwing dass is greater than cohesive force between molecules. A. A. The rangle of contact is small therefore detergents. A. A. The rangle of contact is small therefore detergents. A. A. The rangle of contact is small therefore detergents. A. A. The rangle of contact is small therefore detergents. A. A. The rangle of cont | | following four responses. | area due to surface tension. | | | |
| Reason is correct explanation of the Assertion. Bif both Assertion and Reason are true, but Reason is not correct explanation of the Assertion. C. If Assertion is true, but the Reason is false. D. If Assertion is true, but the Reason is true. A. Art critical temperature, surface tension of a liquid becomes zero. R. At this temperature intermolecular forces for liquid and gases become equal. Liquid can expand without any restriction. 1) A 2) B 3) C 4) D A. Hohe height of a thick is less than calculated height of liquid in the ube, the liquid does not overflow. R: The meniscus of liquid at the top of the tube becomes flat. 1) A 2) B 3) C 4) D A. Hoto sap solution is preferred for washing dirty clothes. 1) A 2) B 3) C 4) D A. The impurities always decreases the surface tension of soap solution decreases on heating. 1) A 2) B 3) C 4) D A. The impurities always decreases the surface tension of a liquid. A. The tangle of contact is small therefore detergents inset the clothes dust in less time 1) A 2) B 3) C 4) D A. The tangle of contact is small angles of contact. R: The adhesive force between molecules of water and glass is greater than objeger drops. B. A. The adpelocomatch of purvater with glass is acut. R: The adhesive force between molecules of water and glass is greater than cohesive force between water molecules. A. A large soap tubble expands while a small bubble shrinks, when they are connected to each other by a capillary tub. A. A large soap subble expands while a small bubble shrinks, when they are connected to each other by a capillary tub. A. A large soap subble increases, as its volume increases. A) A 2) B 3) C 4) D A. A large soap subble expands while a small bubble shrinks, when they are connected to each other by a capillary tub. A. A large soap subble expands while a small bubble shrinks, when they are connected to each other by | | A. If both Assertion and Reason are true and the | b. Small liquid drops are spherical. | | | |
| B.If both Assertion and Reason are true, but Reason is not correct explanation of the Assertion. C. If Assertion is true, but the Reason is false. D. If Assertion is true, but the reason is false. D. If Assertion is funce, surface tension of a liquid becomes zero. R: At this temperature intermolecular forces for liquid and gases become equal. Liquid can expand without any restriction. R: At this temperature intermolecular forces for fliquid in the tube, the liquid does not overflow. R: The meniscus of liquid at the top of the tube becomes flat. A: When height of a tube is best than calculated height of liquid. R: The meniscus of liquid at the top of the tube becomes flat. R: The tube, the liquid does not overflow. R: The meniscus of liquid at the top of the tube becomes flat. R: The adhesive forces and cohesive forces cause friction and cohesive forces cause surface tension. A: Hot soap solution is preferred for washing dirty clothes. R: The change in surface tension of the liquid depends upon the degree of contamination of the impurity. R: The change in surface tension of the liquid depends upon the degree of contamination of the impurity. R: The adhesive force between molecules of water molecules. R: The adhesive force between molecules of water molecules. R: The adhesive force between molecules of water molecules. R: The adhesive force between molecules of water molecules. R: The adhesive force between molecules of water molecules. R: The adhesive force between molecules of water molecules. R: The adhesive force between molecules of water molecules. R: The adhesive force between molecules of water molecules. R: The adhesive force between molecules of water molecules. R: The adhesive force between molecules of water molecules. R: The adhesive force between molecules of the solid. R: A a large soap bubble expands wh | | Reason is correct explanation of the Assertion. | c. Oil is sprayed on stagnated water pools to | | | |
| Reason is not correct explanation of the Assertion. C. If Assertion is true, but the Reason is false. D. If Assertion is true, but the Reason is true. A. Actoritical temperature, surface tension of a liquid becomes zero. R: At this temperature intermolecular forces for liquid any restriction. 1) A 2) B 3) C 4) D A. When height of athick is less than calculated height of the surface of a liquid becomes flat. B. A: Hot scap solution is preferred for washing dirty clothes. R: Surface tension of soap solution decreases on heating. 1) A 2) B 3) C 4) D A. A: The imperiature is small therefore detargents rinse the clothes dust in less time 1) A 2) B 3) C 4) D A. A: The change in surface tension of the liquid depends upon the degree of contact is small therefore detargents rinse the clothes dust in less time 1) A 2) B 3) C 4) D A. A: The imperiation of the sist of the context is small therefore detargents rinse the clothes dust in less time 1) A 2) B 3) C 4) D A. A: The imperiation of the strue context of pure waterwidg glass is greater than cohesive force between water molecules. 1) A 2) B 3) C 4) D A: The imperiation of the liquid desends upon the degree of contact is small therefore detargents rinse the clothes dust in less time 1) A 2) B 3) C 4) D A: The angle of contact for the strue detargents rinse the clothes dust in the strue context of pure waterwidg glass is greater than cohesive force between water molecules. 1) A 2) B 3) C 4) D A: A: Large scap bubble expands while a small bubble strinks, when they are connect to cach other syne a capillary tube. R: The excess pressure (ducto surface tension iniceases, a is troum decreases. a) A 2) B 3) C 4) D B: A: A large scap bubble expands while a small bubble strinks, when they are connect to cach other strue a difficult of the solid. b) the angle of contact between a concrect to a difficult will be oblics.<!--</th--><th></th><th>B.If both Assertion and Reason are true, but</th><th>destroy the mosquito breed by increasing the</th> | | B.If both Assertion and Reason are true, but | destroy the mosquito breed by increasing the | | | |
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| a) Some work mut be expended in bringing molecules on to the surface of a liquid. a) A : When height of a tube is less than calculated height of liquid in the tube, the liquid does not overflow. R: The mensicus of liquid at the top of the tube becomes flat. b) Surface tension is due to cohesive forces. c) Adhensive forces and cohesive forces do not exist simultaneously when a liquid is taken in a tube. c) A : Hot soap solution is preferred for washing dirty clothes. R: Surface tension of soap solution decreases on heating. 1) A 2) B 3) C 4) D c) A: The change in surface tension of the liquid depends upon the degree of contamination of the impurity. c) A. 2) B 3) C 4) D c) A: Detergent should have small angles of contact. R: If angle of contact is small therefore detergents rinse the clothes dust in less time 1) A 2) B 3) C 4) D f) A 2) B 3) C 4) D f. A: The angleofcontactof pure water with glass is acute. R: The adhesive force between molecules of water molecules. f) A 2) B 3) C 4) D f. A: The angleofcontactof pure water with glass is scatter. R: The adhesive force between molecules of water molecules. f) A 2) B 3) C 4) D f. A: The angleofcontactof pure water with glass is acute. R: The adhesive force between molecules of water molecules. f) A 2) B 3) C 4) D f. A: A large soap bubble expands while a smalt bubble shrinks, when they are connected to easion other by a capillary tube. g. A: A large soap bubble expands while a smalt bubble increases, as its volume decreases. g) A 2) B 3) C 4) D g. A: A large soap bubble expands while a smalt bubble increases, as its volume decreases. g) A 2) B 3) C 4) D g. A: A large soap bubble expands while a smalt bubble increases, as its volume decreases. g) A 2) B 3) C 4) D g. A: A large soap bubble increases, as its volume decreases. g) A 2) B 3) | | R: At this temperature intermolecular forces for | 2. Identify the correct statements from the following. | | | |
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| other by a capillary tube. R: The excess pressure (due to surface tension) inside a spherical bubble increases, as its volume decreases. a) A 2) B 3) C 4) D a) A 2) B 3) C 4) D b) the angle of contact will be obtuse, c. the liquid will descend in the capillary tube. d. the liquid will wet the solid. 1) only a & b are correct 2) only a & c are correct 3) a & d are correct 4) a, b & c are correct | | bubble shrinks, when they are connected to each | a) the miniscus will be convey upwords | | | |
| R: The excess pressure (due to surface tension) inside a spherical bubble increases, as its volume decreases. a) A 2) B 3) C 4) D | | other by a capillary tube. | a) the angle of contact will be obtuse | | | |
| inside a spherical bubble increases, as its volume decreases. a) A 2) B 3) C 4) D decreases. a) A 2) B 3) C 4) D decreases. b) C 4) D decreases. c) a c) A c) B c) C 4) D decreases. c) A c) C 4) D decreases. c) C 4) D decreases. c) C 4) D de | | R: The excess pressure (due to surface tension) | c the liquid will descend in the capillary tube | | | |
| decreases. a) A 2) B 3) C 4) D 1) only a & b are correct 2) only a & c are correct 3) a & d are correct 4) a, b & c are correct | | inside a spherical bubble increases, as its volume | d the liquid will wet the solid | | | |
| a) A 2) B 3) C 4) D 3) a & d are correct 4) a, b & c are correct | | decreases. | 1) only a & h are correct 2) only a & a are correct | | | |
| 5) a & d are correct 4) a, b & c are correct | | a) A 2) B 3) C 4) D | 1) only a \ll of are context (2) only a \ll c are context (2) only a \approx c are contex \approx c are context (2) only a \approx c are context (2) only a \approx | | | |
| | | | 5 j a α α a i c c c i i c i i c c i i | | | |

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 101×10^3

size of a

EAMCET-2000 (E) 17. A drop of liquid pressed between two glass 8000 identical water drops combine together plates spreads to a circle of diameter 10 cm. to form a big drop. Then the ratio of the final Thickness of the liquid film is 0.5 mm and surface tension is $70 \times 10^{-3} \text{ Nm}^{-1}$. The force required to surface energy to the initial surface energy of all the drops together is. pull them apart is 1.1:10 2.1:15 3.1:20 4.1:25 1. 4.4 N 2.1.1 N 3. 2.2 N **EAMCET-2000 (M) EAMCET-1993 (E)** 10. The surface energy of a liquid film on a ring of 18. A spherical soap bubble of radius 1 cm is formed area 1.5 m² is (surface tension of liquid=5Nm⁻¹ inside another of radius 3 cm. The radius of 1.0.75 Joule 2.7.5 Joule single soap bubble which maintains the same 3.2.25 Joule 4.3.0 Joule pressure difference as inside the smaller and **EAMCET-1999 (M)** outside the larger soap bubble is -----cm. When a cylindrical tube is dipped vertically into 11 1.1 **EAMCET-1992 (M)** a liquid the angle of contact is 140°. when the tube is dipped with an inclination of 40° the 19. A spherical soap bubble has a radius 'r'. The angle of contact is surface of tension of the soap solution is 'T'. 1.100° 2.140° 3.180° 4.60° The energy needed to double the diameter of **EAMCET-1999 (M)** the bubble at the same temperature is 12. When a capillary tube is lowered into water, the 1. 24 π r²T 2.4 π r²T 3.2 π r²T 4.12 π r²T **EAMCET-1992 (E)** mass of the water raised above the outside level is 5 gm. If the radius of the tube is doubled the 20. Three small identical mercury drops in a mass of water that raises in the tube above the thermally isolated system merge into a single drop. The temperature of the drop will be outside level is than the original temperature of small drops 1.1.25 gm 2.5 gm 3.10 gm 4.20 gm **EAMCET-1998 (E)** 1. less 13. Neglecting gravity, the potential energy of a 3. equal to molecule of a liquid on the surface of a liquid **EAMCET-1991 (M)** The work done to get n smaller identical drops when compared to the potential energy of a 21. molecule inside a liquid is to form a big spherical drop of water is 1. greater 2. less 3. equal proportional to 4. depending upon the liquid some times more, $1.\frac{1}{n^{2/3}-1} \quad 2.\frac{1}{n^{1/3}-1} \quad 3.n^{1/3}-1 \quad 4. \quad n^{4/3}-1$ some times less **EAMCET-1998 (M) EAMCET-1991** 14. Two soap bubbles are blown. In first soap 22. If surface tension of water is $7.3 \times 10^{-2} \text{ N/m}$. bubble excess pressure is 4 times of the second The excess pressure inside a spherical drop of soap bubble. The ratio of the radii of the first radius 1 mm is and second soap bubble is 1.146 2.1:2 1.1:4 3.2:1 4.4:1 **EAMCET-1990 EAMCET-1996 (M)** 23. The surface tension of soap water is 0.04 N.m⁻ 15. A capillary tube when immersed vertically in a liquid ¹. The excess pressure inside a 10 mm diameter rises to 3 cm. If the tube is held immesed in the soap bubble in N/m² will be liquid at an angle of 60° with the vertical, the length 1.4 of the liquid column along the tube will be **EAMCET-1990** 1.2 cm 2.4.5 cm 3.6 cm 4.7.5 cm 24. The rise of a liquid due to surface tension in a **EAMCET-1995 (M)** narrow capillary tube of diamter d is 'h'. When The liquid meniscus in a capillary tube will be 16. the diameter is reduced to d/2, the height will be convex, if the angle of contact is 1. greater than 90° 2. less than 90° 1.h 3. equal to 90° 4. equal to zero **EAMCET-1994 (E) EAMCET-1989 JR. PHYSICS**

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4.3.6 N

4.0.25

2.0.75

3.0.5

2. greater

 N/m^2

2. 1.46 3. 14.6

3.16

2.2h 3. $\frac{h}{2}$

2.8

4. may be less or greater

4. 292

4.32

4.4h

| 25. | At critical temperature surface tension becomes | EAMCET-2006 (E) | | | | |
|-----|---|---|--|--|--|--|
| | 1.0 2.1 3. Infinite 4. Negative | 33. Two soap bubbles combine to form a sing | | | | |
| EAN | ACET-1989 | bubble. In this process, the change in volume | | | | |
| 26. | The fundemantal quantity which has the same | and surface area are respectively 'V' and 'A'. | | | | |
| | power in the dimensional formula of surface | If P is the atmospheric pressure, and T is the | | | | |
| | tension and coefficient of viscosity is | surface tension of the soap solution, the | | | | |
| | 1. Mass 2. Length 3. Time 4. None | following relation is true | | | | |
| EAN | ACET-1988 | 1. $4 PV + 3 TA = 0$ 2. $3 PV - 4 TA = 0$ | | | | |
| 27. | Droplets of a liquid are generally more spherical | 3.4 PV - 3 TA = 0 4.3 PV + 4 TA = 0 | | | | |
| | in shape than large drops of the same liquid | <u>KEY</u> | | | | |
| | because | 1) 1 2) 2 3) 2 4) 2 | | | | |
| | 1. Force of surface tension is equal and opposite | 5) 4 06) 2 07) 1 08)2 | | | | |
| | to the force of gravity | 9) 3 10) 2 11) 2 12) 3 | | | | |
| | 2. Force of surface tension predominates the | 13) 1 14) 1 15) 3 16) 1 | | | | |
| | force of gravity | 17) 3 18) 2 19) 1 20) 2 | | | | |
| | 3. Force of gravity predominates the surface tension | 21) 3 22) 1 23) 4 24) 2 | | | | |
| | 4. Force of surface tension and force of gravity | 25) 1 26) 1 27) 2 28) 2 | | | | |
| | act in the same direction and are equal. | 29) 2 30) 2 31) 1 32) 1 | | | | |
| EAN | ACET-1987 | 33) 4 | | | | |
| 28. | The excess pressure inside a soap bubble is | QUESTIONS FROM OTHER | | | | |
| | 1. Inversely proportional to the surface tension | COMPETITIVE EXAMS | | | | |
| | 2. Inversely proportional to its radius | AIINIS-2002 | | | | |
| | 3. Directly proportional to square of its radius | 01. If the surface tension of water is $0.06 Nm^{-1}$ then the | | | | |
| | 4. Directly proportional to its radius | capillary rise in a tube of diameter 1 mm is $(\theta = 0^{\circ})$ | | | | |
| EAN | ACET-1987 | 1. 3.86 cm 2.3.12 cm 3. 2.44 cm 4.1.22 cm | | | | |
| 29. | The surface tension of a liquid with rise | AIIMS-1999 | | | | |
| | of temperature. | 02. A dry clean needle of diameter 'd' and density ρ | | | | |
| | 1. Increases 2.Decreases | when carefully placed on the surface of water | | | | |
| | 3. Remains same | remains floating. If T is the surface tension of water | | | | |
| | 4. First decreased and then increases | then maximum value for the diameter 'd' of the | | | | |
| | MCE1-2005 (M) | needle for enabling it to float will be | | | | |
| 30. | The pressures inside two soap bubbles are 1.01 | 80π | | | | |
| | and 1.02 atm respectively. The ratio of their | 1. $d = \sqrt{\frac{dp}{T}}$ 2. $d = \sqrt{\frac{dp}{T}}$ | | | | |
| | respective volumes is $1 + 1 = 1 + 2 + 4 = 2 + 4 + 2 = 1 + 2 + 4 + 2 + 2$ | VI g VI g | | | | |
| | 1.10 2.18 3.4 4.2 | $\boxed{8T}$ | | | | |
| | $\mathbf{MCE 1-2005 (E)}$ | $3. d = \sqrt{-31}$ 4. data incomplete | | | | |
| 31. | A thin wire ring of 5 cm radius flat on the surface | $\sqrt{\rho \pi g}$ | | | | |
| | before the film breeks is 20.14 x 10-3N more | JIPMER 1997 CPMT : 1999, 1986 | | | | |
| | then it is after. The surface tension of the liquid | 03. Liquid drops acquire spherical shape due to | | | | |
| | (in Nm ⁻¹) is | 1. gravity2. surface tension | | | | |
| | | 3. viscosity 4. intermolecular attraction | | | | |
| | 1. 80×10^{-3} 2. 87×10^{-3} | MNR 1999;CPMT 1999 | | | | |
| | 3. 90×10^{-3} 4. 98×10^{-3} | 04. A greased from heedle may float on | | | | |
| EAN | ACET-2006 (M) | 1 its shape 2 Archemedes' unlift force | | | | |
| 32. | A wire of length L metres, made of a material of | 3 surface tension effect 4 the statement is wrong | | | | |
| | specific gravity 8 is floating horizontally on the | ROHIL KHAND PFT 1909 | | | | |
| | surface of water. If it is not wet by water, the | 05 The height unto which water will rise in a capillary | | | | |
| | maximum diameter of the wire (in mm) upto | tube will be | | | | |
| | which it can continue to float is (surface tension | 1) maximum when water temperature is $4^{\circ}C$ | | | | |
| | of water is $T = 70 \times 10^{-3} Nm^{-1}$) | 2) minimum when water temperature is 4° C | | | | |
| | 1. 1.5 2. 1.1 3. 0.75 4. 0.55 | 3) minimum when water temperature is 0° C | | | | |

4) same at all temperatures

| AM | UPMT 1999 | MPPMT 1998 |
|-------|--|--|
| 06. | A metal disc of radius 'r' floats on the surface of | 14. At which of the following temperatures the value |
| | water. The water layer goes down and makes an | of suface tension is minimum. |
| | angle θ with the vertical edge of the disc. If it | $\begin{array}{c} 1)4^{\circ}C \\ 1)25^{\circ}C \\ 1007 \\ $ |
| | displaces a weight of W of water and the surface | 15 Mercury does not wet glass wood or iron |
| | tension of waer is T, then weight of disc is | because |
| | 1) 2π rT cos θ 2) 2π rT | 1) cohesive force is less than adhesive force |
| | 3) $2\pi rT \cos \theta + W$ 4) $2\pi rT \cos \theta - W$ | 2) cohesive force is greater than adhesive froce 2 |
| AM | UPMT 1999 | 3) angle of contact is less than 90° |
| 07. | A 10cm long wire is placed horizontally on the | RAJ PMT 1997 |
| | surface of water and is gently pulled up with a | 16. A capillary tube of radius'r' is immersed in water |
| | force of 2 x 10^{-2} N to keep the wire in | and water rises in it to a height H. Mass of water |
| | equilibrium. The surface tension of water in | in the capillary tube is m. If the Capillary of radius |
| | N/m is | that will rise in the capillary tube will be |
| | 1) 0.002 2) 0.001 3) 0.1 4) 0.2 | 1) m 2)2m 3)m/2 4) 4m |
| MPI | PMT 1997, BCECE 1999 | AIIMS 1997 |
| 08. | The pressure of air inside a soap bubble of diameter | 17. What is the pressure inside a water drop of radius |
| | 0.7 cm is 8mm of water above atmospheric | $\lambda_{\rm m}^{-2}$ (surfce tension of water $72 \times 10^{-3} M_{\odot}^{-1}$) |
| | pressure. The surface tension of soap solution is | χ_m (surfection of water $\chi_2 \times 10^{-1}$ χ_m) 1)101203 $\chi_1 = 2^{-2}$ 2)101/37 $\chi_2 = 2^{-2}$ |
| | 1) 9.8 ×10 ⁻¹ N/m 2) 6.8×10 ⁻² N/m | $\frac{1}{101295} \frac{1}{Nm} = \frac{2}{101457} \frac{1}{Nm} = \frac{1}{101457} \frac{1}{Nm} = \frac{1}$ |
| | 3) 7.2 × 10 ⁻² N/m 4) 1.37 × 10 ⁻¹ N/m | JIPMER 1997 |
| MPI | PET 1999 | 18. Two pieces of glass plate one upon the other with |
| 09. | The work done in increasing the size of a soap | a little water between them cannot be separated |
| | film from 10cm x 6cm to 10 cm x 11cm is | easily because of 1) inertia 2) pressure |
| | 3×10^{-4} Joule. The surface tension of the film is | 3) viscosity 4) surface tension |
| | $1)_{1.5 \times 10^{-2} N/m}$ 2) $_{3.0 \times 10^{-2} N/m}$ | MPPET 1997 |
| | 3) $6.0 \times 10^{-2} N/m$ 4) $11.0 \times 10^{-2} N/m$ | 19. One thousand small water drops of equal size |
| | MS 1998 | surface energy to the total initial surface energy is |
| 10. | The amount of a liquid cohesive force per unit | 1) 10:1 2) 1:10 3) 1000:1 4) 1:1000 |
| | length is called | MPPET 1996 |
| | 1) deperession 2) adhesion | 20. The amount of work done in blowing a soap bubble such that its diameter increases from d to D is |
| | 3) apparent weight 4) surface tension | (surface tension of solution=T) |
| AFN | | 1) $4\pi (D^2 - d^2)T$ 2) $8\pi (D^2 - d^2)T$ |
| 11. | If the surface tension of water is $0.06 Nm^{-1}$ then | 3) $\pi (D^2 - d^2)T$ 4) $2\pi (D^2 - d^2)T$ |
| | the capillary rise in the tube of diameter 1 mm is | CPMT 1997 |
| | $(\text{assume } \theta = 0^0)$ | 21. Excess pressure of one soap bubble is four times |
| | 1) $1.22 \text{ cm} 2)2.44 \text{ cm} 3) 3.12 \text{ cm} 4) 3.86 \text{ cm}$ | more than that of other. Then the ratio of volume of first hubble to second one is |
| 12 | Two capillary tubes A and B of radii 0.2 cm and 0.4 | 1)1:64 2)64:1 3)4:1 4)1:2 |
| 12. | cm are dipped in the same liquid. The ratio of heights | MPPET 1996 |
| | through which liquid rises in the tubes A and B is | 22. A capillary tube, made of glass is dipped into |
| | 1) $1:\overline{2}$ 2) $2:\overline{1}$ 3) $1:4$ 4) $4:1$ | 1) mercury rises in the capillary tube |
| AFN | AC 1998 | 2) mercury descends in capillary tube |
| 13. | A big drop of radius R is formed by 1000 small | 3) mercury rises and flows out of capillary tube |
| | droplets of water, then the radius of small drop is | 4) mercury neither rises nor descends in the capillary tube |
| | $1 \frac{R}{R} = 2 \frac{R}{R} = 2 \frac{R}{R} = 4 \frac{R}{R}$ | AMU 1996 |
| | $\frac{1}{2}$ $\frac{2}{5}$ $\frac{5}{6}$ $\frac{4}{10}$ | |
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| 3. 7 CPM 24. MF 25. | 23. The surface tension of water is T. The weight (W) of water supported by surface tension in capillary tubue of radius Ris T.4π R² T.2π R T.R T/2π R 24. The quantity on which the rise of liquid in a capillary tube does not depend is Density of liquid 2) radius of capillary tube angle of contact 4) atmospheric pressure MP PMT 1995 25. Work done in blowing a soap bubble of radius 10cm is (surface tension of soap solution is 3/100 N/m | | 28. | 28. A vessel, whose bottom has round holes with dian eter of 1mm is filled with water Assuming that su face tension acts only at holes, then the maximum height to which the water can be filled in vessed without leakage is (J & K CET 2002) (Given surface tension of water is 75 X 10⁻³ N/r and g = 10m / s² 1)3 cm 2) 0.3 cm 3)3 mm 4)3 m 29. If two soap bubbles of different radii are corn nected by a tube. (AIEEE 2004) 1) air flows from the bigger bubbles to the smaller bubble till the sizes become equal. 2) air flows from bigger bubble to the smaller bubble till the sizes are interchanged 3) air flows from the smaller bubble to the bigger | | | | m- ur- im sel //m on- the ler ger. |
|---------------------------------|--|--|-----|--|-------|------|-------|--|
| 41 | 1)/J.JUX10 J | 2)37.00×10 J | | | K | EY | | |
| 1 - | $3)150.72 \times 10^{-4} J$ | 4) 75.36J | | 01) 3 | 02)3 | 03)2 | 04) 3 | |
| UMF″ ⊿2€ | IEET 1995 If σ is surface tension | the work done in breaking a | | 01)3 | 0235 | 03)2 | 08)2 | |
| J. , | big drop of radius 'R' | into 'n' drops of equal raidus is | | 05)2 | 06) 3 | 07)3 | 08)2 | |
| 4 | _ | BHU 1995 | | 09)2 | 10) 4 | 11)2 | 12)2 | |
| 4 | 1) $Rn^{2/3}\sigma$ | 2) $(n^{2/3}-1)\sigma R^2$ | | 13)4 | 14)4 | 15)2 | 16)2 | |
| 4 | 3) $4\pi R^2 (n^{1/3} - 1)\sigma$ | 4) $\pi R^2 (n^{1/3} - 1) \sigma$ | | 17)2 | 18)4 | 19)2 | 20)4 | |
| _7. | A soap bubble of ra excess pressure insid | dius 'r' is formed in air. The de the bubble is | | 21)1 | 22)2 | 23)2 | 24)4 | |
| 4 | $1)\frac{4T}{r} \qquad 2)\frac{2T}{r}$ | 3) $P_0 + \frac{2T}{r} 4$) $P_0 - \frac{4T}{r}$ | | 25)1 29)3 | 26)3 | 27)1 | 28) 3 | |
| | | * * | * * | * | | | | |