

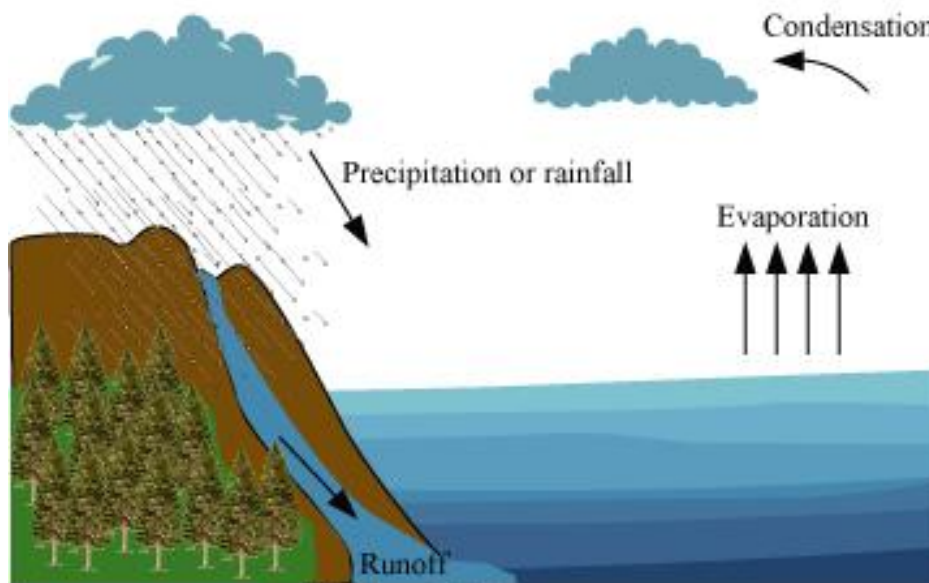
Water

- **Sources of water:** The common sources of water are ponds, lakes, rivers, wells, and reservoirs.
- **Water cycle:** Water undergoes different processes in the environment and is found in different states during these processes. This cyclic process through which water circulates in the environment is called the **water cycle**.
- **Importance of water**
 - Water is necessary for germination of seeds, transportation of nutrients from soil and food from the leaves to different parts of the plant, in preparation of food through photosynthesis.
 - Aquatic animals and plants get their nutrients as well as oxygen supply from the water. These substances are present in water in dissolved form.
 - Water is used for many other purposes such as in cooking, cleaning, industrial work, running hydroelectric and thermal electric power plants. Sea water is also used as a medium of transportation.
- A solution has two components, namely the solvent and the solute.
- **Solvent** is that part of the solution in which the other component is dissolved. In other words, solvent is that component of a mixture that is present in large amounts.
- **Solute** is that part of the solution that is dissolved in the solution. This is present in a lesser quantity as compared to the solvent. Also, more than one solute can be present in a solution.
- **Properties of a solution**
 - It is a homogeneous mixture of solutes and solvents
 - The solute particles in a solution are extremely small in size. They are less than 1 nm (10^{-9} m) in diameter.
 - Solute particles are not visible to the naked eye.
 - As a result of the small size of the solute particles, a solution does not scatter light.
 - Solute particles being small in size get dissolved in the solvent. Hence, the solute cannot be separated from the solution by filtration.
 - Solute particles do not settle down when left undisturbed.
 - The addition of harmful substances to water which causes its physical, chemical and biological properties to change is called **water pollution**.

- **Uses of Water**

Water is used for many purposes like drinking, washing clothes and utensils, generating electricity, bathing, irrigation etc.

- Water is essential for life.
 - About 71% of the earth's surface is covered with water. It is present as ground water, in seas, oceans, rivers, lakes, ice caps, and in atmosphere.
 - Wells, rivers, ground water reservoirs or lakes are sources of fresh water.
 - Ocean and sea water is not potable because they contain large amount of dissolved salts.
 - Water is a necessity for every form of life. It is used for domestic activities, irrigation, industrial purposes, etc.
 - 22nd March is celebrated as World Water day.
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- **Water Cycle:** The circulation of water between water bodies, lands and clouds is known as water cycle.



- The change of water into its vapour is known as **evaporation**.
- The change of vapour back into water is known as **condensation**.
- Fall of water from clouds is known as **precipitation** or **rainfall**.

- **Runoff** is the process by which rain water returns back to the water bodies.
 - **Clouds** are formed during the process of condensation.
 - Rain, snow, hail, etc. replenish rivers, lakes, and wells and recharge the ground water.
1. Water is present in three forms.
 2. Solid form – snow and ice
 3. Liquid form – oceans, lakes, rivers, and underground water
 4. Gaseous form – water vapour in the atmosphere
 5. The three forms of water circulate through the water cycle and keep the total amount of water constant on the earth.
- In free state, water occurs in solid, liquid, and gaseous state while in combined state, it is found in proteins, carbohydrates, etc.
 - It is colourless, odourless, and tasteless.
 - It boils at 100°C and freezes at 0°C .
 - Pure water is a poor conductor of electricity. It has high latent heat of vaporisation and fusion and specific heat capacity.
 - Water is a universal solvent as it dissolves maximum number of substances.
 - It is stable to heat.
 - Water has anomalous expansion property, i.e. it expands on cooling below 4°C and it has maximum density at 4°C

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- Water Cycle:** The circulation of water between water bodies, lands and clouds is known as water cycle.
- o The change of water into its vapour is known as **evaporation**.
 - o The change of vapour back into water is known as **condensation**.

- o Fall of water from clouds is known as **precipitation or rainfall**.
 - o **Runoff** is the process by which rain water returns back to the water bodies.
 - o Clouds are formed during the process of **condensation**.
 - o Rain, snow, hail, etc. replenish rivers, lakes, and wells and recharge the ground water.
 - o In India, most of the rainfall occurs during the monsoon season.
 - o Heavy rain in a short period of time causes flood. No rain for a long period causes **drought**.
 - o Most of the potable water becomes available to us as **ground water**.
 - o As the number of tube-wells in an area increases, the level of ground water decreases.
 - o There is an immediate requirement to harvest water. The basic idea behind rainwater harvesting is '**Catch water where it falls**'.
 - o **In rainwater harvesting method**, the stored rainwater is used to recharge ground water.
 - o **In rooftop rainwater harvesting system**, the rainwater is collected from rooftop in a storage tank. It allows the water to go into the pit through pipes. This water seeps and refills the ground water.
- The addition of harmful substances to water, as a result of which its physical, chemical, and biological properties get altered, is called **water pollution**.
- **Types of water pollutants**
- **Domestic sewage**
 - o It is composed of food wastes, detergents, and disease-causing pathogens.
 - o The bacteria present in faecal matter of mammals indicate the pollution levels in a river and if such water is consumed, it may cause various diseases.
- **Industrial waste**
 - o It is rich in toxic chemicals such as arsenic, fluorides, and lead.
 - o It causes toxicity in plants and animals.
 - o It affects the soil by causing changes in its acidity and growth of worms.
- **Agricultural waste**
 - o It is rich in agricultural pesticides and weedicides.
 - o It causes ground water pollution.
 - o It causes an increase in the population of algae in water.

- When these algae die, they are acted upon by decomposers, which use lots of oxygen dissolved in water leading to depletion of dissolved oxygen.
- This results in the death of fish and other aquatic organisms.
- **Release of Superheated Water**
 - The release of superheated water from some industries and nuclear power plants causes thermal pollution of the water bodies.
 - The abrupt change in the temperature of water body can kill the fish and other organisms adapted to particular temperature range.

Methods of preventing water pollution

- Industrial waste must be chemically treated to remove harmful substances before dumping into the water bodies.
- Disposal of human and animal excreta into water should be avoided.
- Sewage water must be treated before releasing into the rivers.
- **Conservation of water**
 - Reusing the waste water from the kitchen (water that has been used to wash vegetables, etc.) to water the plants in the garden
 - Turning the tap off while brushing or shaving
 - Checking for leaky taps and fixing them up
 - Rainwater harvesting
 - Using improved farming and irrigation techniques
 - Preventing pollution of water
 - Conserving and replenishing ground water
 - Proper removal of silt from water bodies
 - Preventing cutting of trees
- **Prevention of water pollution**
 - Proper treatment of industrial waste and domestic waste before their disposal into rivers.
 - Strict implementation of environmental laws in industrial units.

- Reusing water used in kitchens (such as to wash vegetables) for watering plants.
 - Getting the leaky taps checked and preventing wastage of water.
- **Potable water**
 - Water that is fit for drinking is called potable water.
 - Methods of obtaining potable water.
- **Physical methods**
 - Sedimentation and filtration
 - Boiling of water
 - Use of domestic filters such as candle type filter
- **Chemical method**
 - Use of chlorine tablets
 - Infusion of ozone gas
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