GENERAL SCIENCE, Paper-I

(Physical Sciences)
(English Version)
Parts A and B

Time: 21/2 Hours

Maximum Marks: 50

Instructions:

Answer the questions under Part 'A' on a separate answer book.

 Write the answers to the questions under Part 'B' on the question paper itself and attach it to the answer book of Part 'A'.

Part A

Time: 2 Hours

Marks: 35

SECTION I

 $5 \times 2 = 10$

Notes: 1. Answer any five questions, choosing at least two from each group.

2. Each question carries two marks.

Group - A

- What role does specific heat play in keeping a watermelon cool for a long time after removing it from a fridge on a hot day? Explain.
- 2. State any two differences between real and virtual images.
- 3. Write the lens Maker's formula and explain the terms in it.
- 4. A force of 8N acts on a rectangular conductor 20 cm long placed perpendicular to a magnetic field. Determine the magnetic field induction if the current in the conductor is 40 A.

Group - B

- 5. A shiny brown colored element 'X' on heating in air becomes black in color.
 Can you predict the element 'X' and the black colored substance formed?
 How do you support your predictions?
- 6. What is neutralization reaction? Give two examples.
- 7. What is the difference between roasting and calcination? Give one example for each.
- 8. An element has atomic number 19. Where would you expect this element in the periodic table? Why?

SECTION II

 $4 \times 1 = 4$

- Notes: 1. Answer any four questions from the following.
 - 2. Each question carries one mark.
- 9. Which mirror is used as rear-view mirror in the vehicles?
- Find the speed of light in a transparent medium, whose refractive index is
 3/2.
- 11. Define the power of lens.
- 12. $Fe_2O_3 + 2Al \longrightarrow Al_2O_3 + 2Fe$ Name the compound which is oxidized in the above reaction.
- 13. Write the electronic configuration of chromium.
- 14. Which method is suitable to enrich sulphide ores?

SECTION III

- Notes: 1. Answer any four questions, choosing at least two from each group.
 - 2. Each question carries four marks.

Group - A

- 15. Kavya can see distant objects clearly but cannot see objects at near distance. With what eye defect is she suffering? Draw the diagrams showing the defected eye and its correction.
- 16. How the optical fibres are working? What are the various uses of optical fibres in our daily life?
- 17. Find the radii of curvature of a convexo-concave convergent lens made of glass with refractive index n = 1.5 having focal length of 24 cm. One of the radii of curvature is double than the other.
- 18. State Ohm's law. Suggest an experiment to verify it and explain the procedure.

Group - B

- 19. Five solutions A, B, C, D and E when tested with universal indicator showed pH as 4, 1, 11, 7 and 9 respectively, which solution is
 - A) neutral
 - B) strongly alkaline
 - C) strongly acidic
 - D) weakly acidic
 - E) weakly alkaline

Arrange the pH in increasing order of hydrogen ion (H+) concentration.

20. What is hybridization? Explain the formation of Boron trifluoride (BF_3) molecule by hybridization.

19E (A)

- 21. How do you appreciate the role of electronic configuration of the atoms of elements in periodic classification?
- 22. Write IUPAC names for the following carbon compounds.

A)
$$CH_3 OH$$

$$CH_3 - CH - CH - CH_2 - CH_3$$

B)
$$CH_3 \quad Cl \\ | \quad | \quad | \\ CH_3 - CH_2 - CH - CH - CH_2 - COOH$$

C)
$$CH_3 - CH = C - CH_3$$
 Br

D)
$$CH_3 \qquad CH_3 \\ CH_3 - CH - CH_2 - CH_2 - CH - CH_3$$

SECTION IV

 $1 \times 5 = 5$

- Notes: 1. Answer any one of the following questions.
 - 2. Each question carries five marks.
- 23. Draw a neat diagram of electric motor. Name the parts.
- 24. Draw a neat diagram showing acid solution in water conducts electricity.

GENERAL SCIENCE, Paper-I

(Physical Sciences)

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Parts A and B

Time: 21/2 Hours

Maximum Marks: 50

Part B

Attach Part 'B' question paper to the main answer book of Part 'A'.

Marks: 15 Time: 30 Minutes Instructions: Answer all questions. 1. Each question carries ½ mark. 2. Answers are to be written in the question paper only. 3. Marks will not be awarded in case of any overwriting, rewriting or erased 4. answers. Write the 'CAPITAL LETTER' showing the correct answer for the following I. $20 \times \frac{1}{2} = 10$ questions in the brackets provided against them. Specific heat $S = \dots$ 1. (A) $Q/\Delta t$ $Q\Delta t$ We get a diminished image with a concave mirror, when the 2. object is placed at Focal point (F) (A)

between the Pole (P) and Focal point (F)

at the center of a Curvature (C)

(B)

(C)

9E	(B)								
1	3.	At a critical angle of incidence, the angle of refraction is							
		(A)	450	(B)	1800	1000			
		(C)	900	(D)	300				
	4.	Focal length (f) of the plano-convex lens is, when its							
		radius of curvature of the surface is R and n is the refractive							
			ex of the lens.			1			
			f = R		f = R/2				
		(C)	f = R/(n-1)	(D)	f = (n-1)/R				
	5.	Dur	ge.	1					
		(A)	frequency	(B)	wavelength				
		(C)	speed of light	(D)	all of these	3376			
	6.	Myopia can be corrected by using							
		(A)	convex mirror	(B)	concave mirror				
		(C)	double convex lens	(D)	double concave lens				
	7.	A charge is moved from a point A to a point B . The work done							
		to move unit charge during this process is called							
		(A)	potential at A						
		(B)	potential at B						
		(C)	$\operatorname{nd} B$						
		(D)							
	8.	A circuit draws a current of 2 A from a 12 V battery, its resultant							
		resistance is							
		(A)	12Ω	(B)	2Ω				
		(C)	6Ω	(D)	18Ω				
	9.	Which one of the following converts mechanical energy into							
		elect		[
		(A)	Generator	(B)	Battery				
		(C)	Motor	(D)	Switch				
	10.	The blue color of sky is due to							
		(A)	diffraction of light	(B)	interference of light				

(D)

scattering of light

polarization of light

19E	(B)							11 - 3
	11.	The reaction for formation of hydrogen chloride from hydrogen						
•		and chlorine represents following type of chemical reaction.						J
		(A)	Decomposition		(B)	Displacement		
		(C)	Double displace	ment	(D)	Combination		
	12.	2. Which gas is produced when metal carbonates react with aci						
		(A)	Oxygen		(B)	Carbon dioxide		
		(C)	Nitrogen		(D)	Hydrogen		
	13.	The state of the s						
	dis		ne 'L' shell of an at	tom is			L]
3,523		(A) (C)	8		(B) (D)	4 16		
	14.	Electrons enters into orbital after filling the 3d orbital.						. 1
		(A)	4s	de y	(B)	5s		,
36		(C)	4p		(D)	5 <i>p</i>	Nagegie.	
957.5	15.	Which one of the following element belongs to 3 rd period and III A group?						1
		(A)	Sodium		(B)	Potassium		
		(C)	Aluminium	**************************************	(D)	Argon	11.01	
	16.	Number of elements present in 3 rd period of the long form of periodic table.					ľ	1
		(A)	2 (B) 8	3	(C)	18 (D) 32		
	17.	Which one of the following types of medicines is used for treating indigestion?						1
		(A)	Antibiotic		(B)	Analgestic		
		(C)	Antacid		(D)	Antiseptic		
	18.	The	ore of aluminium	is			[]
		(A)	magnesite		(B)	galena		
	4	(C)	gypsum		(D)	hauvite		

1011	(D)			200						
	19.	Whi	ch one of the follow	wing is uns	atura	ated hy	vdrocarbon? []		
		(A)	C_2H_6		(B)	C_3H_8				
		(C)	C_3H_6		(D)	CH_4				
	20.	When sodium metal is dropped in ethanol gas will be released.								
		(A) (C)	carbon dioxide oxygen		(B) (D)	meth hydr				
II.	Fill	in the	e following blanks	with suital	ole ar	nswers	$5 \times \frac{1}{2} = 2\frac{1}{2}$	2		
	21.	The	eye lens can				ngth due to working o			
	22.	The relation between focal length and radius of curvature is given by								
	23.	The lens which can form real and virtual images is								
	24.	Three resistors of values 2Ω , 4Ω and 6Ω are connected in parallel. The resultant resistance is								
	25.	The S.I. unit of magnetic field induction is								
III.	Mat	atch the following by writing the letter of the correct answer in the ackets, choosing from Group B . $5 \times \frac{1}{2} = 2$								
•		Gre	oup 'A'			1	Group B'			
	26.	Pla	ster of Paris	[]	(A)	$CaOCl_2$			
	27.	Gyr	osum	[.].	(B)	$NaHCO_3$			
	28.	Ble	aching powder	[]	(C)	$Na_2CO_3 \cdot 10H_2O$			
	29.	Bal	king soda	[]	(D)	$CaSO_4 \cdot 1/2 \ H_2O$			
	30.		shing soda	[]	(E)	$CaSO_4 \cdot 2H_2O$			
						(F) (G)	$CuSO_4 \cdot 5H_2O$ NaCl			
48					- 14	1				