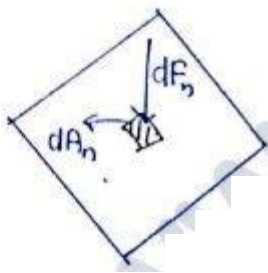


Fluid static

Pressure Measurement:-

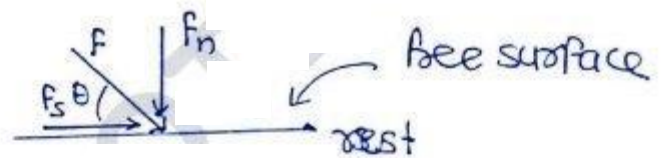
Pressure :- Normal force per unit area exerted by the fluid known as pressure.

pressure is compressive in nature as per the pascal when the fluid is at rest the pressure has no direction i.e. infinite direction. so pressure is scalar in nature.



$$P = \frac{dF_n}{dA} \quad \frac{N}{m^2} = Pa$$

rest
 $P = (-\sigma)$
 $\tau = 0$



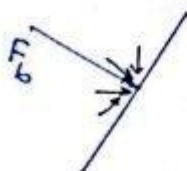
$$\begin{aligned} F_n &= P \sin \theta \\ F_s &= F \cos \theta \end{aligned} \quad \left\{ \begin{array}{l} \text{it is not necessary} \\ \text{that there will be no} \\ \text{shear stress but} \\ \text{shear stress must be 0} \end{array} \right.$$

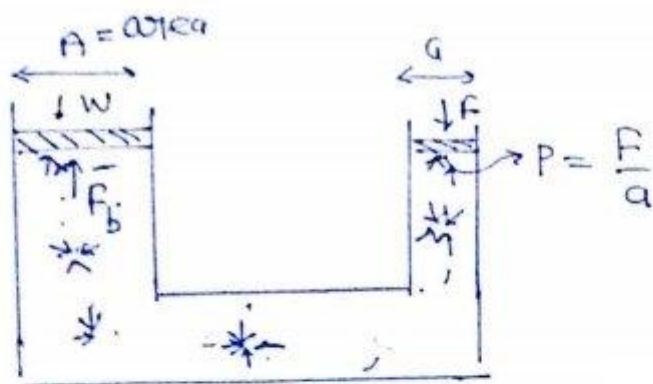
free surface $\tau = 0$

$$F_s = 0, \cos \theta = 0$$

$$\theta = 90^\circ$$

* Always act $\perp$ r





$$F_p = P \times A$$

$$F_p = W$$

$$P \times A = W$$

$$\frac{F}{a} \times A = W$$

$$\Rightarrow \frac{W}{F} = \frac{A}{a}$$

$$A > a$$

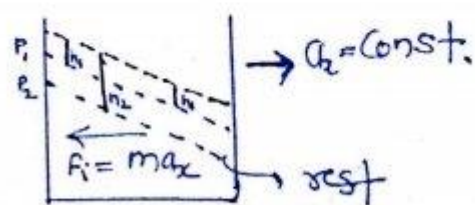
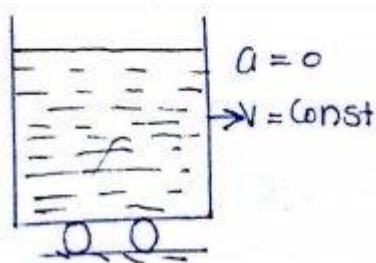
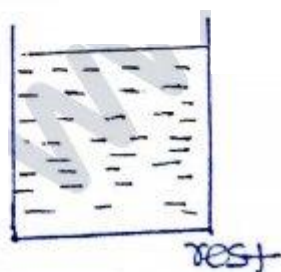
$$\boxed{W > F}$$

Validity of Pascal's law: →

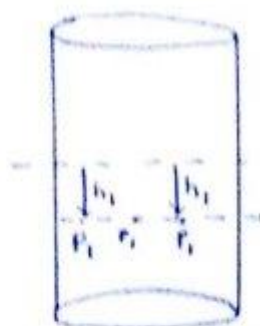
(i) It is valid for Compressible, incompressible, viscous, non viscous fluid but the fluid must be at rest.

(ii) If the fluid in a container, container is moving with constant velocity, uniform accⁿ and constant angular velocity.

(iii) Fluid is ideal.



P_1, P_2 Const.



Pressure at every radius will be same

$$P \neq f(r)$$

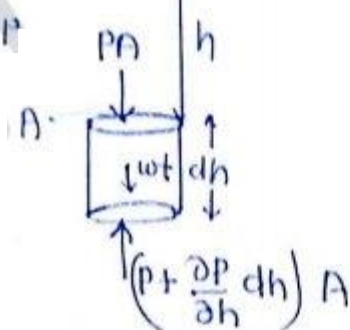
$$P = f(z)$$



const pressure line

Hydrostatic Load:-

our convention



$$\vec{a} = a_x \hat{i} + a_y \hat{j} + a_z \hat{k}$$

$\downarrow \quad \downarrow \quad \downarrow$
 $x \quad y \quad z$

$$\sum F_{net} = m \vec{a}$$

$$PA + wt - \left(P + \frac{\partial P}{\partial h} dh \right) A = 0$$

$$PA + \omega \times A dh - P A - \left(\frac{\partial P}{\partial h} \right) A dh$$

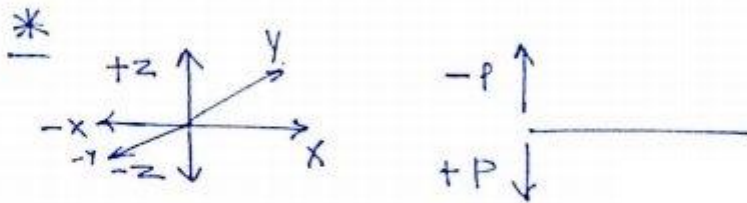
$$\frac{\partial P}{\partial h} = \omega = \rho g$$

Incompressible fluid

$$\rho \neq f(P)$$

$$dP = \rho g dh$$

$$dP \propto dh$$

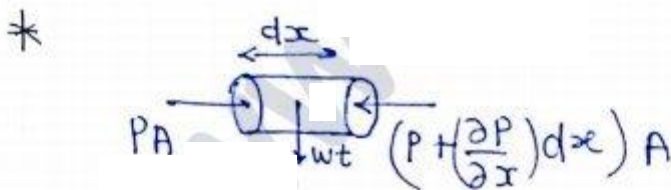


$$dP \propto (-dz)$$

↓ ↓
+ve -(-ve)

$$dP = -\rho g dz$$

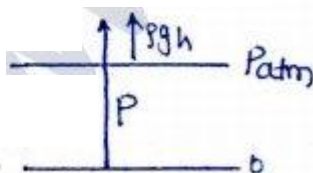
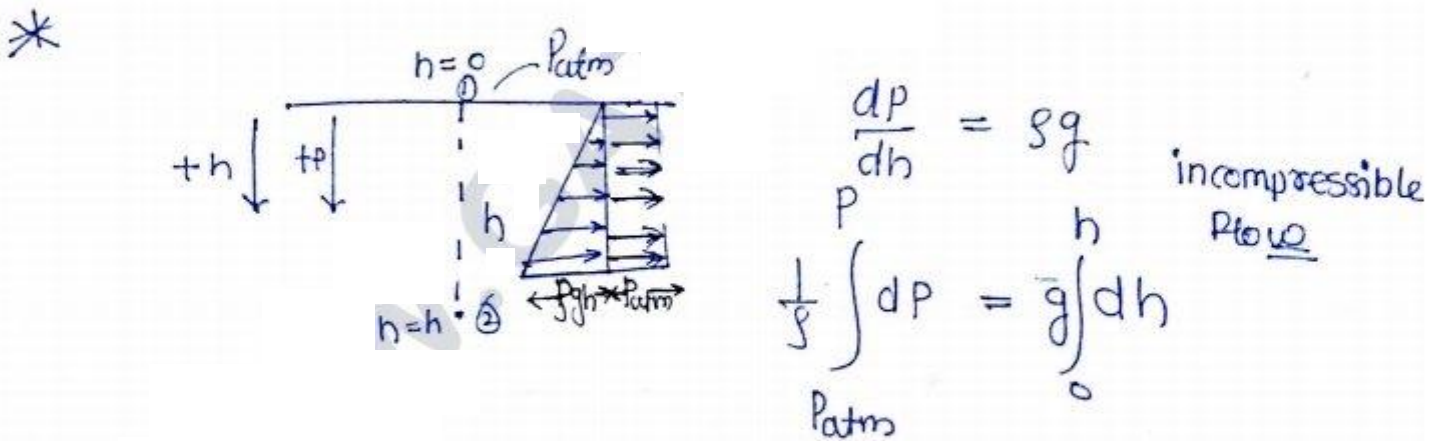
$$\boxed{\frac{dP}{dz} = -\rho = -\rho g}$$



$$PA - \left(P + \left(\frac{\partial P}{\partial x} \right) dx \right) A = 0$$

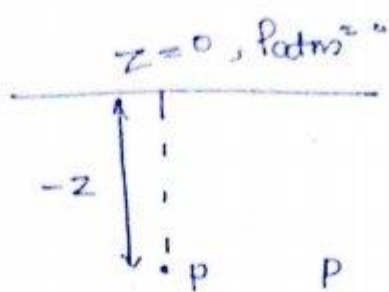
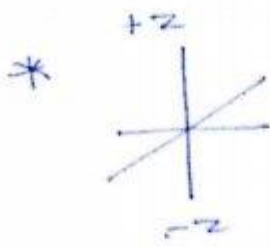
$$\frac{\partial P}{\partial x} = 0$$

Pressure remains constant at Datum.



$$\boxed{P = P_{atm} + \rho gh} \quad \text{Absolute}$$

$$\boxed{P = \rho gh} \quad \text{Gauge. (P}_{atm} = 0)$$



$$\frac{dp}{dz} = -\omega \Rightarrow \int_0^p dp = -\rho g \int_0^{-z} dz$$

~~$P = \rho g h$~~ ~~$P = \rho g h$~~ Gauge

$P = \rho g z$ Gauge pressure

Gas (in Compressible fluid)

$$dp = \rho g dh$$

$$dp = \frac{\rho}{\rho T} g dh$$

$$\int_{P_{atm}}^P \frac{dp}{\rho} = \int_0^h \frac{g}{\rho T} dh$$

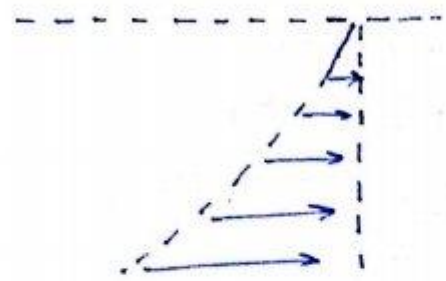
$$\ln \left(\frac{P}{P_{atm}} \right) = + \frac{gh}{RT}$$

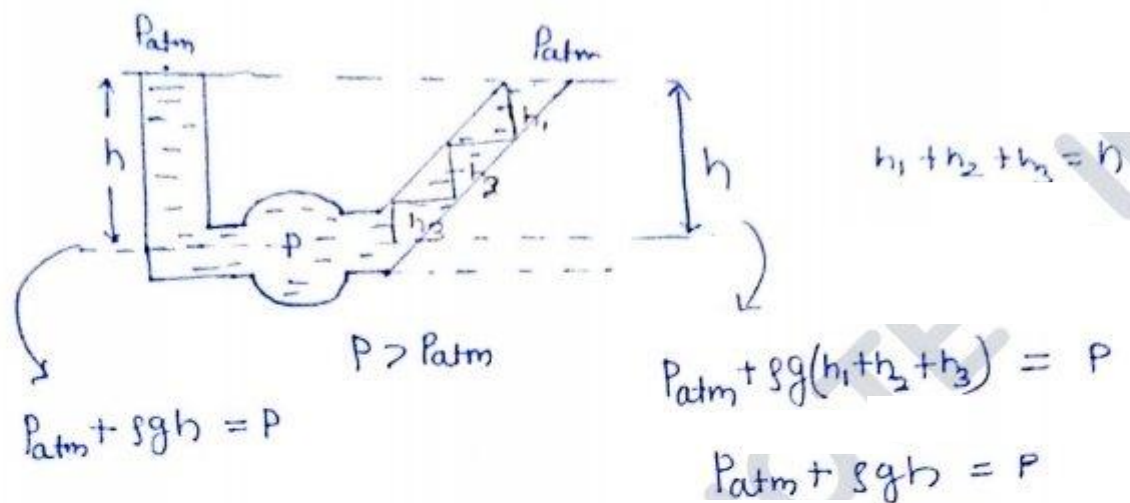
*
* $P = P_{atm} e^{\frac{gh}{RT}}$

ideal Gas

$$P = \rho R T \rightarrow \text{Const.}$$

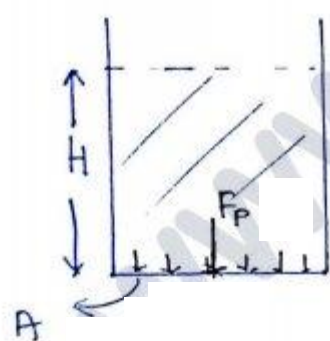
$$\rho = \frac{P}{RT}$$



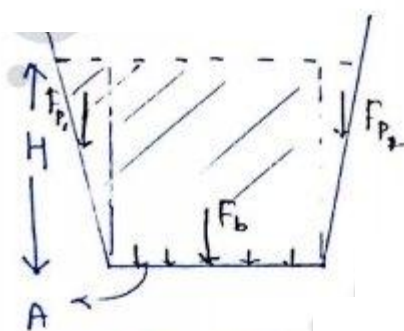


Pressure remain same on datum.

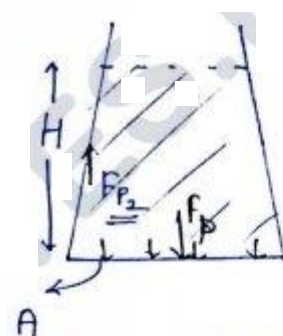
Hydrostatic Paradox:-



$$\begin{aligned}
 W_t &= \rho \times \text{Vol.} \\
 W_t &= \rho A H (\downarrow) \\
 F_b &= P A = \rho g H A (\downarrow)
 \end{aligned}$$



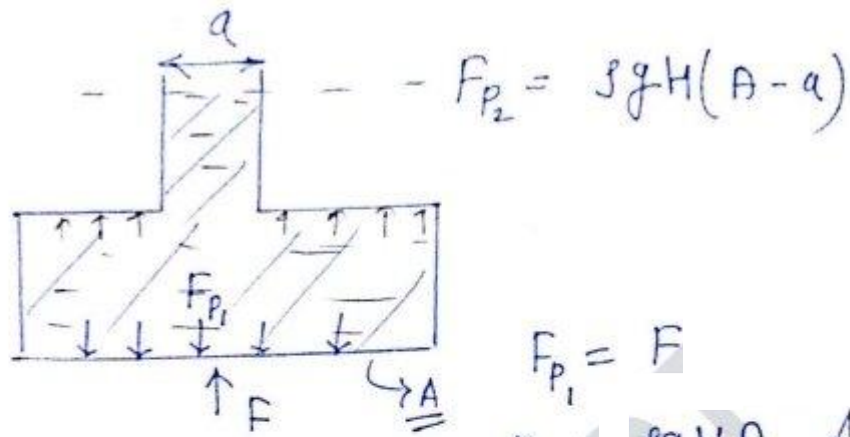
$$\begin{aligned}
 F_b &= P \times A \\
 &= \rho g H A \\
 W_t &> F_b \\
 \downarrow F_{p1} + \downarrow F_b &= W_t
 \end{aligned}$$



$$\begin{aligned}
 F_p &= \rho g H A \\
 W_t &< F_p \\
 W_t &= \downarrow F_p - \uparrow F_{p2}
 \end{aligned}$$

* Hydrostatic force depends on the depth of container and surface area only for the given fluid.

Q.3

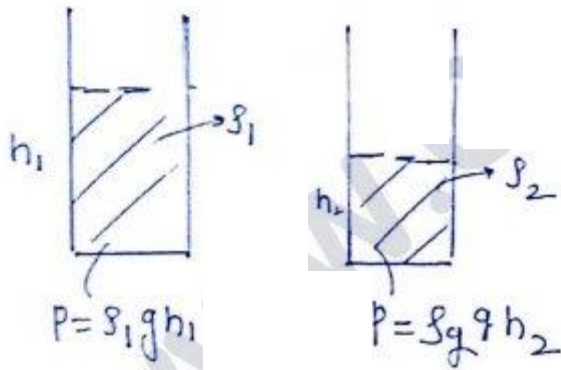


$$F_{P1} = F$$

$$F = \rho g H A$$

Conversion of one liquid column to another:-

$$P = P$$



$$\rho_1 g h_1 = \rho_2 g h_2$$

$$\rho_1 h_1 = \rho_2 h_2$$

$$h_2 = \frac{\rho_1}{\rho_2} h_1$$

$$\rho_1 g h_1 = \rho_2 g h_2$$

$$\omega_1 h_1 = \omega_2 h_2$$

$$h_2 = \frac{\omega_1}{\omega_2} h_1$$

$$\frac{\rho_1 h_1}{\rho_w} = \frac{\rho_2 h_2}{\rho_w}$$

$$\rho_1 h_1 = \rho_2 h_2$$

$$h_2 = \frac{\rho_1 h_1}{\rho_2}$$

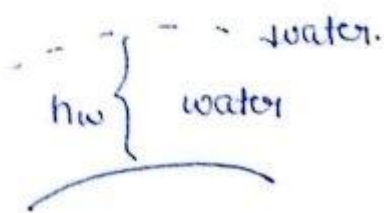
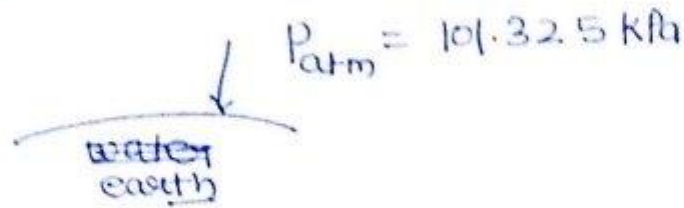
if $\rho_2 = 1$ (water.)

$$h_2 = \rho_1 h_1$$

Atmospheric Pressure:

The pressure exerted by the environmental mass on earth surface is known as atmospheric pressure.

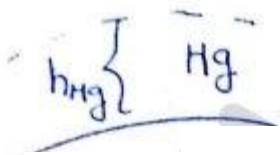
$$P_{\text{atm}} = 101.325 \text{ kPa}$$



$$\rho_w h_w = P_{\text{atm}}$$

$$9810 \times h_w = 101.325 \times 10^3 \text{ Pa}$$

$$h_w = 10.3 \text{ m of water}$$



$$\rho_{\text{Hg}} h_{\text{Hg}} = 101.325 \times 10^3$$

$$h_{\text{Hg}} = \frac{101.325 \times 10^3}{13.6 \times 9810}$$

$$h_{\text{Hg}} = 760 \text{ mm of Hg}$$

$$h_{\text{Hg}} = 76 \text{ cm of Hg}$$

Units of Pressure:-

$$P = \frac{F}{A} \quad \frac{\text{N}}{\text{m}^2} \text{ (S.I.)}$$

$$P_{\text{atm}} = 101.325 \times 10^3 \text{ Pa}$$

$$\frac{\text{N}}{\text{m}^2} = \text{Pa}$$

$$10^3 \frac{\text{N}}{\text{m}^2} = \text{kPa}, \quad 10^6 \frac{\text{N}}{\text{m}^2} = \text{MPa}$$

$$P_{atm} = 101.325 \times 10^3 Pa = 101325 \times 10^5 Pa$$

$$\approx 1 \times 10^5 Pa \approx 1 \times 10^5 Pa$$

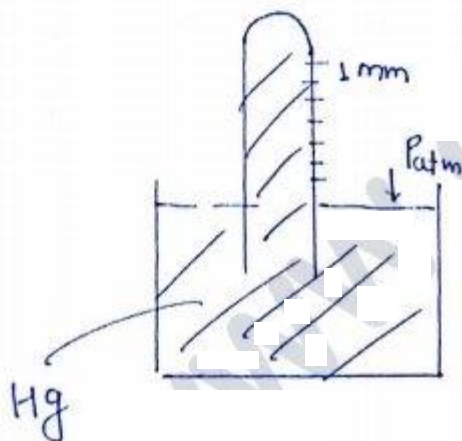
$$10^5 Pa = \frac{kgf}{cm^2}$$

$$10^5 Pa = 1 bar$$

$$1 atm = 1 bar$$

$$1 bar = \frac{kgf}{cm^2}$$

Barometry. (By Torricelli)



$$1 Torr = 1 mm \text{ of Hg}$$

$$1 Torr = 10^{-3} m \text{ of Hg}$$

$$= 10^{-3} \times 13.6 \times 9810$$

$$1 Torr = 133.33 Pa$$

F.P.S. (Foot Pounds Sec.)

$$Psi = \frac{Lbf}{(inch)^2}$$

Pounds Force per sq. inch

Pounds Force

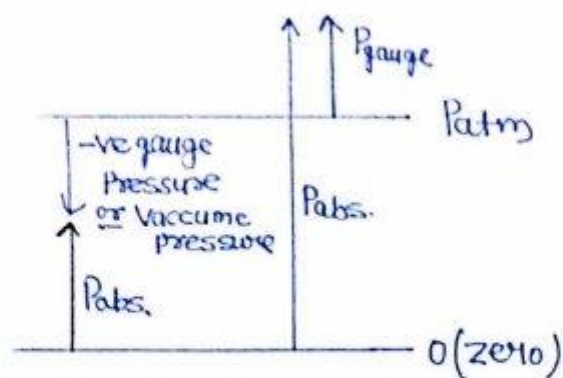
$$Lbf = 0.453 kgf$$

$$Lbf = 0.453 \times 9.81 N$$

$$inch = 2.54 cm$$

$$1 atm = 14.7 Psi$$

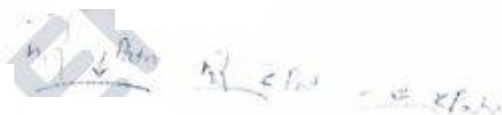
Absolute Pressure, Gauge Pressure, -ve gauge Pressure
Vacuum pressure.



eg -30 kPa

30 kPa Vacuum

30 kPa (-ve gauge pressure)



There may be positive gauge pressure or negative gauge pressure but there can not be negative absolute pressure.

Gauges are calibrated as per their local envt.

Pressure measure from atmospheric pressure is gauge pressure and pressure ~~pre~~ measured from zero pressure is absolute pressure.

Classification of pressure measurement device.

① on the basis of liq. column deflection (Manometers)

(i) simple manometers

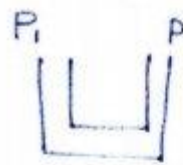


(a) Piezometers

(b) U-tube manometer

(c) single column manometers

(ii) Differential Manometers



(a) U-tube differential manometer

(b) Inverted tube manometer.

(2) on the basis of elastic properties. (Mechanical Gauge)

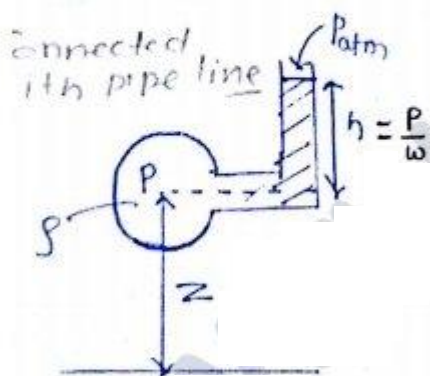
↓
Bourdon tube
Gauge

Piezometers:- It is the simple manometer at one end open to atmosphere and another end is connected where pressure is to measure.

Disadvantages: (i) Can not measure Gases pressure

(ii) Can not measure Vacuum pressure

(iii) Gives large deflection at high pressure for lighter liquids so moderate pressure should be measure.



$$P - \rho g h = P_{atm}$$

$$P = P_{atm} + \rho g h$$

→ Abs.

$$P_{atm} = 0$$

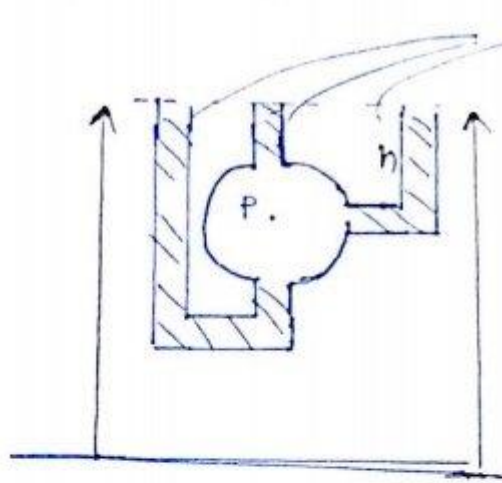
$$P = \rho g h \text{ Gauge}$$

$$P = \omega h$$

$$h = \frac{P}{\omega}$$

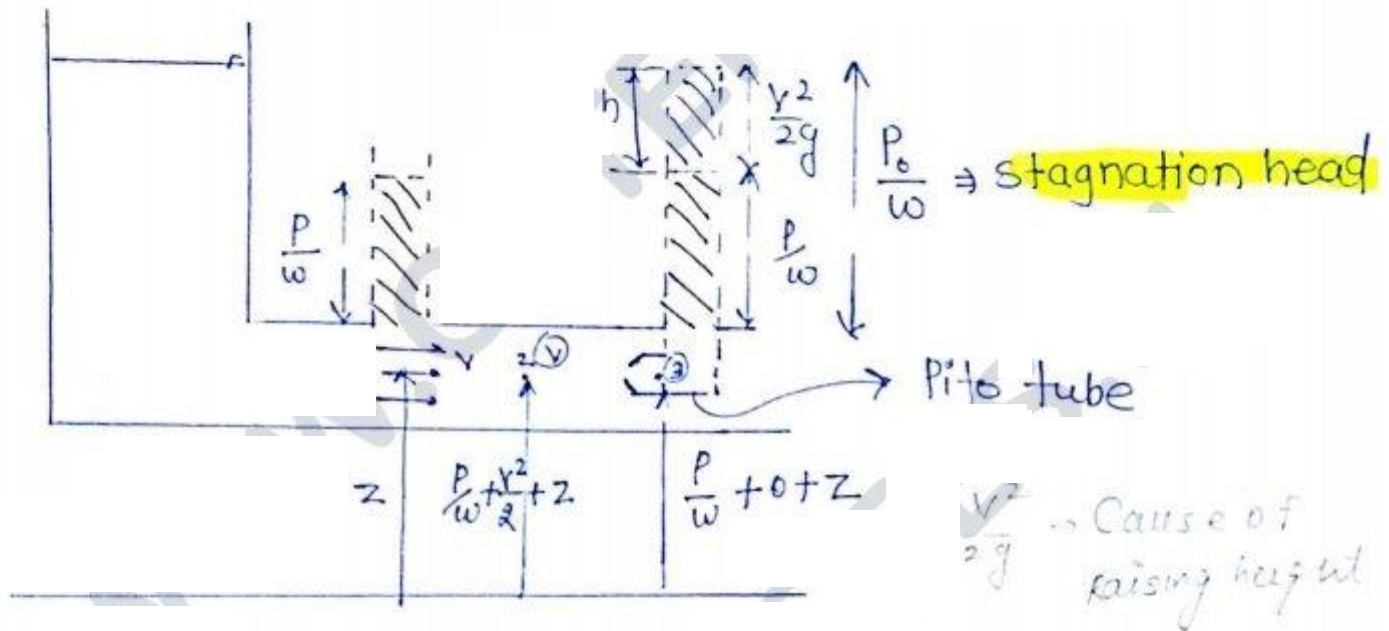
*** Piezometric head = $\frac{P}{\omega} + Z$**

static head datum head



All Gives Same Reading

Piezometer Gives same reading from datum



$$h = \frac{V^2}{2g} \Rightarrow V = \sqrt{2gh}$$

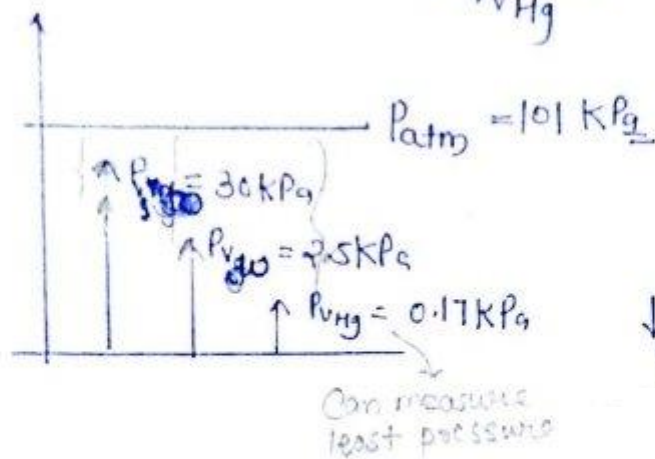
Vapour Pressure: → Pressure exerted (partial pressure) by the saturated vapour in the close container on the liquid surface at given temperature is known as saturated vapour pressure.

at 20°C $P_{\text{water}} = 2.5 \text{ kPa}$

100°C $P_{\text{w}} = 100 \text{ kPa} = 1 \text{ atm}$

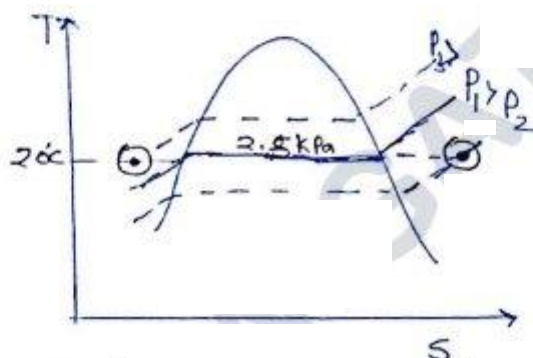
at 20°C Gasoline $P_{Vg} = 30 \text{ kPa}$

$$P_{VHg} = 0.17 \text{ kPa}$$



20°C

$$\downarrow \frac{P}{\rho} + \frac{V^2}{2g} = c \quad \begin{matrix} V_{\max} \\ P_{\min} \end{matrix} \quad \uparrow \frac{P}{\rho} + \frac{V^2}{2g} = c$$



→ at 20° above P-Vapour
sat liquid

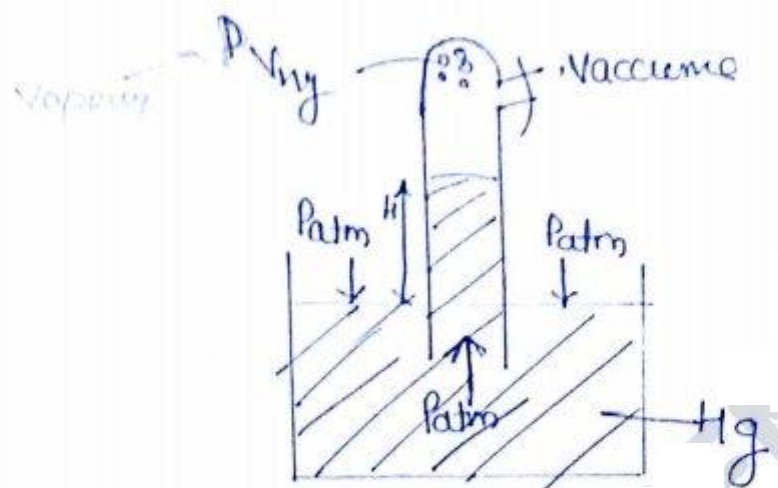
→ at 20° below P-Vapour
saturated steam.

* at const. temp if P below P_{vapour}
then fluid (eg, water) evaporate
bubble forms

Cavitation: In the fluid if the abs pressure of fluid becomes less than corresponding temp. Vapour pressure the bubbles are formed and if these bubbles comes in a region of high pressure the bubbles gets collapse the pitting, cavitation phenomena takes place it errod the surface.

eg. at succion side of pump, exit of reaction turbine etc.

Barometer:- it is an instrument which measure local atmospheric pressure



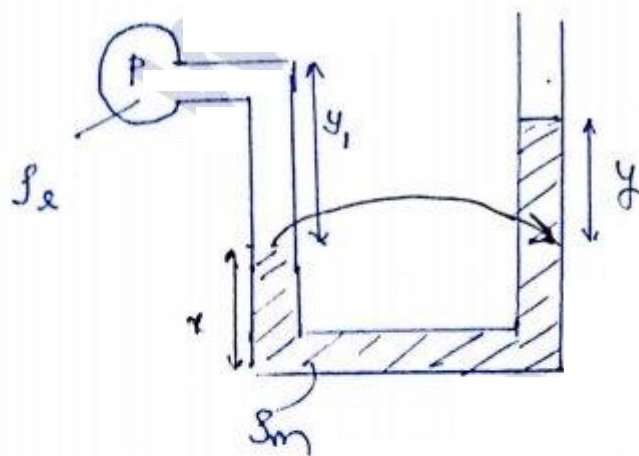
$$P_{vHg} + \omega_{Hg} H = P_{atm}$$

$$\downarrow 0$$

$$P_{atm} = \omega_{Hg} H$$

* Mercury (Hg) used in barometer because of **high density** and **low Vapour pressure**.

U-Tube manometer:->



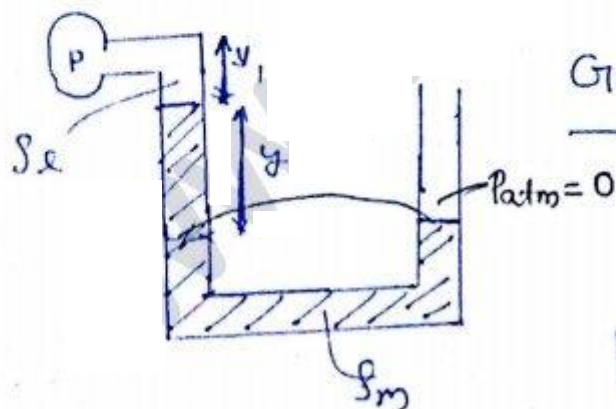
$$P + S_e g y_1 + S_m g x - S_m g x - S_m g y = P_{atm}$$

$$P = P_{atm} + S_m g y - S_e g y_1$$

↳ Absolute

$$P_{atm} = 0$$

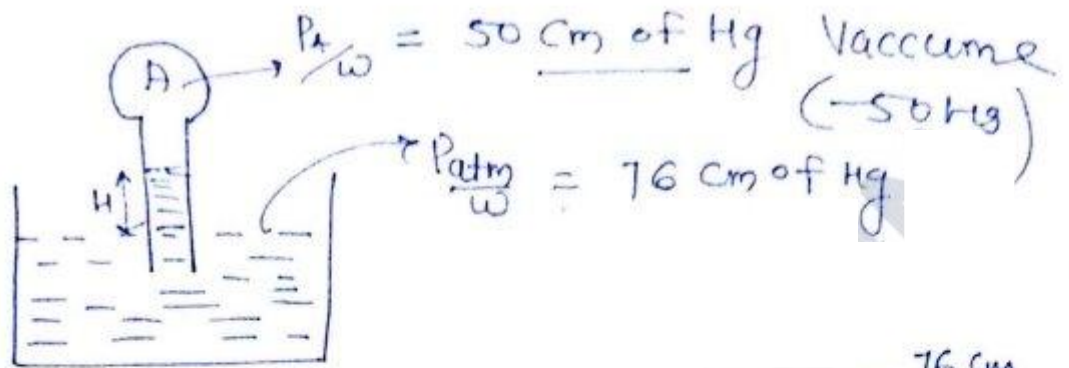
$$\text{Gauge} \Rightarrow P = S_m g y - S_e g y_1$$



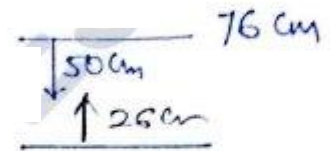
$$0 - S_m g y - S_e g y_1 = P$$

$$P = -S_m g y - S_e g y_1 \quad \text{Gauge}$$

Q.8
0.12



$$\frac{P_A}{w} + H = 76$$



Atmos. Pressure $26 + H = 76$

$$H = 50 \text{ cm of Hg } \underline{\underline{Ans}}$$

or Consider $P_{atm} = 0$

$$0 - H = -50$$

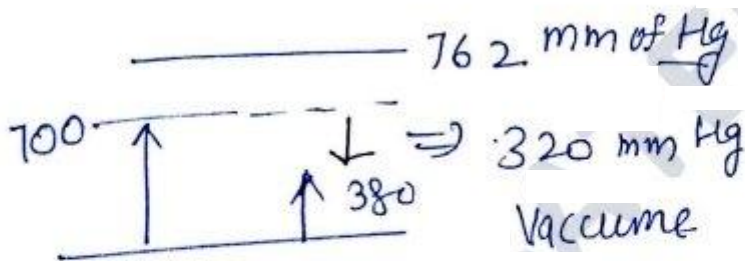
$\Rightarrow$ Gauge

$$H = 50 \text{ cm}$$

Q.13

$$P_{atm} = 762 \text{ mm of Hg}$$

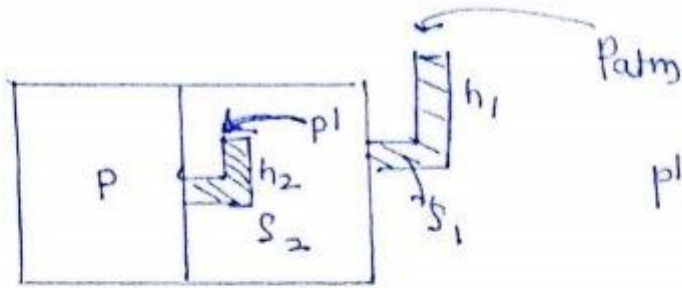
700 mm local envt. reading.



$$(P)_{Vacc} = 700 - 380 = 320 \text{ mm Hg Vacuum}$$

2.14

*

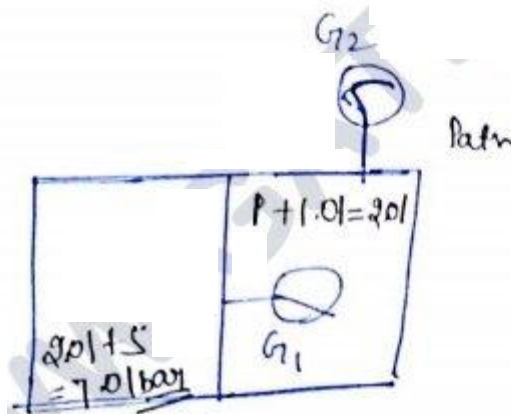


$$P^1 = P_{atm} + S_1 g h_1$$

$$P = P^1 + S_2 g h_2$$

$$P = P_{atm} + S_1 g h_1 + S_2 g h_2$$

Q. 19



$$P_{G1} = 5 \text{ bar}$$

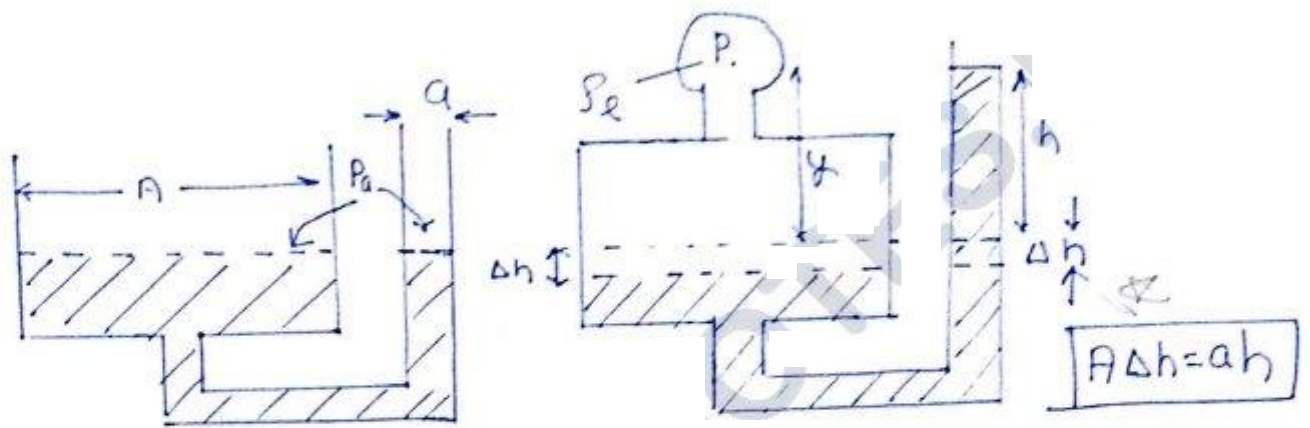
$$P_{G2} = 4 \text{ bar}$$

$$P_{atm} = 1.01 \text{ bar}$$

$$P = 1.01 + 1 + 5$$

$$\underline{\underline{P = 7.01 \text{ bar}}}$$

Single Column manometer:—



$$P + \rho_l g y + \rho_l g \Delta h - \rho_m g \Delta h - \rho_m g h = P_{atm}$$

Q14

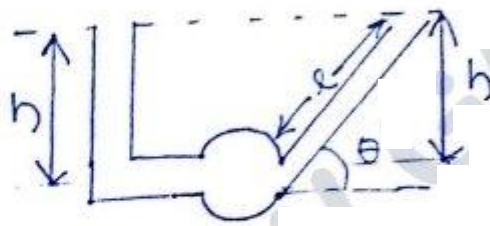
$$A = 500a$$

$$\text{error} = (h + \Delta h) - h$$

$$A\Delta h = ah$$

$$\% \text{ error} = \frac{\Delta h}{h} \times 100 = \frac{a}{A} \times 100 = \frac{100}{500} = \underline{\underline{0.2\%}}$$

Sensitivity of manometer:—

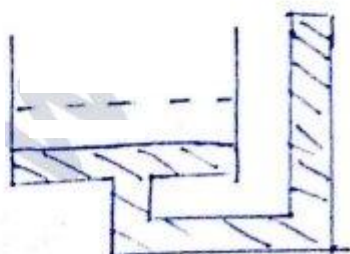


$$l \sin \theta = h$$

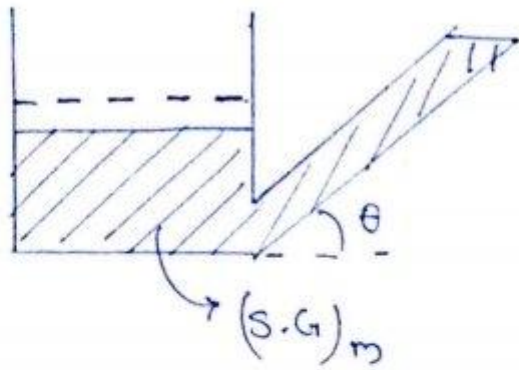
$$\text{sensitivity } (S \propto \frac{l}{h})$$

$$S \propto \frac{1}{\sin \theta}$$

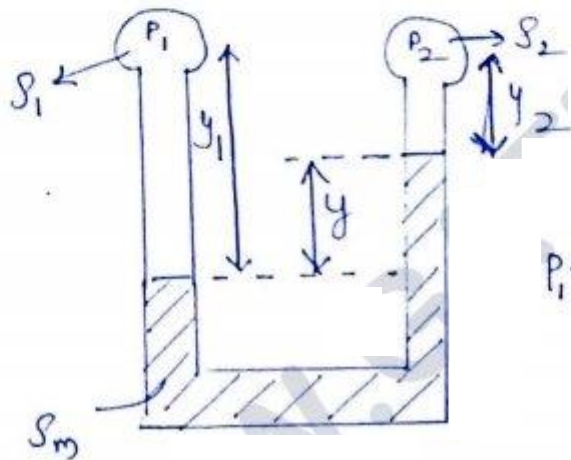
$$\theta \downarrow \rightarrow \sin \theta \downarrow \rightarrow S \uparrow$$



$$S \propto \frac{1}{(\rho_l/A)}, \quad S \propto \frac{1}{(\rho_m/g)}$$



For Differential manometer

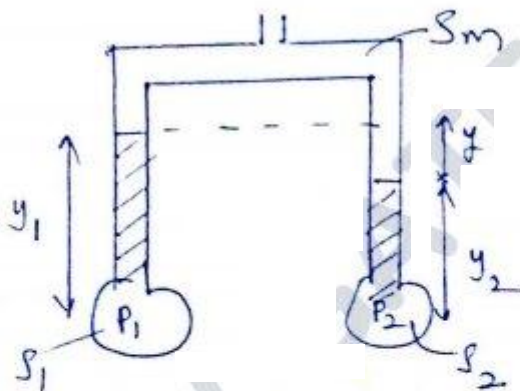


$$P_1 + s_1 g y_1 - s_m g y - s_2 g y_2 = P_2$$

$$P_1 - P_2 = s_m g y - s_1 g y_1 + s_2 g y_2$$

Inverted tube manometry

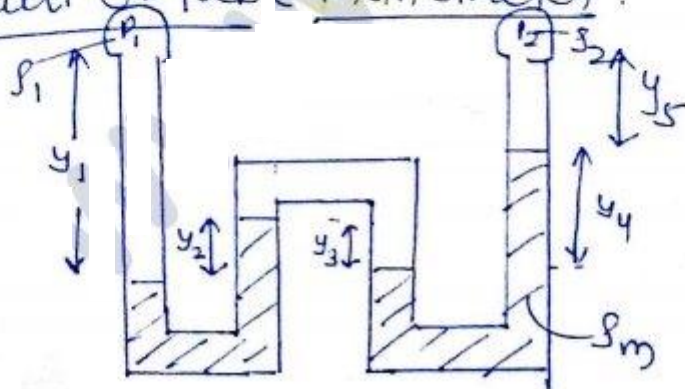
used for to measure low pressure diff.



$$P_1 - s_1 g y_1 + s_m g y + s_2 g y_2 = P_2$$

$$P_1 - P_2 = -s_m g y + s_1 g y_1 - s_2 g y_2$$

Multi U-tube manometry :- for high pressure diff



$$P_1 - P_2 = -s_1 g y_1 + s_m g y_2 - s_m g y_3 + s_m g y_4 + s_2 g y_5$$