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Roll No.....

Total No. of Questions: 28]

[Total No. of Printed Pages: 4

XIIARKDN20

2002-B

CHEMISTRY

Time: 3 Hours]

[Maximum Marks: 70

Section-A

1 each

(Very Short Answer Type Questions)

T. What is a 'Semiconductor'?

- 3. What is IUPAC name of $K_3[Fe(CN)_6]$?
- 4. When phenol is heated with HI in presence of red phosphorus it give:
 - (A) ethyl iodide

(B) ethane

(C) ethylene

- (D) ether
- 5. The IUPAC name of:

is butane-2-amine.

(True/False)

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Turn Over

D--2-B

- 6. Vinegar is a dilute solution of
- 7. Vitamin 'C' is:
 - (A) Paracetamol

(B) Lactic acid

(C) Citric acid

(D) Ascorbic acid

Section-B

2 each

(Short Answer Type Questions-I)

- What is a fuel cell? State two advantages of H₂-O₂ fuel cell over ordinary cell.
- 9. Derive the equation for a rate constant for a first order reaction.
- 10. Write down the electronic configuration of :
 - (i) Cu⁺
 - (ii) Co³⁺

Or

Why are transition metals called d-block elements?

11. What are ambidentate ligands? Give examples.

Or

Give an example of ionization isomerism in coordination compounds.

- 12. Discuss Fittig's reaction.
- 13. Give uses of Chloroform.

Section-C

3 each

(Short Answer Type Questions-II)

- Describe Frankel defect in crystals.
- 15. State and explain Faraday's laws of electrolysis.
- 16. What is the difference between order of a reaction and its molecularity?

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- 17. What do you understand by the terms?
 - (i) Roasting
 - (ii) Calcination
 - (iii) Smelting
- 18. What are lyophobic and lyophilic sols? Give their chief characteristics.
- 19. What are interhalogen compounds? How are they classified?
- 20. Explain the following:
 - (i) Transition metals act as catalysts
 - (ii) Transition metals form coloured complexes
- 21. What are primary, secondary and tertiary amines? Give one example of each.

Or

Explain why aromatic amines are less basic than ammonia and aliphatic amines.

- 22. What are Alcohols? Give any two methods of preparation of alcohols.
- 23. What are Carbohydrates? How are they classified?

Or

Give sources and deficiency diseases of Vitamin A, Vitamin C and Vitamin D.

24. (i) Identify the type of polymer:

- (ii) Write the structure of Dacron.
- (iii) Define addition polymer.

Or

Give preparation and uses of Nylon 66.

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- 25. Describe the following with suitable examples of chemicals :
 - (i) Antipyretics
 - (ii) Antiseptics

Or

What are Detergents? Why are detergents preferred over soaps?

Section-D

5 each

(Long Answer Type Questions)

26. What are colligative properties? Derive the relationship between the elevation in boiling point and molecular mass of a non-volatile solute in solution.

Or

Define molarity and molality. Calculate molality of 2.5 g of ethanoic acid (CH₃COOH) in 75 g of Benzene.

27. Name the elements of nitrogen family based on the electronic configuration explain the valence of N and P. Give common oxidation states of Nitrogen.

Or

How is sulphuric acid manufactured by contact process? Why is sulphuric acid syrupy (viscous)?

28. What are carboxylic acids? Give any four methods of preparations of carboxylic acids.

Or

Explain the following reactions by giving an example:

- (i) Aldol condensation
- (ii) Wolf-Kashner reduction
- (iii) Rosenmund reduction

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