

1. Heat

1. Latent heat of fusion for ice is _____.
2. S.I Unit of specific heat is _____.
3. Specific heat _____.
4. Latent heat of vaporization of water is _____.
5. The process of converting solid into liquid is called _____.
6. The amount of a water vapor present in air is called _____.
7. _____ is the reverse process of evaporation.
8. Evaporation is a _____ phenomenon.
9. Conservation of steam into liquid is called _____.
10. The water droplets condensed on cold surfaces is called _____.
11. 1 Calorie = _____ joule
12. The temperature of a steel rod is 330K. Its temperature °C is _____.
13. _____ is used as a coolant.
14. Rate of evaporation depends on _____, _____, _____.
15. Which of the following is a warming process _____? ()
a) Evaporation b) Condensation c) Boiling d) All the above
16. The temperature of a steel rod is 330K. Its temperature °C is _____. ()
a) 45°C b) 57°C c) 59°C d) 63°C
17. Specific heat S = _____. ()
a) $\frac{Q}{\Delta t}$ b) $Q \Delta t$ c) $\frac{Q}{m \Delta t}$ d) $\frac{m \Delta t}{Q}$
18. _____ is a cooling process. ()
a) Boiling b) Evaporation c) Condensation d) All the above
19. _____ is used as a coolant. ()
a) Benzene b) Kerosene c) Grease d) Water
20. 1 Calorie = _____ joule ()
a) 4.186 b) 0.45 c) 41.86 d) 0.0418

21. Which one has highest specific heat? ()
 a) Benzene b) Lead c) Water d) Kerosene
22. Which of the following is surface phenomenon? ()
 a) Evaporation b) Condensation c) Freezing d) Melting
23. Rate of evaporation depends on _____ ()
 a) Surface Area b) Humidity c) Temperature d) All the above
24. The phase changes from gas to liquid is called _____ ()
 a) Boiling b) Evaporation c) Condensation d) Humidity

Answers

- 1) 80 cal / gm 2) J Kg⁻¹k⁻¹ 3) $\left(s = \left[\frac{Q}{m\Delta t} \right] \right)$
- 4) 540 cal/gm 5) Melting 6) Humidity
- 7) Condensation 8) Cooling 9) Condensation
- 10) Dew 11) 4.186 12) 57°C
- 13) Water 14) Surface Area, Humidity, Temperature
- 15) b 16) b 17) c
- 18) b 19) d 20) a
- 21) c 22) a 23) d
- 24) c