

# JCECE ENGINEERING ENTRANCE EXAM.

**SOLVED PAPER 2006**

## Physics

1. Which of the following rays can be polarised?

- (a) Water wave and sound wave
- (b) Sound wave and radio wave
- (c) X-rays and water wave
- (d) Light wave and X-ray

2. Quantum nature can prove :

- (a) interference
- (b) photoelectric effect
- (c) diffraction
- (d) polarisation

3. Which one has highest binding energy per nucleon?

- (a)  $\text{Fe}^{56}$
- (b)  $\text{Li}^6$
- (c)  $\text{U}^{235}$
- (d)  $\text{Ca}^{40}$

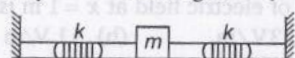
4. Huygen's wave theory can't explain :

- (a) interference
- (b) photoelectric effect
- (c) diffraction
- (d) all of these

5. If the refractive index of a glass prism is  $\cot(A/2)$  and  $A$  is angle of prism, then angle of minimum deviation is :

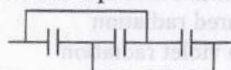
- (a)  $\left(\frac{\pi}{2} - A\right)$
- (b)  $\left(2\pi - \frac{A}{2}\right)$
- (c)  $\left(\frac{\pi - A}{2}\right)$
- (d)  $(\pi - 2A)$

6. If  $f$  is the frequency when mass  $m$  is attached to a spring of spring constant  $k$ , then new frequency for this arrangement, is :



- (a)  $f/2$
- (b)  $\sqrt{2}f$
- (c)  $f/\sqrt{2}$
- (d)  $2\sqrt{2}f$

7. What is equivalent capacitance of the network? Each capacitor has  $1\mu\text{F}$  capacitance :



- (a)  $\frac{1}{3}\mu\text{F}$
- (b)  $2\mu\text{F}$
- (c)  $\frac{3}{2}\mu\text{F}$
- (d)  $3\mu\text{F}$

8. The two capacitors  $C_1$  and  $C_2$  are charged to potentials  $V_1$  and  $V_2$  and then connected in parallel. There will be no flow of energy, if :

- (a)  $C_1 V_1 = C_2 V_2$
- (b)  $V_1 = V_2$
- (c)  $C_1 = C_2$
- (d)  $\frac{C_1}{V_1} = \frac{C_2}{V_2}$

9. Which is not the unit of electric field?

- (a)  $\frac{\text{N}}{\text{C}}$
- (b)  $\frac{\text{N} \cdot \text{m}}{\text{C}}$
- (c)  $\frac{\text{V}}{\text{m}}$
- (d)  $\frac{\text{J}}{\text{C} \cdot \text{m}}$

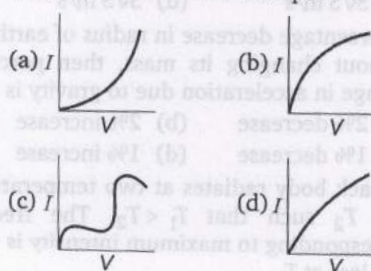
10. If a body moves for 2 s with 15 m/s velocity towards east and then moves with 5 m/s velocity for 8 s towards north, then average velocity is :

- (a) 5 m/s
- (b) 15 m/s
- (c) 30 m/s
- (d) 7.5 m/s

11. For monoatomic gas which is correct?

- (a)  $C_V = \frac{3}{2}R$
- (b)  $C_P = \frac{5}{2}R$
- (c)  $C_P - C_V = 2R$
- (d)  $\frac{C_P}{C_V} = \frac{3}{5}$

12. Which is correct relationship for diode?



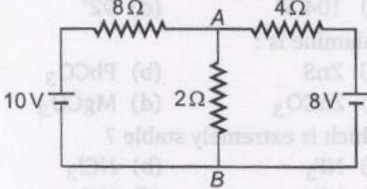
13. In a properly biased transistor :

- (a) both depletion layers are equally large
- (b) both depletion layers are equally small
- (c) emitter-base depletion layer is large but base-collector depletion layer is small
- (d) emitter-base depletion layer is small but base-collector depletion layer is large



14. A wire has resistance  $20\ \Omega$ . If its length is increased three times its initial length, then new resistance is :  
 (a)  $40\ \Omega$  (b)  $80\ \Omega$   
 (c)  $60\ \Omega$  (d)  $180\ \Omega$
15. A circular coil of diameter  $d$  is rotated in electric field such that electric flux is changed from zero to maximum value  $\phi$  then, electric field is :  
 (a)  $\frac{\phi}{\pi d^2}$  (b)  $\frac{2\phi}{\pi d^2}$   
 (c)  $\frac{4d^2}{\pi\phi^2}$  (d)  $\frac{4\phi}{\pi d^2}$
16. Work done in rotating a bar magnet from  $0$  to angle  $\theta$  is :  
 (a)  $MH(1 - \cos\theta)$  (b)  $\frac{M}{H}(1 - \cos\theta)$   
 (c)  $\frac{M}{H}(\cos\theta - 1)$  (d)  $MH(\cos\theta - 1)$
17. If a convex lens of refractive index  $1.44$  is dipped in liquid of refractive index  $1.49$ , then it behaves as :  
 (a) concave lens (b) convex lens  
 (c) mirror (d) none of these
18. If a source approaches and recedes from observer with same velocity, the ratio of frequencies (apparent) is  $6 : 5$ , then velocity of source is : ( $v_s = 330\text{ m/s}$ )  
 (a)  $20\text{ m/s}$  (b)  $10\text{ m/s}$   
 (c)  $30\text{ m/s}$  (d)  $33\text{ m/s}$
19. If a boy swings in a circle so the minimum and maximum height from ground is  $3\text{ m}$  and  $6\text{ m}$ , then, its maximum velocity is :  
 (a)  $5\sqrt{2}\text{ m/s}$  (b)  $2\sqrt{5}\text{ m/s}$   
 (c)  $3\sqrt{5}\text{ m/s}$  (d)  $5\sqrt{3}\text{ m/s}$
20. If percentage decrease in radius of earth is  $1\%$  without changing its mass, then percentage change in acceleration due to gravity is :  
 (a)  $2\%$  decrease (b)  $2\%$  increase  
 (c)  $1\%$  decrease (d)  $1\%$  increase
21. A black body radiates at two temperatures  $T_1$  and  $T_2$  such that  $T_1 < T_2$ . The frequency corresponding to maximum intensity is :  
 (a) less at  $T_1$   
 (b) more at  $T_1$   
 (c) equally in the two cases  
 (d) cannot say
22. If temperature is increased by  $1\text{ K}$  at constant volume, then work done on the gas is :  
 (a)  $\frac{5}{2}R$  (b)  $\frac{3}{2}R$   
 (c) zero (d)  $\frac{1}{2}R$
23. A body cools from  $75^\circ\text{C}$  to  $70^\circ\text{C}$  in time  $t_1$ , from  $70^\circ\text{C}$  to  $65^\circ\text{C}$  in time  $t_2$  and from  $65^\circ\text{C}$  to  $60^\circ\text{C}$  in time  $t_3$ , then :  
 (a)  $t_3 > t_2 > t_1$  (b)  $t_1 > t_2 > t_3$   
 (c)  $t_2 > t_1 = t_3$  (d)  $t_1 > t_2 > t_3$
24. A gas is at  $27^\circ\text{C}$ . Its volume is doubled keeping pressure constant, then final temperature is :  
 (a)  $600^\circ\text{C}$  (b)  $327\text{ K}$   
 (c)  $327^\circ\text{C}$  (d)  $273^\circ\text{C}$
25. If the volume of gas is changed from  $V_1$  to  $V_2$  isothermally, then work done is :  
 (a)  $RT \ln \frac{V_1}{V_2}$  (b)  $RT \ln \frac{V_2}{V_1}$   
 (c)  $R(T_2 - T_1) \ln \frac{V_2}{V_1}$  (d)  $R(V_2 - V_1) \ln \frac{T_2}{T_1}$
26. If energy is supplied to a gas isochorically, increase in internal energy is  $dU$  then :  
 (a)  $dQ = dU + dW$  (b)  $dQ = dU - dW$   
 (c)  $dQ = dU$  (d)  $dQ = -dU$
27. A nucleus  ${}^A_ZX$  emits one  $\alpha$  and  $2\beta$  particles, then final nucleus is :  
 (a)  $Y^{A-2}_Z-2$  (b)  $Y^{A-4}_Z-4$   
 (c)  $Y^{A-4}_Z-4$  (d)  $X^A_Z$
28. The fringe width for red light is approximately how many times that for violet light in Young's slit experiment?  
 (a) 2 times (b) 3 times  
 (c) Equal (d)  $1/2$  times
29. A person sees clearly at a distance of  $100\text{ cm}$ , then power of lens used to see object at  $40\text{ cm}$  is :  
 (a)  $3\text{ D}$  (b)  $-3\text{ D}$   
 (c)  $-1.5\text{ D}$  (d)  $+1.5\text{ D}$
30. The electric potential  $V$  is given as a function of distance  $x$  (metre) by  $V = (5x^2 + 10x - 4)\text{V}$ . Value of electric field at  $x = 1\text{ m}$  is :  
 (a)  $-23\text{ V/m}$  (b)  $11\text{ V/m}$   
 (c)  $6\text{ V/m}$  (d)  $-20\text{ V/m}$
31. In an image convertor tube fluorescent material is bombarded by :  
 (a) visible radiation  
 (b) electron radiation  
 (c) infrared radiation  
 (d) ultra violet radiation
32. A particle executes simple harmonic motion with a frequency  $f$ . The frequency with which its kinetic energy oscillates is :  
 (a)  $f/2$  (b)  $f$   
 (c)  $2f$  (d)  $4f$



33. The work done by the centripetal force  $F$  when the body completes one rotation around the circle of radius  $R$  is :  
 (a)  $2\pi RF$  (b)  $2RF$   
 (c)  $RF$  (d) zero
34. The unit mass having  $\vec{r} = 8\hat{i} - 4\hat{j}$  and  $\vec{v} = 8\hat{i} + 4\hat{j}$  in its angular momentum is :  
 (a) 64 unit in  $-\hat{k}$  direction  
 (b) 64 unit in  $+\hat{k}$  direction  
 (c) 64 unit in  $+\hat{j}$  direction  
 (d) 64 unit in  $+\hat{i}$  direction
35. Which is nuclear fusion direction?  
 (a) Hydrogen to helium  
 (b) Uranium to krypton  
 (c) Hydrogen to water  
 (d) Neutron to proton
36. If an AC produces same heat as that produced by a steady current of 4 A, then peak value of current is :  
 (a) 4 A (b) 1.56 A  
 (c) 5.6 A (d) 1.41 A
37. In LCR circuit  $f = \frac{50}{\pi}$  Hz,  $V = 50$  volt,  $R = 300\Omega$ . If  $L = 1$  H and  $C = 20\mu\text{C}$ , then voltage across capacitor is :  
 (a) zero (b) 20 V  
 (c) 30 V (d) 50 V
38. If two forces each of 2 N are inclined at  $60^\circ$ , then resultant force is :  
 (a) 2 N (b)  $2\sqrt{5}$  N  
 (c)  $3\sqrt{2}$  N (d)  $4\sqrt{2}$  N
39. A block of mass 10 kg is placed on a rough horizontal surface whose coefficient of friction is 0.5. If a horizontal force of 100 N is applied on it, then acceleration of block will be :  
 (a)  $10\text{ m/s}^2$  (b)  $5\text{ m/s}^2$   
 (c)  $15\text{ m/s}^2$  (d)  $0.5\text{ m/s}^2$
40. The potential difference across an instrument in an AC circuit of frequency  $f$  is  $V$  and the current through it is  $I$  such that  $V = 5 \cos 2\pi ft$  volt and  $I = 2 \sin 2\pi ft$  amp. The power dissipated in the instrument is :  
 (a) zero (b) 10 W  
 (c) 5 W (d) 2.5 W
41. If ratio of intensities of interfering waves is 16 : 9, then ratio of maximum to minimum intensity is :  
 (a) 49 : 1 (b) 225 : 81  
 (c) 3 : 1 (d) 9 : 1
42. The power of a lens, a short sighted person uses is  $-2$  D. Find the maximum distance of an object which he can see without spectacles :  
 (a) 25 cm (b) 50 cm  
 (c) 100 cm (d) 10 cm
43. The first overtone frequency of a wave on string of length 2 m is 250 Hz. Then, its velocity is :  
 (a) 1000 m/s (b) 25 m/s  
 (c) 500 m/s (d) 10 m/s
44. What is the current flowing in arm AB ?  
  
 (a)  $\frac{35}{4}$  A (b)  $\frac{13}{7}$  A  
 (c)  $\frac{5}{7}$  A (d)  $\frac{7}{5}$  A
45. A projectile is fired making an angle  $2\theta$  with horizontal with velocity 4 m/s. At any instant it makes an angle  $\theta$ , then its velocity is :  
 (a)  $4 \cos \theta$   
 (b)  $4(2 \cos \theta - \sec \theta)$   
 (c)  $2(\sec \theta + 4 \cos \theta)$   
 (d)  $4(\sec \theta + \cos \theta)$
46. If path difference becomes  $(2n - 1)\frac{\lambda}{2}$  then :  
 (a) white fringe is formed  
 (b) bright fringe is formed  
 (c) uniform illumination is obtained  
 (d) dark fringe is formed
47. If the intensity of fringe at wavelength  $\lambda$  is  $K$ , then its intensity at wavelength  $\lambda/2$  is :  
 (a)  $\frac{K}{2}$  (b)  $K$   
 (c) zero (d)  $\sqrt{2}K$
48. A positively charged particle moving with velocity  $v$  enters a region of space having a uniform magnetic field  $B$ . The particle will experience the large deflecting force, when the angle between  $v$  and  $B$  is :  
 (a)  $0^\circ$  (b)  $45^\circ$   
 (c)  $90^\circ$  (d)  $180^\circ$



49. In a step-up transformer the turn ratio is 1 : 8. A lead accumulator (emf = 6 V) is connected across the primary coil of the transformer. The voltage across the secondary coil is :

- (a) 48 V (b) 0.75 V  
(c) 14 V (d) zero

50. The mass of a lift is 500 kg. When it ascends with an acceleration of  $2 \text{ m/s}^2$ , the tension in the cable will be : ( $g = 10 \text{ m/s}^2$ )

- (a) 6000 N (b) 5000 N  
(c) 4000 N (d) 1000 N

## Chemistry

1. A water molecule can form maximum number of H-bond which is equal to :

- (a) 1 (b) 2  
(c) 3 (d) 4

2. Bond angle in  $\text{H}_2\text{O}$  is :

- (a)  $109^\circ 28'$  (b)  $107^\circ 10'$   
(c)  $104.5^\circ$  (d)  $92^\circ$

3. Calamine is :

- (a) ZnS (b)  $\text{PbCO}_3$   
(c)  $\text{ZnCO}_3$  (d)  $\text{MgCO}_3$

4. Which is extremely stable ?

- (a)  $\text{NF}_3$  (b)  $\text{NCl}_3$   
(c)  $\text{NBr}_3$  (d)  $\text{NH}_3$

5.  $\text{CH}_3\text{COCl}$  does not react with :

- (a) diethyl ether (b) phenol  
(c) ethanol (d) aniline

6. In  $\text{SO}_2$  hybridisation is :

- (a)  $sp$  (b)  $sp^3$   
(c)  $dsp^2$  (d)  $sp^2$

7. Lowest melting point chloride is :

- (a) LiCl (b) NaCl (c) KCl (d) CsCl

8. Half-life of a substance is 6 min. If its initial amount is 32 g, then amount present after 18 min is:

- (a) 4 g (b) 8 g (c) 16 g (d) 2 g

9. If calcium acetate and calcium formate react, then product formed is :

- (a) acetaldehyde (b) acetic acid  
(c) formic acid (d) ethyl formate

10. Reduction with aluminium isopropoxide in excess of isopropyl alcohol is called Meerwein Ponderf Verley reduction (MPV). What will be the final product when cyclohex-2-enone is selectively reduced in MPV reaction?

- (a) Cyclohexanol (b) Cyclohex-2-enol  
(c) Cyclohexanone (d) Benzene

11. If pH of a solution is 4, then  $\text{H}^+$  is :

- (a)  $10^4$  (b)  $10^{10}$   
(c)  $10^{-4}$  (d)  $10^{-10}$

12. When sodium nitrate is heated above  $600^\circ\text{C}$ , then :

- (a) only  $\text{Na}_2\text{O}$  is formed  
(b) only  $\text{N}_2$  is formed  
(c) only  $\text{O}_2$  is formed  
(d) all are formed

13. Which of the following produces  $\text{Cl}_2$  gas?

- (a)  $\text{NaCl} + \text{HNO}_3$  (b)  $\text{MnO}_2 + \text{HCl}$   
(c)  $\text{KMnO}_4 + \text{HCl}$  (d)  $\text{HCl} + \text{HNO}_3$

14. Which is correctly arranged as increasing size?

- (a)  $\text{F} < \text{O} < \text{C} < \text{Cl} < \text{Br}$   
(b)  $\text{C} < \text{O} < \text{F} < \text{Cl} < \text{Br}$   
(c)  $\text{Cl} < \text{Br} < \text{F} < \text{C} < \text{O}$   
(d)  $\text{O} < \text{F} < \text{C} < \text{Cl} < \text{Br}$

15.  $\text{NH}_3$  is absorbed by :

- (a) ozone (b) CaO  
(c) pyragallol (d)  $\text{CaCl}_2$

16. 1.25 g  $\text{NH}_3$  contains how many atoms?

- (a)  $10^{23}$  (b)  $2 \times 10^{23}$   
(c)  $6 \times 10^{13}$  (d)  $4 \times 10^{23}$

17. Which of the following has smallest bond angle?

- (a) Ethane (b) Ethene  
(c) Ethyne (d) Ethanol

18. Chloroform in air is oxidised to :

- (a)  $\text{CCl}_4$  (b) dichloromethane  
(c) phosgene (d) oxygen

19. Gypsum is :

- (a)  $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$   
(b)  $\text{CaSO}_4 \cdot \frac{1}{2}\text{H}_2\text{O}$   
(c)  $\text{MgSO}_4 \cdot 2\text{H}_2\text{O}$   
(d)  $\text{CuSO}_4$

20. Which is not soluble in water?

- (a)  $\text{PbSO}_4$  (b)  $\text{CdSO}_4$   
(c)  $\text{Bi}(\text{SO}_4)_2$  (d)  $\text{CuSO}_4$

21. Which of the following is colour red?

- (a)  $\text{Cu}_2\text{O}$  (b)  $\text{CuF}$   
(c)  $\text{ZnF}_2$  (d)  $\text{ZnCl}_2$

22. How many unpaired electrons are present in  $[\text{Cr}(\text{NH}_3)_5\text{Br}]^{3+}$  ?  
 (a) 1 (b) 2  
 (c) 3 (d) 4
23.  $\text{S} + \frac{3}{2}\text{O}_2 \longrightarrow \text{SO}_3 \Delta H = 2x$ ,  
 $\text{SO}_2 + \frac{1}{2}\text{O}_2 \longrightarrow \text{SO}_3 \Delta H = y$ ,  
 then heat of formation of  $\text{SO}_2$  is :  
 (a)  $2x - y$  (b)  $2x + y$   
 (c)  $x + y$  (d)  $\frac{2x - y}{2}$
24. Reagent (catalyst) used in Friedel-Craft's alkylation reaction is :  
 (a)  $\text{AlCl}_3$  (b) anhyd.  $\text{AlCl}_3$   
 (c)  $\text{N}_2$  (d) He
25. Catalyst used in making  $\text{H}_2\text{SO}_4$  in contact process is :  
 (a)  $\text{V}_2\text{O}_5$  (b)  $\text{Fe}_2\text{O}_3$   
 (c)  $\text{Cr}_2\text{O}_3$  (d)  $\text{CrO}_3$
26. When acetamide is reacted with  $\text{NaOBr}$ , then product formed is :  
 (a) ethanamine (b) methanamine  
 (c) methanamide (d) ethanenitrile
27. Isocyanide is prepared by :  
 (a) Friedel Craft's reaction  
 (b) Wurtz's reaction  
 (c) Williamson synthesis  
 (d) Carbylamine reaction
28. If rate of diffusion of  $\text{CH}_4$  is twice than that of a gas x, then its molecular mass is :  
 (a) 64 g (b) 16 g  
 (c) 32 g (d) 8 g
29. Which one is not Lewis acid?  
 (a)  $\text{BeF}_2$  (b)  $\text{SnCl}_4$   
 (c)  $\text{AlCl}_3$  (d)  $\text{BF}_3$
30. Natural gas mainly consists of:  
 (a) methane (b) butane  
 (c) propane (d) ethane + octane
31. Phenol is treated with Zn to form:  
 (a) benzoic acid (b) benzyl alcohol  
 (c) benzene (d) benzoquinone
32. Which one is isoelectronic with  $\text{CO}$ ?  
 (a)  $\text{N}_2^-$  (b)  $\text{N}_2^+$   
 (c)  $\text{CN}^-$  (d)  $\text{NO}$
33. How much volume of 1M  $\text{H}_2\text{SO}_4$  is required to neutralize 20 mL of 1M  $\text{NaOH}$ ?  
 (a) 10 mL (b) 20 mL  
 (c) 5 mL (d) 15 mL
34. Fenton's reagent is:  
 (a)  $\text{SnCl}_2 + \text{HCl}$  (b)  $\text{AgNO}_3 + \text{NH}_4\text{OH}$   
 (c)  $\text{CuSO}_4 + \text{NaOH}$  (d)  $\text{FeSO}_4 + \text{H}_2\text{O}_2$
35.  $\text{Cu}^{2+} + \text{Ag} \longrightarrow \text{Cu} + \text{Ag}^+$  oxidation half reaction is :  
 (a)  $\text{Cu}^{2+} \rightarrow \text{Cu}$  (b)  $\text{Ag} \rightarrow \text{Ag}^+$   
 (c)  $\text{Cu} \rightarrow \text{Cu}^{2+}$  (d) all of these
36.  $\text{C}_4\text{H}_{10}\text{O}$  has how many isomeric alcohols?  
 (a) 1 (b) 2  
 (c) 3 (d) 4
37. IUPAC name of  $\text{CH}_3 - \text{C} \equiv \text{C} - \text{HC} - (\text{CH}_3)_2$  is :  
 (a) 4-methyl-2-pentyne  
 (b) 1,1-dimethyl-2-butyne  
 (c) 2-methyl-4-pentyne  
 (d) 4,4-dimethyl-2-butyne
38. Fehling test is given by:  
 (a) glucose (b) fructose  
 (c) sucrose (d) all of these
39. Number of isomeric primary amine of molecular formula  $\text{C}_4\text{H}_{11}\text{N}$  is :  
 (a) 1 (b) 2  
 (c) 3 (d) 4
40. For reaction  $3\text{X} + \text{Y} \rightleftharpoons \text{X}_3\text{Y}$   $\Delta H = +ve$ , amount of  $\text{X}_3\text{Y}$  can be changed by:  
 (a) changing temperature  
 (b) changing pressure  
 (c) changing temperature, pressure,  
 (d) changing temperature, pressure, adding catalyst
41. Which one is most ionic?  
 (a)  $\text{P}_2\text{O}_5$  (b)  $\text{MnO}_2$   
 (c)  $\text{Mn}_2\text{O}_7$  (d)  $\text{P}_2\text{O}_3$
42. Which one gives  $\text{I}_2$  on reaction with  $\text{KI}$ ?  
 (a)  $\text{Ag}_2\text{SO}_4$  (b)  $\text{CuSO}_4$   
 (c)  $\text{PbSO}_4$  (d)  $\text{CdSO}_4$
43. Colour of the solution when  $\text{KI}$  reacts with  $\text{Br}_2$  is :  
 (a) blue (b) black  
 (c) red (d) no change
44. Finely divided iron combines with  $\text{CO}$  to give:  
 (a)  $\text{Fe}(\text{CO})_5$  (b)  $\text{Fe}_2(\text{CO})_9$   
 (c)  $\text{Fe}_3(\text{CO})_{12}$  (d)  $\text{Fe}(\text{CO})_6$
45. Which of the following is pyramidal?  
 (a)  $\text{PCl}_3$  (b)  $\text{CO}_3^{2-}$   
 (c)  $\text{SO}_2$  (d)  $\text{NO}_3^-$
46. Which of the following conducts electricity?  
 (a) Crystal  $\text{NaCl}$  (b) Diamond  
 (c) Molten  $\text{KBr}$  (d) Sulphur



47. Ratio of kinetic energy of hydrogen and helium gas at 300 K is :  
 (a) 2 : 1 (b) 4 : 5  
 (c) 1 : 1 (d) 1 : 2
48. Which of the following has highest energy?  
 (a)  $n = 2, l = 1$   
 (b)  $n = 3, l = 2$   
 (c)  $n = 3, l = 1$   
 (d)  $n = 2, l = 0$
49. Oxidation state of phosphorus in pyrophosphoric acid is :  
 (a) +5 (b) +3 (c) +4 (d) +1
50. If the 75% of a first order reaction is complete in 8 min, then time taken to decompose 50% of its initial amount is :  
 (a) 2 min (b) 4 min  
 (c) 12 min (d) 1 min

## Mathematics

1. If  $\alpha \neq \beta$  and  $\alpha^2 = 5\alpha - 3, \beta^2 = 5\beta - 3$ , then the equation having  $\alpha/\beta$  and  $\beta/\alpha$  as its roots is :  
 (a)  $3x^2 + 19x + 3 = 0$   
 (b)  $3x^2 - 19x + 3 = 0$   
 (c)  $3x^2 - 19x - 3 = 0$   
 (d)  $x^2 - 16x + 1 = 0$
2. If  $y = (x + \sqrt{1 + x^2})^n$ , then  $(1 + x^2) \frac{d^2y}{dx^2} + x \frac{dy}{dx}$  is :  
 (a)  $n^2y$  (b)  $-n^2y$   
 (c)  $-y$  (d)  $2x^2y$
3. If  $1, \log_3 \sqrt{(3^{1-x} + 2)}, \log_3 (4 \cdot 3^x - 1)$  are in AP, then  $x$  equals :  
 (a)  $\log_3 4$  (b)  $1 - \log_3 4$   
 (c)  $1 - \log_4 3$  (d)  $\log_4 3$
4. A problem in mathematics is given to three students A, B, C and their respective probability of solving the problem is  $\frac{1}{2}, \frac{1}{3}$  and  $\frac{1}{4}$ . Probability that the problem is solved, is :  
 (a)  $\frac{3}{4}$  (b)  $\frac{1}{2}$  (c)  $\frac{2}{3}$  (d)  $\frac{1}{3}$
5. The angle of elevation of a tower at a point distant  $d$  metres from its base is  $30^\circ$ . If the tower is 20 m high, then the value of  $d$  is :  
 (a)  $10\sqrt{3}$  m (b)  $\frac{20}{\sqrt{3}}$  m  
 (c)  $20\sqrt{3}$  m (d) 10 m
6.  $l, m, n$  are the  $p$ th,  $q$ th and  $r$ th terms of a GP and all positive, then  $\begin{vmatrix} \log l & p & 1 \\ \log m & q & 1 \\ \log n & r & 1 \end{vmatrix}$  equals :  
 (a) 3 (b) 2  
 (c) 1 (d) zero
7.  $\lim_{x \rightarrow 0} \frac{\sqrt{1 - \cos 2x}}{\sqrt{2x}}$  is equal to :  
 (a)  $\lambda$  (b)  $-1$   
 (c) zero (d) does not exist
8. A triangle with vertices (4, 0), (-1, -1), (3, 5) is :  
 (a) isosceles and right angled  
 (b) isosceles but not right angled  
 (c) right angled but not isosceles  
 (d) neither right angled nor isosceles
9.  $\cot^{-1}(\sqrt{\cos \alpha}) - \tan^{-1}(\sqrt{\cos \alpha}) = x$ , then  $\sin x$  is equal to :  
 (a)  $\tan^2\left(\frac{\alpha}{2}\right)$  (b)  $\cot^2\left(\frac{\alpha}{2}\right)$   
 (c)  $\tan \alpha$  (d)  $\cot\left(\frac{\alpha}{2}\right)$
10. A plane which passes through the point (3, 2, 0) and the line  $\frac{x-3}{1} = \frac{y-6}{5} = \frac{z-4}{4}$  is :  
 (a)  $x - y + z = 1$  (b)  $x + y + z = 5$   
 (c)  $x + 2y - z = 1$  (d)  $2x - y + z = 5$
11. The solution of the equation  $\frac{d^2y}{dx^2} = e^{-2x}$  is :  
 (a)  $\frac{e^{-2x}}{4}$  (b)  $\frac{e^{-2x}}{4} + cx + d$   
 (c)  $\frac{1}{4}e^{-2x} + cx^2 + d$  (d)  $\frac{1}{4}e^{-2x} + c + d$
12.  $\lim_{x \rightarrow \infty} \left( \frac{x^2 + 5x + 3}{x^2 + x + 2} \right)^x$  is equal to :  
 (a)  $e^4$  (b)  $e^2$  (c)  $e^3$  (d)  $e$
13. The domain of  $\sin^{-1} \left[ \log_3 \left( \frac{x}{3} \right) \right]$  is :  
 (a) [1, 9] (b) [-1, 9]  
 (c) [-9, 1] (d) [-9, -1]
14. The value of  $2^{1/4} \cdot 4^{1/8} \cdot 8^{1/16} \dots \infty$  is :  
 (a) 1 (b) 2 (c)  $3/2$  (d) 4

15. Fifth term of a GP is 2, then the product of its 9 terms is :  
 (a) 256 (b) 512  
 (c) 1024 (d) none of these
16.  $\int_0^{10\pi} |\sin x| dx$  is :  
 (a) 20 (b) 8  
 (c) 10 (d) 18
17.  $I_n = \int_0^{\pi/4} \tan^n x dx$ , then  $\lim_{n \rightarrow \infty} n[I_n + I_{n+2}]$  is equal to :  
 (a)  $\frac{1}{2}$  (b) 1  
 (c)  $\infty$  (d) zero
18.  $\int_{-\pi}^{\pi} \frac{2x(1 + \sin x)}{1 + \cos^2 x} dx$  is :  
 (a)  $\frac{\pi^2}{4}$  (b)  $\pi^2$   
 (c) zero (d)  $\frac{\pi}{2}$
19. The period of the function  $f(x) = \sin^4 x + \cos^4 x$  is :  
 (a)  $\pi$  (b)  $\frac{\pi}{2}$   
 (c)  $2\pi$  (d) none of these
20. If  $x^y = e^{x-y}$ , then  $\frac{dy}{dx}$  is :  
 (a)  $\frac{1+x}{1+\log x}$  (b)  $\frac{1-\log x}{1+\log x}$   
 (c) not defined (d)  $\frac{\log x}{(1+\log x)^2}$
21. The two curves  $x^3 - 3xy^2 + 2 = 0$  and  $3x^2y - y^3 - 2 = 0$  :  
 (a) cut at right angles  
 (b) touch each other  
 (c) cut at an angle  $\frac{\pi}{3}$   
 (d) cut at an angle  $\frac{\pi}{4}$
22. The function  $f(x) = \cot^{-1} x + x$  increases in the interval :  
 (a)  $(1, \infty)$   
 (b)  $(-1, \infty)$   
 (c)  $(-\infty, \infty)$   
 (d)  $(0, \infty)$
23. The greatest value of  $f(x) = (x+1)^{1/3} - (x-1)^{1/3}$  on  $[0, 1]$  is :  
 (a) 1 (b) 2  
 (c) 3 (d)  $1/3$
24.  $\int \frac{dx}{x(x^n+1)}$  is equal to :  
 (a)  $\frac{1}{n} \log \left( \frac{x^n}{x^n+1} \right) + c$   
 (b)  $\frac{1}{n} \log \left( \frac{x^n+1}{x^n} \right) + c$   
 (c)  $\log \left( \frac{x^n}{x^n+1} \right) + c$   
 (d) none of the above
25. The area bounded by the curve  $y = 2x - x^2$  and the straight line  $y = -x$  is given by :  
 (a)  $\frac{9}{2}$  sq unit  
 (b)  $\frac{43}{6}$  sq unit  
 (c)  $\frac{35}{6}$  sq unit  
 (d) none of these
26. The differential equation of all non-vertical lines in a plane is :  
 (a)  $\frac{d^2y}{dx^2} = 0$  (b)  $\frac{d^2x}{dy^2} = 0$   
 (c)  $\frac{dy}{dx} = 0$  (d)  $\frac{dx}{dy} = 0$
27. Given two vectors  $\hat{i} - \hat{j}$  and  $\hat{i} + 2\hat{j}$  the unit vector coplanar with the two vectors and perpendicular to first is :  
 (a)  $\pm \frac{1}{\sqrt{2}} (\hat{i} + \hat{j})$  (b)  $\frac{1}{\sqrt{5}} (2\hat{i} + \hat{j})$   
 (c)  $\pm \frac{1}{\sqrt{2}} (\hat{i} + \hat{k})$  (d) none of these
28. The vector  $\hat{i} + x\hat{j} + 3\hat{k}$  is rotated through an angle  $\theta$  and doubled in magnitude, then it becomes  $4\hat{i} + (4x-2)\hat{j} + 2\hat{k}$ . The value of  $x$  is :  
 (a)  $\left\{-\frac{2}{3}, 2\right\}$  (b)  $\left\{\frac{1}{3}, 2\right\}$   
 (c)  $\left\{\frac{2}{3}, 0\right\}$  (d)  $\{2, 7\}$
29. A parallelopiped is formed by planes drawn through the points  $(2, 3, 5)$  and  $(5, 9, 7)$  parallel to the coordinate planes. The length of a diagonal of the parallelopiped to piped is :  
 (a) 7 (b)  $\sqrt{38}$   
 (c)  $\sqrt{155}$  (d) none of these



30. The equation of the plane containing the line  $\frac{x-x_1}{l} = \frac{y-y_1}{m} = \frac{z-z_1}{n}$  is  $a(x-x_1) + b(y-y_1) + c(z-z_1) = 0$ , where :  
 (a)  $ax_1 + by_1 + cz_1 = 0$   
 (b)  $al + bm + cn = 0$   
 (c)  $\frac{a}{l} = \frac{b}{m} = \frac{c}{n}$   
 (d)  $lx_1 + my_1 + nz_1 = 0$
31. A and B play a game where each is asked to select a number from 1 to 25. If the two numbers match, both of them win a prize. The probability that they will not win a prize in a single trial is :  
 (a)  $\frac{1}{25}$  (b)  $\frac{24}{25}$   
 (c)  $\frac{2}{25}$  (d) none of these
32. The equation of the directrix of the parabola  $y^2 + 4y + 4x + 2 = 0$  is :  
 (a)  $x = -1$  (b)  $x = 1$   
 (c)  $x = -\frac{3}{2}$  (d)  $x = \frac{3}{2}$
33. Let  $T_n$  denote the number of triangles which can be formed using the vertices of a regular polygon of  $n$  sides. If  $T_{n+1} - T_n = 21$ , then  $n$  equals :  
 (a) 5 (b) 7 (c) 6 (d) 4
34. In a triangle  $ABC$ ,  $2ca \sin \frac{A-B+C}{2}$  is equal to :  
 (a)  $a^2 + b^2 - c^2$  (b)  $c^2 + a^2 - b^2$   
 (c)  $b^2 - c^2 - a^2$  (d)  $c^2 - a^2 - b^2$
35. For  $x \in \mathbb{R}$ ,  $\lim_{x \rightarrow \infty} \left( \frac{x-3}{x+2} \right)^x$  is equal to :  
 (a)  $e$  (b)  $e^{-1}$  (c)  $e^{-5}$  (d)  $e^5$
36. The incentre of the triangle with vertices  $(1, \sqrt{3})$ ,  $(0, 0)$  and  $(2, 0)$  is :  
 (a)  $\left(1, \frac{\sqrt{3}}{2}\right)$  (b)  $\left(\frac{2}{3}, \frac{1}{\sqrt{3}}\right)$   
 (c)  $\left(\frac{2}{3}, \frac{\sqrt{3}}{2}\right)$  (d)  $\left(1, \frac{1}{\sqrt{3}}\right)$
37. If the vectors  $\vec{a}$ ,  $\vec{b}$  and  $\vec{c}$  from the sides  $BC$ ,  $CA$  and  $AB$  respectively, of a triangle  $ABC$ , then :  
 (a)  $\vec{a} \cdot \vec{b} + \vec{b} \cdot \vec{c} + \vec{c} \cdot \vec{a} = 0$   
 (b)  $\vec{a} \times \vec{b} = \vec{b} \times \vec{c} = \vec{c} \times \vec{a} = \vec{0}$   
 (c)  $\vec{a} \cdot \vec{b} = \vec{b} \cdot \vec{c} = \vec{c} \cdot \vec{a} = 0$   
 (d)  $\vec{a} \times \vec{a} + \vec{a} \times \vec{c} + \vec{c} \times \vec{a} = \vec{0}$
38. If  $\omega$  is an imaginary cube root of unity, then  $(1 + \omega - \omega^2)^7$  equals :  
 (a)  $128 \omega$  (b)  $-128 \omega$   
 (c)  $128 \omega^2$  (d)  $-128 \omega^2$
39.  $\sin^2 \theta = \frac{4xy}{(x+y)^2}$  is true, if and only if :  
 (a)  $x + y \neq 0$  (b)  $x = y, x \neq 0, y \neq 0$   
 (c)  $x = y$  (d)  $x \neq 0, y \neq 0$
40. The radius of the circle passing through the foci of the ellipse  $\frac{x^2}{16} + \frac{y^2}{9} = 1$  and having its centre at  $(0, 3)$  is :  
 (a) 4 (b) 3 (c)  $\sqrt{12}$  (d)  $\frac{7}{2}$
41. If  $(\omega \neq 1)$  is a cubic root of unity, then  $\begin{vmatrix} 1 & 1+i+\omega^2 & \omega^2 \\ 1-i & -1 & \omega^2-1 \\ -i & -1+\omega-i & -1 \end{vmatrix}$  equals :  
 (a) zero (b) 1  
 (c)  $i$  (d)  $\omega$
42. Let  $f(2) = 4$  and  $f'(2) = 4$ . Then  $\lim_{x \rightarrow 2} \frac{x f(2) - 2f(x)}{x-2}$  is given by :  
 (a) 2 (b)  $-2$   
 (c)  $-4$  (d) 3
43. Three straight lines  $2x + 11y - 5 = 0$ ,  $24x + 7y - 20 = 0$  and  $4x - 3y - 2 = 0$  :  
 (a) form a triangle  
 (b) are only concurrent  
 (c) are concurrent with on line bisecting the angle between the other two  
 (d) none of the above
44. A straight line through the point  $(2, 2)$  intersects the lines  $\sqrt{3}x + y = 0$  and  $\sqrt{3}x - y = 0$  at the points  $A$  and  $B$ . The equation to the line  $AB$ , so that triangle  $OAB$  is equilateral is :  
 (a)  $x - 2 = 0$  (b)  $y - 2 = 0$   
 (c)  $x + y - 4 = 0$  (d) none of these
45. The greatest distance of the point  $P(10, 7)$  from the circle  $x^2 + y^2 - 4x - 2y - 20 = 0$  is :  
 (a) 10 (b) 15  
 (c) 5 (d) none of these



46. The equation of the ellipse whose foci are  $(\pm 2, 0)$  and eccentricity  $\frac{1}{2}$  is :

- (a)  $\frac{x^2}{12} + \frac{y^2}{16} = 1$  (b)  $\frac{x^2}{16} + \frac{y^2}{12} = 1$   
 (c)  $\frac{x^2}{16} + \frac{y^2}{8} = 1$  (d) none of these

47. The equation of the chord joining two points  $(x_1, y_1)$  and  $(x_2, y_2)$  on the rectangular hyperbola  $xy = c^2$  is :

- (a)  $\frac{x}{x_1 + x_2} + \frac{y}{y_1 + y_2} = 1$   
 (b)  $\frac{x}{x_1 - x_2} + \frac{y}{y_1 - y_2} = 1$   
 (c)  $\frac{x}{y_1 + y_2} + \frac{y}{x_1 + x_2} = 1$   
 (d)  $\frac{x}{y_1 - y_2} + \frac{y}{x_1 - x_2} = 1$

48. Let  $R$  be the resultant of  $P$  and  $Q$  and if  $\frac{P}{3} = \frac{Q}{7} = \frac{R}{5}$ , the angle between  $P$  and  $R$  is :

- (a)  $\cos^{-1}\left(\frac{11}{14}\right)$  (b)  $\cos^{-1}\left(-\frac{11}{14}\right)$   
 (c)  $\frac{2\pi}{3}$  (d)  $\frac{5\pi}{6}$

49. Two bodies of different masses  $m_1$  and  $m_2$  are dropped from different heights  $h_1$  and  $h_2$ . The ratio of the time taken by the two bodies to fall through these distances is :

- (a)  $h_1 : h_2$  (b)  $\sqrt{h_1} : \sqrt{h_2}$   
 (c)  $h_1^2 : h_2^2$  (d)  $h_2 : h_1$

50. If  $\text{var}(x) = 8.25$ ,  $\text{var}(y) = 33.96$  and  $\text{cov}(x, y) = 10.2$ , then the correlation coefficient is :

- (a) 0.89 (b) -0.98  
 (c) 0.61 (d) -0.16

## ANSWERS

### PHYSICS

- |         |         |         |         |         |         |         |         |         |         |
|---------|---------|---------|---------|---------|---------|---------|---------|---------|---------|
| 1. (d)  | 2. (b)  | 3. (a)  | 4. (b)  | 5. (d)  | 6. (b)  | 7. (d)  | 8. (b)  | 9. (b)  | 10. (a) |
| 11. (b) | 12. (a) | 13. (d) | 14. (d) | 15. (d) | 16. (a) | 17. (a) | 18. (c) | 19. (d) | 20. (b) |
| 21. (a) | 22. (c) | 23. (a) | 24. (c) | 25. (b) | 26. (c) | 27. (c) | 28. (a) | 29. (d) | 30. (d) |
| 31. (b) | 32. (c) | 33. (d) | 34. (b) | 35. (a) | 36. (c) | 37. (d) | 38. (b) | 39. (b) | 40. (a) |
| 41. (a) | 42. (b) | 43. (c) | 44. (b) | 45. (b) | 46. (d) | 47. (c) | 48. (c) | 49. (d) | 50. (a) |

### CHEMISTRY

- |         |         |         |         |         |         |         |         |         |         |
|---------|---------|---------|---------|---------|---------|---------|---------|---------|---------|
| 1. (d)  | 2. (c)  | 3. (c)  | 4. (a)  | 5. (b)  | 6. (d)  | 7. (d)  | 8. (a)  | 9. (a)  | 10. (b) |
| 11. (c) | 12. (d) | 13. (b) | 14. (b) | 15. (d) | 16. (b) | 17. (c) | 18. (c) | 19. (a) | 20. (a) |
| 21. (a) | 22. (c) | 23. (a) | 24. (b) | 25. (a) | 26. (b) | 27. (d) | 28. (a) | 29. (b) | 30. (a) |
| 31. (c) | 32. (c) | 33. (a) | 34. (d) | 35. (b) | 36. (d) | 37. (a) | 38. (a) | 39. (d) | 40. (c) |
| 41. (d) | 42. (b) | 43. (b) | 44. (a) | 45. (a) | 46. (c) | 47. (c) | 48. (b) | 49. (a) | 50. (b) |

### MATHEMATICS

- |         |         |         |         |         |         |         |         |         |         |
|---------|---------|---------|---------|---------|---------|---------|---------|---------|---------|
| 1. (b)  | 2. (a)  | 3. (b)  | 4. (a)  | 5. (c)  | 6. (d)  | 7. (d)  | 8. (a)  | 9. (a)  | 10. (a) |
| 11. (b) | 12. (a) | 13. (a) | 14. (b) | 15. (b) | 16. (a) | 17. (b) | 18. (b) | 19. (b) | 20. (d) |
| 21. (a) | 22. (c) | 23. (b) | 24. (a) | 25. (a) | 26. (a) | 27. (a) | 28. (a) | 29. (a) | 30. (b) |
| 31. (b) | 32. (d) | 33. (b) | 34. (b) | 35. (c) | 36. (d) | 37. (b) | 38. (d) | 39. (b) | 40. (a) |
| 41. (a) | 42. (c) | 43. (c) | 44. (b) | 45. (b) | 46. (b) | 47. (a) | 48. (b) | 49. (b) | 50. (c) |

## HINTS & SOLUTIONS

### Physics

1. **Key Idea :** Only transverse waves can be polarised.

Light waves and X-rays are transverse, hence the vibrations of the electric vector must occur in a plane perpendicular to the direction of propagation of light and hence, give rise to polarised light.

**Note :** Longitudinal waves cannot be polarised.

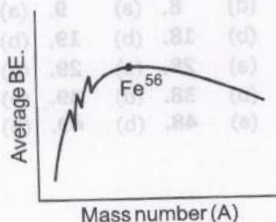
2. According to quantum theory light travels in small bundles or packets of energy known as photons. The energy of photon in such case is  $h\nu$ , where  $h$  is Planck's constant and  $\nu$  the frequency of light. The intensity of light depends on the number of these photons. Einstein explained the phenomenon of photoelectric effect on the basis of Planck's quantum theory.

**Note :** Interference, diffraction and polarisation proved the wave nature of light.

3. **Key Idea :** Binding energy per nucleon measures the stability of nucleus.

In order to compare the stability of the nuclei of different atoms, binding energy per nucleon is determined. Higher the binding energy per nucleon more stable is the nucleus. The curve has almost a flat maximum roughly from  $A = 50$  to  $A = 80$  corresponding to an average BE per nucleon of about 8.5 MeV.

The nuclei having mass number between 50 and 80 are most stable. Iron ( $\text{Fe}^{56}$ ) ( $A = 56$ ) having BE per nucleon of about 8.8 MeV is most stable.

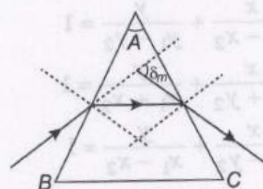


4. Light shows dual character at times it behaves as a wave and at times as particle. While interference, diffraction show the wave nature of light, particle nature is supported by photoelectric effect.

5. **Key Idea :**  $\sin(90^\circ - \theta) = \cos \theta$

The refractive index ( $\mu$ ) of a prism of angle  $A$ , and minimum deviation  $\delta_m$  is given by

$$\mu = \frac{\sin\left(\frac{A + \delta_m}{2}\right)}{\sin A/2}$$



Given,  $\mu = \cot \frac{A}{2}$

$$\therefore \cot \frac{A}{2} = \frac{\sin\left(\frac{A + \delta_m}{2}\right)}{\sin(A/2)}$$

$$\Rightarrow \frac{\cos A/2}{\sin A/2} = \frac{\sin\left(\frac{A + \delta_m}{2}\right)}{\sin(A/2)}$$

$$\Rightarrow \cos \frac{A}{2} = \sin\left(\frac{A + \delta_m}{2}\right)$$

$$\therefore \sin\left(90^\circ - \frac{A}{2}\right) = \sin\left(\frac{A + \delta_m}{2}\right)$$

$$\Rightarrow 90^\circ - \frac{A}{2} = \frac{A + \delta_m}{2}$$

$$\Rightarrow 180^\circ - A = A + \delta_m$$

$$\Rightarrow \delta_m = 180^\circ - 2A = \pi - 2A$$

6. **Key Idea :** The system executes SHM.

The frequency of oscillation of spring mass system is given by

$$f = \frac{1}{2\pi} \sqrt{\frac{k}{m}}$$

For the given arrangement

$$k' = k_1 + k_2 = k + k = 2k$$

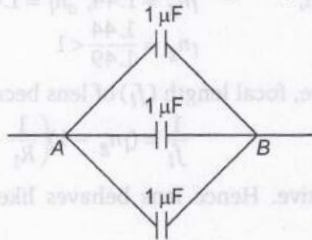
Hence, frequency of oscillation is

$$f' = \frac{1}{2\pi} \sqrt{\frac{k'}{m}}$$

$$\Rightarrow f' = \frac{1}{2\pi} \sqrt{\frac{2k}{m}} = \sqrt{2}f$$



7. The given circuit can be rearranged as follows, it consists of three capacitors connected in parallel, hence equivalent resistance of the combination is  $C' = C_1 + C_2 + C_3$



Given,  $C' = 1 + 1 + 1 = 3\mu F$

8. When a conductor of capacitance  $C$  is charged, it acquires a potential

$$V = \frac{q}{C}$$

When connected in parallel, equivalent capacitance of combination is

$$C' = C_1 + C_2$$

Also, energy flow  $\left(E = \frac{1}{2} C' V^2\right)$  will not take place, if potential across capacitors is same.

9. The electric field ( $E$ ) intensity is given by

$$E = -\frac{dV}{dx}$$

Its unit is V/m.

Also,  $F = qE$

$$\Rightarrow E = \frac{F}{q}$$

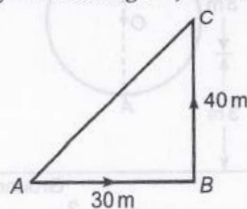
Hence, unit of  $E$  is  $\frac{N}{C}$

Also  $E = \frac{W}{qx}$

Hence unit is  $\frac{J}{C \cdot m}$

10. **Key Idea :** Average velocity =  $\frac{\text{Displacement}}{\text{Time taken}}$

Let body move along AB, then BC.



Distance travelled in 2s.

$$S_1 = 15 \times 2 = 30 \text{ m}$$

Distance travelled in 8s

$$S_2 = 5 \times 8 = 40 \text{ m}$$

$$\begin{aligned} \therefore \text{Displacement } AC &= \sqrt{(30)^2 + (40)^2} \\ &= \sqrt{900 + 1600} = \sqrt{2500} \\ &= 50 \text{ m} \end{aligned}$$

$$\begin{aligned} \text{Hence average velocity} &= \frac{\text{displacement}}{\text{time taken}} \\ &= \frac{50}{8+2} = 5 \text{ m/s} \end{aligned}$$

11. **Key Idea :** Monoatomic gas has three translational degrees of freedom.

Specific heat at constant volume ( $C_V$ ) =  $\frac{f}{2} R$   
and specific heat at constant pressure ( $C_P$ )

$$C_P = \left(\frac{f}{2} + 1\right) R$$

For monoatomic gas  $f = 3$

$$\therefore C_V = \frac{3}{2} R, C_P = \left(\frac{3}{2} + 1\right) R = \frac{5}{2} R$$

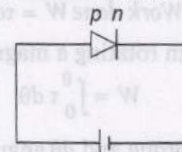
From Mayor's formula

$$C_P - C_V = R$$

$$\therefore \frac{5}{2} R - \frac{3}{2} R = R$$

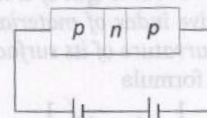
$$\text{and } \gamma = \frac{C_P}{C_V} = \frac{5/2}{3/2} = \frac{5}{3}$$

12. When for a  $p$ - $n$  junction diode when  $p$ -type is connected to positive of battery and  $n$ -type to negative then the diode is forward biased, and a constant current flows.



Hence, correct variation is shown by graph (a).

13. When a transistor is properly biased emitter base junction is forward biased so the size of depletion layer becomes small due to which majority carriers move from emitter to base. The collector-base junction is reverse biased, due to which the collector-base depletion layer becomes large.



- 14. Key Idea :** When length is increased volume remains same and specific resistance constant. The resistance of a wire of length  $l$  is given by

$$R = \rho \frac{l}{A} \quad \dots(i)$$

When wire is stretched, volume remains constant, hence,

$$V = Al = \text{constant} \quad \dots(ii)$$

Multiply and divide Eq. (i) by  $l$ , we get

$$R = \frac{\rho l^2}{Al}$$

$$\frac{R_2}{R_1} = \left(\frac{l_2}{l_1}\right)^2$$

$$\text{Given, } l_2 = 3l_1$$

$$\frac{R_2}{20} = \left(\frac{3l_1}{l_1}\right)^2 = 9$$

$$\Rightarrow R_2 = 20 \times 9 = 180 \Omega$$

- 15. From Gauss's theorem**

$$\oint E \cdot ds = \phi$$

where  $E$  is electric field intensity,  $s$  the surface area,  $\phi$  the flux.

$$\text{Given, } s = \pi r^2 = \pi \left(\frac{d}{2}\right)^2 = \frac{\pi d^2}{4}$$

where  $r$  is radius and  $d$  the diameter.

$$\phi = E \times \frac{\pi d^2}{4}$$

$$\Rightarrow E = \frac{4\phi}{\pi d^2}$$

- 16. Key Idea :** Work done  $W = \tau d\theta$

Work done in rotating a magnet is given by

$$W = \int_0^\theta \tau d\theta$$

where  $\tau$  is torque and  $d\theta$  angular charge

$$\text{Also, } \tau = MH \sin \theta$$

$$W = \int_0^\theta MH \sin \theta d\theta$$

$$\Rightarrow W = MH \int_0^\theta \sin \theta d\theta$$

$$\Rightarrow W = MH[1 - \cos \theta]_0^\theta$$

$$\Rightarrow W = MH[1 - \cos \theta + \cos 0]$$

$$\Rightarrow W = MH[1 - \cos \theta]$$

- 17. Key Idea :** Focal length of a lens depends upon the refractive index of material of lens and the radius of curvature of its surfaces.

From lens formula

$$\frac{1}{f} = (n - 1) \left( \frac{1}{R_1} - \frac{1}{R_2} \right)$$

When lens is dipped in liquid, its focal length is

$$l n_g = \frac{a n_l}{a n_l}$$

Given,

$$l n_g = 1.44, a n_l = 1.49$$

$\therefore$

$$l n_g = \frac{1.44}{1.49} < 1$$

Hence, focal length ( $f_l$ ) of lens becomes

$$\frac{1}{f_l} = (l n_g - 1) \left( \frac{1}{R_1} - \frac{1}{R_2} \right)$$

negative. Hence lens behaves like a concave lens.

**Note :** If focal length was positive it will behave like a convex lens, when infinite it behaves like a plane transparent plate.

- 18. From Doppler's effect, the perceived frequency when source approaches observer is**

$$n' = n \left( \frac{v}{v - v_s} \right) \quad \dots(i)$$

When source recedes the observer

$$n'' = n \left( \frac{v}{v + v_s} \right) \quad \dots(ii)$$

From Eqs. (i) and (ii), we get

$$\frac{n'}{n''} = \frac{v + v_s}{v - v_s}$$

$$\frac{6}{5} = \frac{330 + v_s}{330 - v_s}$$

$$\Rightarrow 1980 - 6v_s = 1650 + 5v_s$$

$$\Rightarrow 11v_s = 1980 - 1650 = 330$$

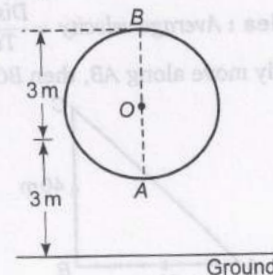
$$v_s = \frac{330}{11} = 30 \text{ m/s}$$

- 19. Key Idea :** Maximum velocity at lowest point is  $\sqrt{5}$  times maximum velocity at highest point.

Diameter AB of circle is 3m

Therefore, radius  $r = \frac{3}{2} \text{ m}$

$$\text{velocity at A}(v) = \sqrt{5gr}$$



$$\text{Given, } g = 10 \text{ m/s}^2, r = \frac{3}{2} \text{ m}$$

$$\therefore v = \sqrt{5 \times 10 \times \frac{3}{2}} = 5\sqrt{3} \text{ m/s}$$



20. The acceleration due to gravity  $g$  is given by

$$g = \frac{GM}{R^2} \quad \dots(i)$$

where  $G$  is gravitational constant,  $M$  the mass,  $R$  the radius of earth.

Taking log on both sides, we get

$$\log g = \log G + \log M - 2 \log R$$

Differentiating, we get

$$\frac{\Delta g}{g} = 0 + 0 - \frac{2\Delta R}{R}$$

$$\text{Given, } \frac{\Delta R}{R} = -1\%$$

$$\therefore \frac{\Delta g}{g} = -2(-1\%) = 2\% \text{ increase}$$

21. From Wein's displacement law

$$\lambda_m T = \text{constant} \quad \dots(i)$$

where  $\lambda_m$  corresponds to maximum wavelength,  $T$  is absolute temperature.

$$\text{Also, } v = n\lambda \quad \dots(ii)$$

where  $v$  is velocity,  $n$  the frequency,  $\lambda$  the wavelength.

From Eqs. (i) and (ii), we get

$$\frac{v}{n} \cdot T = \text{constant}$$

Given  $T_1 < T_2$ , hence frequency corresponding to maximum intensity is less at  $T_1$ .

22. Work done on a gas is given by

$$W = P\Delta V$$

where  $P$  is pressure,  $\Delta V$  the change in volume.

Given  $\Delta V = 0$ , therefore

$$W = P \times 0 = 0$$

23. From Newton's law of cooling

$$\frac{dH}{dt} = K \left( \frac{\theta_1 + \theta_2}{2} - \theta_0 \right)$$

where  $\theta_0$  is temperature of surrounding,  $\frac{\theta_1 + \theta_2}{2}$

the temperature of body.

Hence,  $t_3 > t_2 > t_1$ .

24. **Key Idea :** Since pressure is constant, gas obeys Charles's law.

From Charles's law

$$\frac{V}{T} = \text{constant}$$

where  $V$  is volume,  $T$  the temperature.

$$\therefore \frac{V_2}{V_1} = \frac{T_2}{T_1}$$

$$\text{Given } T_1 = 27^\circ\text{C} = 300\text{ K}, V_2 = 2V_1$$

$$\therefore 2 \times 300 = T_2$$

$$\Rightarrow T_2 = 600\text{ K}$$

$$\text{In centigrade } T_2 = 600 - 273 = 327^\circ\text{C}.$$

25. **Key Idea :** Ideal gas equation is

$$PV = RT$$

Work done in isothermal process is

$$W = \int_{V_1}^{V_2} P dV$$

From ideal gas law  $PV = RT$

$$\therefore W = \int_{V_1}^{V_2} \left( \frac{RT}{V} \right) dV = RT \int_{V_1}^{V_2} \frac{dV}{V}$$

$$\Rightarrow W = RT \ln \left( \frac{V_2}{V_1} \right)$$

26. **Key Idea :** In isochoric process, volume is constant.

$$\text{Work done} = P\Delta V,$$

where  $P$  is pressure,  $\Delta V$  the change in volume.

For an isochoric process  $\Delta V = 0$ ,

$$\text{Therefore } dW = PdV = 0$$

From first law of thermodynamics

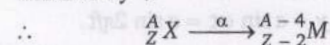
$$dQ = dU + dW$$

$$\text{where } dW = 0$$

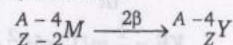
$$\therefore dQ = dU$$

27. **Key Idea :**  $\beta$ -particles are fast moving electrons.

When  $\alpha$ -particle is emitted by an atom, its atomic number decreases by 2 and mass number by 4.



When  $\beta$ -particle is emitted, the atomic number increases by 1, while mass number remains unchanged, therefore



28. The fringe width ( $W$ ) is given by

$$W = \frac{D\lambda}{d}$$

where  $\lambda$  is wavelength,  $d$  the distance between coherent sources,  $D$  the distance between source and screen.

$$\therefore \frac{W_R}{W_V} = \frac{\lambda_R}{\lambda_V}$$

$$\lambda_R \text{ (red light) is } 6400\text{--}7900 \text{ \AA},$$

$$\lambda_V \text{ (violet light) is } 4000\text{--}4500 \text{ \AA}$$

$$\therefore W_R \approx 2W_V$$

**29. Key Idea :** Power =  $\frac{1}{\text{focal length (m)}}$

From lens formula

$$\frac{1}{f} = \frac{1}{v} - \frac{1}{u}$$

Given,  $v = -100 \text{ cm}, u = -40 \text{ cm}$

$$\therefore \frac{1}{f} = -\frac{1}{100} - \frac{1}{-40}$$

$$\Rightarrow \frac{1}{f} = -\frac{1}{100} + \frac{1}{40}$$

$$\Rightarrow \frac{1}{f} = \frac{-1 + 2.5}{100} = \frac{1.5}{100}$$

Hence, power of lens is

$$P = \frac{100}{f(\text{cm})} = 100 \times \frac{1.5}{100} = 1.5 \text{ D}$$

**30. Key Idea :**  $E = -\frac{dV}{dx}$

Electric field  $E$  is given by

$$E = -\frac{dV}{dx}$$

Given,  $V = 5x^2 + 10x - 4$

$$\therefore E = \frac{dV}{dx} = 10x + 10$$

At  $x = 1, E = -(10 \times 1 + 10) = -20 \text{ V/m}$

**31.** The bombardment of fluorescent material with electron radiation cause fluorescence in image convertor tube.

**32.** For a particle executing SHM the displacement equation is given by

$$y = a \sin \omega t = a \sin 2\pi ft$$

$$\text{kinetic energy } KE = \frac{1}{2}mv^2 = \frac{1}{2}m\left(\frac{dy}{dt}\right)^2$$

$$\text{where } \frac{dy}{dt} = 2\pi fa \cos 2\pi ft$$

$$\therefore KE = \frac{1}{2}m\{(2\pi fa)^2 \cos^2 2\pi ft\}$$

$$KE \propto (1 + \cos 4\pi ft)$$

$\cos 4\pi ft$  changes periodically with frequency  $2f$ .

**33. Key Idea :** Work done  $(W) = \text{Force}(F) \times \text{displacement}(d)$

For a body rotating in circular motion, centripetal force is directed towards the centre of the circle when body completes one rotation, its displacement is zero, hence

Work done = force  $\times$  displacement

$$W = F \times 0 = 0$$

**34. Key Idea :** Angular momentum  $\vec{L} = m(\vec{r} \times \vec{v})$

For a body of mass  $m$  rotating with velocity  $v$  in a circle of radius  $r$ , the angular momentum is given by

$$\vec{L} = m(\vec{r} \times \vec{v})$$

For unit mass  $m = 1$

$$\therefore |\vec{L}| = (8\hat{i} - 4\hat{j}) \times (8\hat{i} + 4\hat{j})$$

$$|\vec{L}| = \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 8 & -4 & 0 \\ 8 & 4 & 0 \end{vmatrix}$$

$$|\vec{L}| = \hat{i}(0 - 0) - \hat{j}(0 - 0) + \hat{k}(32 + 32)$$

$$\Rightarrow |\vec{L}| = 64\hat{k} \text{ unit.}$$

**35.** Hydrogen bomb is also known as nuclear fusion bomb based upon the fusion of heavy hydrogen nuclei since fusion takes place under the extreme conditions of high pressure and high temperature, a fission bomb must be used as igniter of a fusion bomb.

**36.** The peak value of current ( $I_0$ ) is given by

$$I_0 = \sqrt{2} I_{\text{rms}}$$

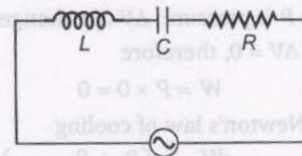
where  $I_{\text{rms}}$  is known as root mean square value of current.

Given,  $I_{\text{rms}} = 4 \text{ A}$

$$\therefore I_0 = \sqrt{2} \times 4 = 5.6 \text{ A}$$

**37.** For an LCR circuit

the impedance ( $Z$ ) is given by



$$Z = \sqrt{R^2 + (X_L - X_C)^2}$$

where  $X_L = \omega L = 2\pi fL$

$$\text{and } X_C = \frac{1}{\omega C} = \frac{1}{2\pi fC}$$

Given,  $f = \frac{50}{\pi} \text{ Hz}, R = 300 \Omega, L = 1 \text{ H},$

$$C = 20 \mu\text{F} = 20 \times 10^{-6} \text{ F.}$$

$$\therefore Z = \sqrt{(300)^2 + \left(2\pi \times \frac{50}{\pi} \times 1 - \frac{1}{2\pi \times \frac{50}{\pi} \times 20 \times 10^{-6}}\right)^2}$$



$$Z = \sqrt{90,000 + (100 - 500)^2}$$

$$Z = \sqrt{90,000 + 16,0000} = 500 \Omega$$

Hence, current in circuit is given by

$$i = \frac{V}{Z} = \frac{50}{500} = 0.1 \text{ A}$$

Voltage across capacitor is,

$$V_C = iX_C = \frac{i}{2\pi fC} = \frac{0.1}{2\pi \times \frac{50}{\pi} \times 20 \times 10^{-6}}$$

$$= \frac{0.1 \times 10^6}{100 \times 20}$$

$$\Rightarrow V_C = 50 \text{ V.}$$

38. The resultant force is given by

$$R = \sqrt{A^2 + B^2 + 2AB \cos \theta}$$

Given,  $A = 2 \text{ N}$ ,  $B = 2 \text{ N}$ ,  $\theta = 60^\circ$

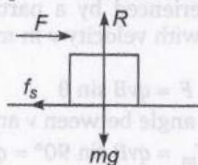
$$\therefore R = \sqrt{2^2 + 2^2 + 2 \times 2 \times 2 \cos 60^\circ}$$

$$\Rightarrow R = \sqrt{16 + 4}$$

$$\Rightarrow R = \sqrt{20}$$

$$\Rightarrow R = 2\sqrt{5} \text{ N}$$

39. Force that produces acceleration is



$$F' = F - \mu R$$

Given,  $F = 100 \text{ N}$ ,

$$F = \mu R = 0.5 \times 10 \times 10$$

$$\therefore F' = 100 - 50 = 50 \text{ N}$$

$$\text{Also, acceleration} = \frac{F'}{m} = \frac{50}{10} = 5 \text{ m/s}^2$$

40. **Key Idea :** Average power dissipated depends on phase difference between voltage and current.

Average power dissipated in an AC circuit is given by.

$$P = VI \cos \phi$$

where  $V$  is voltage,  $I$  the current,  $\phi$  the phase difference

$$\text{Given, } V = 5 \cos 2\pi ft$$

$$\text{Using } \sin(90^\circ + \theta) = \cos \theta$$

we have

$$V = 5 \sin \left( 2\pi ft + \frac{\pi}{2} \right)$$

$$\text{and } I = 2 \sin 2\pi ft$$

Hence, phase difference between  $V$  and  $I$  is

$$\phi = \frac{\pi}{2}$$

$$p = VI \cos \phi = VI \cos \frac{\pi}{2} = 0$$

**Note :** The given circuit is a wattless circuit, because average power dissipated is zero.

41. **Key Idea :** Intensity  $\propto (\text{amplitude})^2$ .

We know that

$$I = ka^2$$

where  $a$  is amplitude and  $I$  the intensity

$$\frac{I_{\max}}{I_{\min}} = \frac{(a_1 + a_2)^2}{(a_1 - a_2)^2}$$

$$\text{Given } \frac{I_1}{I_2} = \frac{16}{9} = \frac{a_1^2}{a_2^2}$$

$$\Rightarrow \frac{a_1}{a_2} = \frac{4}{3}$$

$$\therefore \frac{I_{\max}}{I_{\min}} = \frac{(4+3)^2}{(4-3)^2} = \frac{7^2}{1} = \frac{49}{1}$$

42. **Key Idea :** Image is formed at focus.

We know that

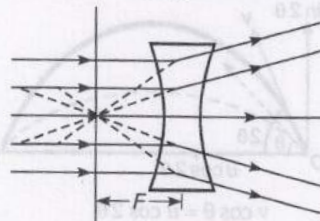
$$\text{Power } (P) = \frac{1}{\text{focal length}(m)}$$

$$\text{Given } P = -2 \text{ D}$$

$$\therefore \text{Focal length} = -\frac{1}{2} = -0.5 \text{ m}$$

$$\Rightarrow F = -50 \text{ cm}$$

Hence, maximum distance of an object, which he can see without spectacles is 50 cm.



43. The first overtone of a wave on string is given by

$$n = \frac{v}{\lambda}$$

where  $\lambda$  is wavelength,  $v$  the velocity and  $n$  the frequency.

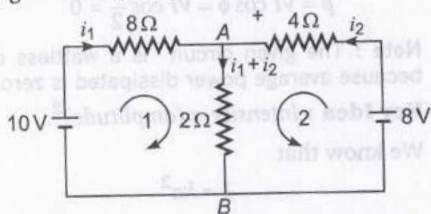
Given,  $\lambda = l$  (in first overtone)

$$\therefore n = \frac{v}{l}$$

$$\Rightarrow v = nl = 250 \times 2 = 500 \text{ m/s.}$$

**44. Key Idea :** As per Kirchhoff's law  $\sum iR = V$

Let  $i_1, i_2$  be current in the two loops of the given circuit then



Applying Kirchhoff's law to loop 1, we get

$$(\sum iR = V)$$

$$8i_1 + 2(i_1 + i_2) = 10$$

$$\Rightarrow 10i_1 + 2i_2 = 10$$

$$\Rightarrow 5i_1 + i_2 = 5 \quad \dots(i)$$

Applying Kirchhoff's law to loop 2, we get

$$4i_2 + 2(i_1 + i_2) = 8$$

$$\Rightarrow 6i_2 + 2i_1 = 8$$

$$\Rightarrow i_1 + 3i_2 = 4 \quad \dots(ii)$$

From Eqs. (i) and (ii), we get

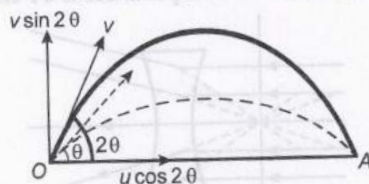
$$i_1 = \frac{11}{14} \text{ A}, i_2 = \frac{15}{14} \text{ A}$$

Hence current flowing in arm AB is

$$i_1 + i_2 = \frac{11}{14} + \frac{15}{14} = \frac{26}{14} = \frac{13}{7} \text{ A}$$

**45. Key Idea :** Horizontal component of velocity remains same.

Let  $v$  be the velocity, when projected with angle  $\theta$ , then equating the horizontal velocities in both the cases, we get



$$v \cos \theta = u \cos 2\theta$$

$$v = \frac{u \cos 2\theta}{\cos \theta}$$

$$\text{where } \sec \theta = \frac{1}{\cos \theta}$$

$$\therefore v = u \cos 2\theta \sec \theta$$

$$\text{Using } \cos 2\theta = 2\cos^2 \theta - 1, \text{ we get}$$

$$\text{Given, } u = 4 \text{ m/s, we get}$$

$$v = 4(2\cos^2 \theta - 1) \sec \theta$$

$$\Rightarrow v = 4(2\cos \theta - \sec \theta)$$

**46. Key Idea :** Path difference is odd multiple of  $\frac{\lambda}{2}$ .

$$\text{Given path difference} = (2n - 1) \frac{\lambda}{2}$$

which is an odd multiple of  $\frac{\lambda}{2}$ .

Hence, destructive interference takes place and dark fringe is formed.

**47. Key Idea :** Resultant intensity is given by

$$I_R = I_1 + I_2 + 2\sqrt{I_1 I_2} \cos \phi$$

Resultant intensity due to two waves of intensities  $I_1, I_2$  is given by

$$I_R = I_1 + I_2 + 2\sqrt{I_1 I_2} \cos \phi$$

when path difference is  $\lambda, \phi = 2\pi$

when path difference is  $\frac{\lambda}{2}, \phi = \pi$

$$\therefore I_{R1} = I + I + 2I \cos 2\pi$$

$$I_{R1} = 4I = K$$

$$I_{R2} = I + I + 2I \cos \pi$$

$$I_{R2} = 0$$

**48. Key Idea :** Maximum force is experienced when angle between  $v$  and  $B$  is  $90^\circ$ .

Force experienced by a particle of charge  $q$ , travelling with velocity  $v$  in magnetic field  $B$  is given by

$$F = qvB \sin \theta$$

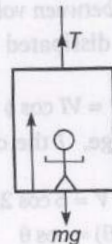
where  $\theta$  is angle between  $v$  and  $B$

$$\therefore F_m = qvB \sin 90^\circ = qvB$$

**49. Key Idea :** Lead accumulator battery is a DC power supply source.

Transformer is based on the principle of mutual induction and works on AC power supply since lead accumulator is a DC source of supply, hence voltage across secondary is zero.

**50.** The resultant acceleration ( $a$ ) is provided by resultant of tension ( $T$ ) and weight ( $mg$ ). Therefore,



$$T - mg = ma$$

$$\Rightarrow T = m(g + a)$$

$$\text{Given } m = 500 \text{ kg, } g = 10 \text{ m/s}^2, a = 2 \text{ m/s}^2$$

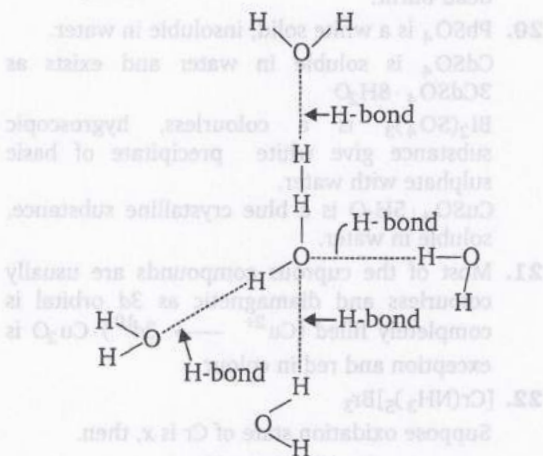
$$\therefore T = 500(10 + 2)$$

$$\Rightarrow T = 6000 \text{ N}$$

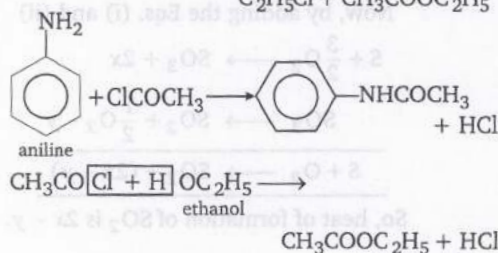
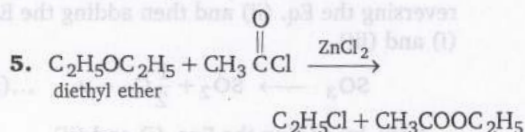


## Chemistry

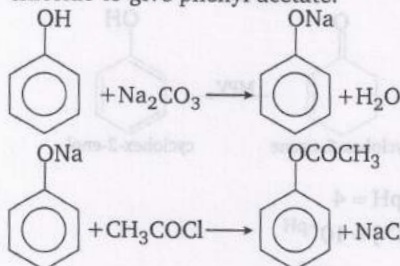
1. A water molecule can form maximum four hydrogen bonds which is clear from the given structure



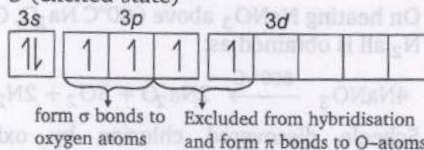
2. The H—O—H bond angle in water is  $104^{\circ}28'$ , in accordance with the VSEPR prediction of slightly less than tetrahedral due to the presence of lone pairs of electrons on oxygen atom. Thus the orbitals used for bonding by O are close to  $sp^3$  hybrids.
3.  $\text{ZnCO}_3$  is known as calamine. It is an important ore of zinc. It is also called smithsonite in the USA.
4. Nitrogen forms trihalides of the formula  $\text{NX}_3$ . Among halides of nitrogen,  $\text{NF}_3$  is extremely stable due to strong N—F bond in  $\text{NF}_3$ . Small size of both nitrogen and fluorine atom causes strength of N—F bond.  $\text{NCl}_3$  is explosive yellow oil.  $\text{NBr}_3$  and  $\text{NI}_3$  are unstable compounds and known only as ammonia complexes e.g.,  $\text{NBr}_3 \cdot 6\text{NH}_3$ .



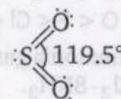
Acetyl chloride does not react directly with phenol. Firstly phenol reacts with alkali to form sodium phenoxide which reacts with acetyl chloride to give phenyl acetate.



6. S (excited state) =



It has two bonds and one lone pair. The S—O double bond arises from  $p\pi - d\pi$  bonding due to lateral overlap of  $p$ -orbitals of oxygen with  $d$ -orbitals of sulphur. The bond between S and O results from  $sp^2$  hybridization in S-atom.



7. Alkali metal halides have high melting point. The melting points of alkali metal chloride decrease from NaCl to CsCl. (except Li).
- |       |        |        |       |       |
|-------|--------|--------|-------|-------|
| LiCl  | NaCl   | KCl    | RbCl  | CsCl  |
| 833 K | 1081 K | 1043 K | 990 K | 918 K |
- So, NaCl has highest melting point and CsCl has lowest melting point.

8.  $t_{1/2} = 6 \text{ min}$

$$N_0 = 32 \text{ g}$$

$$\text{Total time} = 18 \text{ min}$$

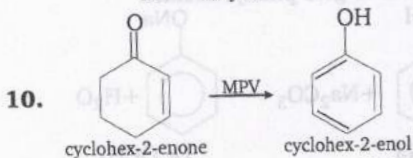
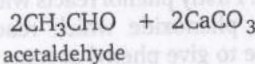
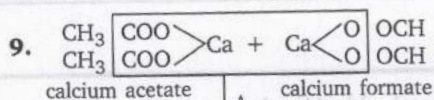
$$\frac{N}{N_0} = \left(\frac{1}{2}\right)^n$$

$$n = \frac{\text{Total time}}{\text{Half-life}} = \frac{18}{6} = 3$$

$$\frac{N}{32} = \left(\frac{1}{2}\right)^3$$

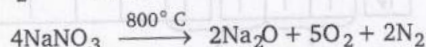
$$\frac{N}{32} = \frac{1}{8}$$

$$N = \frac{32}{8} = 4 \text{ g}$$

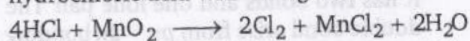


11.  $\text{pH} = 4$   
 $[\text{H}^+] = 10^{-\text{pH}}$   
 $= 10^{-4}$

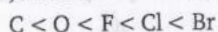
12. On heating  $\text{NaNO}_3$  above  $600^\circ\text{C}$   $\text{Na}_2\text{O}$ ,  $\text{O}_2$  and  $\text{N}_2$  all is obtained as:



13. Scheele discovered chlorine by oxidising hydrochloric acid with manganese dioxide



14. Atomic radius decreases in period from left to right and increase in a group from top to bottom. So, the correct answer is:



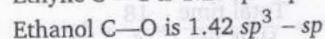
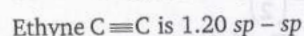
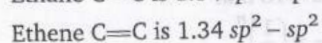
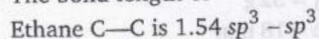
15.  $\text{CaCl}_2$  absorbs ammonia gas to form an addition compound  $\text{CaCl}_2 \cdot 8\text{NH}_3$ .



16. No. of atoms = No. of gram molecular weight  $\times 6.02 \times 10^{23} \times \text{atomicity}$

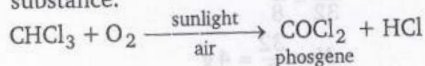
No. of atoms in 1.25 g of  $\text{NH}_3$   
 $= \frac{1.25}{17} \times 6.02 \times 10^{23} \times 4$   
 $= 2.2 \times 10^{23}$

17. The bond length of

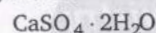


So, the smallest bond angle is of ethyne.

18. Chloroform when exposed to air and sunlight, oxidised to phosgene, a highly poisonous substance.



19. Gypsum is hydrated calcium sulphate.



Gypsum when heated strongly loses whole of its water of hydration. It does not set like plaster of Paris and so, gypsum is said to be dead burnt.

20.  $\text{PbSO}_4$  is a white solid, insoluble in water.

$\text{CdSO}_4$  is soluble in water and exists as  $3\text{CdSO}_4 \cdot 8\text{H}_2\text{O}$

$\text{Bi}_2(\text{SO}_4)_3$  is a colourless, hygroscopic substance give white precipitate of basic sulphate with water.

$\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  is a blue crystalline substance, soluble in water.

21. Most of the cuprous compounds are usually colourless and diamagnetic as  $3d$  orbital is completely filled ( $\text{Cu}^{2+} \longrightarrow 3d^{10}$ ).  $\text{Cu}_2\text{O}$  is exception and red in colour.

22.  $[\text{Cr}(\text{NH}_3)_5]\text{Br}_3$

Suppose oxidation state of Cr is  $x$ , then.

$x + (5 \times 0) + (-1 \times 3) = 0$

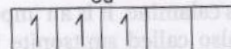
$x - 3 = 0$

$x = +3$

$_{24}\text{Cr} = 1s^2, 2s^2, 2p^6, 3s^2, 3p^6, 3d^5, 4s^1$

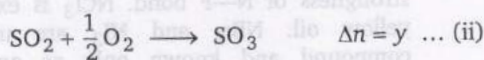
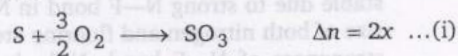
$\text{Cr}^{3+} = 1s^2, 2s^2, 2p^6, 3s^2, 3p^6, 3d^3$

$3d^3$

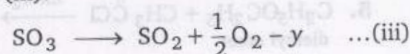


So, three unpaired electrons are present in  $[\text{Cr}(\text{NH}_3)_5]\text{Br}_3$ .

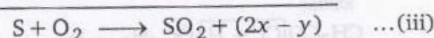
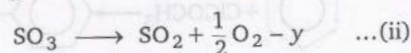
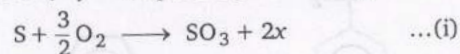
23. Given:



Heat of formation of  $\text{SO}_2$  can be calculated by reversing the Eq. (ii) and then adding the Eqs. (i) and (iii).



Now, by adding the Eqs. (i) and (ii)

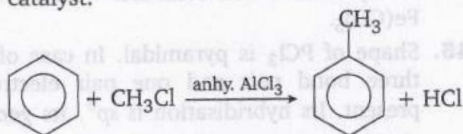


So, heat of formation of  $\text{SO}_2$  is  $2x - y$ .

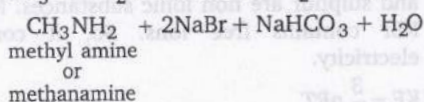


**24. Friedel-Craft's reaction:** In this reaction alkyl group is introduced in the benzene ring in presence of anhy.  $\text{AlCl}_3$

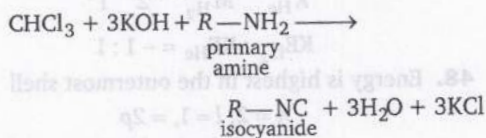
**25.** Even at  $400-500^\circ\text{C}$  the rate of reaction is very low. Therefore, to increase the reaction velocity a suitable catalyst is used. Now a days most of the sulphuric acid plants use  $\text{V}_2\text{O}_5$  as a catalyst.



**26.** When acetamide is treated with  $\text{NaBr}$  or  $\text{Br}_2/\text{NaOH}$ , Hofmann degradation reaction takes place and a primary amine is obtained as:  
 $\text{CH}_3\text{CONH}_2 + \text{NaOBr}$  or  $\text{Br}_2/\text{NaOH} \longrightarrow$



**27.** Isocyanides are prepared by carbylamine reaction. In this treated with chloroform, and  $\text{NaOH}$  or  $\text{KOH}$  to give isocyanide. (unpleasant smelling compound)



**28.** If  $r_x = r$  then  $r_{\text{CH}_4} = 2r$   
 $M_{\text{CH}_4} = 12 + 4 = 16$   
 $M_x = ?$

We know that,

$$\frac{r_{\text{CH}_4}}{r_x} = \sqrt{\frac{M_x}{M_{\text{CH}_4}}}$$

$$\frac{2r}{r} = \sqrt{\frac{M_x}{16}}$$

$$(2)^2 = \frac{M_x}{16}$$

$$4 = \frac{M_x}{16}$$

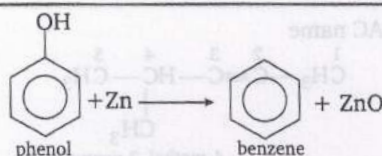
or

$$M_x = 64$$

**29.** Lewis acids are the compounds having a tendency to accept a pair of electron. e.g.,  $\text{BF}_3$ ,  $\text{AlCl}_3$ ,  $\text{BeF}_2$ , etc. So,  $\text{SnCl}_4$  is not a Lewis acid.

**30.** Natural gas mainly consists of  $\text{CH}_4$  (methane)

**31.** Phenol when treated with  $\text{Zn}$  powder, benzene is obtained as :



**32.** Isoelectronic species are the species having same number of electrons.

$$\text{Number of electrons in CO} = 6 + 8 = 14$$

$$\text{Number of electrons in } \text{N}_2^- = 14 + 1 = 15$$

$$\text{Number of electrons in } \text{N}_2^+ = 14 - 1 = 13$$

$$\text{Number of electrons in } \text{CN}^- = 6 + 7 + 1 = 14$$

$$\text{Number of electrons in NO} = 7 + 8 = 15$$

So,  $\text{CN}^-$  and  $\text{CO}$  are isoelectronic species.

**33.** Firstly molarity is converted into normality as  
 Molarity  $\times$  mol. wt. = Normality  $\times$  Eq. wt.

For  $\text{H}_2\text{SO}_4$

$$1 \times 98 = N \times 49$$

$$N = \frac{98}{49} = 2$$

For  $\text{NaOH}$ ,

$$1\text{M} = 1\text{N}$$

Now from normality equation

$$\text{H}_2\text{SO}_4 = \text{NaOH}$$

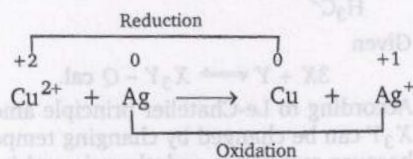
$$N_1 V_1 = N_2 V_2$$

$$2 \times V_1 = 1 \times 20$$

$$V_1 = \frac{20}{2} = 10 \text{ mL of } \text{H}_2\text{SO}_4$$

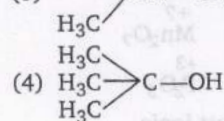
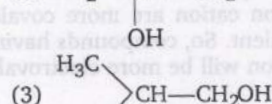
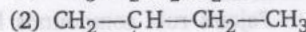
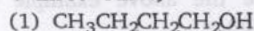
**34.** A solution of  $\text{H}_2\text{O}_2$  with  $\text{Fe}^{2+}$  (Ferrous) salt is known as Fenton's reagent. ( $\text{FeSO}_4 + \text{H}_2\text{O}_2$ )

**35.**

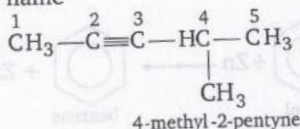


So, oxidation half reaction is  $\text{Ag} \longrightarrow \text{Ag}^+$

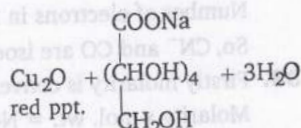
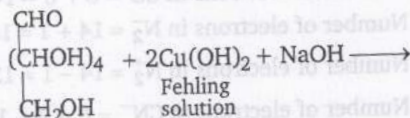
**36.**  $\text{C}_4\text{H}_{10}\text{O}$  form total seven isomers (four alcohol + three ether)



37. IUPAC name



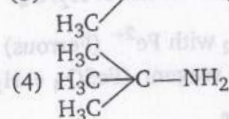
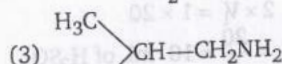
4-methyl-2-pentyne

38. Glucose reduces Fehling solution to give red ppt. of  $\text{Cu}_2\text{O}$  as:

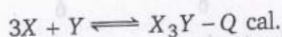
Fructose and sucrose does not respond to Fehling solution.

39. Number of primary amines formed from  $\text{C}_4\text{H}_{11}\text{N}$  is four:

- as  
(1)  $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{NH}_2$   
(2)  $\text{CH}_3 - \underset{\text{NH}_2}{\text{CH}} - \text{CH}_2 - \text{CH}_3$



40. Given

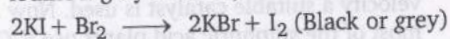


According to Le-Chatelier principle amount of  $X_3Y$  can be changed by changing temperature because reaction is endothermic and it can be changed by changing in pressure because number of moles of reactant and products are different. But can never change by adding catalyst because the above example is (a) reversible reaction.

41. According to Fajan's rule compounds having large charge on cation are more covalent or least electrovalent. So, compounds having low charge on cation will be more electrovalent or ionic.

So,  $\text{P}_2\text{O}_3$  will be most ionic.42.  $\text{CuSO}_4$  reacts with KI to give  $\text{I}_2$  as:

43. Bromine liberates iodine from KI solution because of its oxidising property. Colour of iodine is grey black so, black colour is obtained.

44. Finely divided iron combines with CO to give  $\text{Fe}(\text{CO})_5$ .45. Shape of  $\text{PCl}_3$  is pyramidal. In case of  $\text{PCl}_3$ , three bond pair and one pair electrons is present. Its hybridisation is  $sp^3$ . Its geometry should be tetrahedral but actually it is **pyramidal** due to the presence of **lone pair of electron**.

46. Solid NaCl contains immobile ions. Diamond and sulphur are non ionic substances. Molten KBr contains free ions. So, it conducts electricity.

$$47. \text{KE} = \frac{3}{2} nRT$$

$$\text{KE} \propto n$$

$$\text{KE} \propto \frac{w}{M}$$

$$\frac{\text{KE}_{\text{H}_2}}{\text{KE}_{\text{He}}} = \frac{M_{\text{He}}}{M_{\text{H}_2}} = \frac{2}{2} = 1$$

$$\text{KE}_{\text{H}_2} : \text{KE}_{\text{He}} = -1 : 1$$

48. Energy is highest in the outermost shell

$$n = 2, l = 1, = 2p$$

$$n = 3, l = 2 = 3d$$

$$n = 3, l = 1 = 3p$$

$$n = 2, l = 0 = 2s$$

According to aufbau's principle 3d has highest energy

49. Pyrophosphoric acid is



$$(+1 \times 4) + 2x + (-2 \times 7) = 0$$

$$+ 4 + 2x - 14 = 0$$

$$2x = 14 - 4$$

$$2x = 10$$

$$x = +5$$

50. First order kinetic equation is:

$$k = \frac{2.303}{t} \log \frac{a}{(a-x)}$$

$$\text{or } t = \frac{2.303}{k} \log \frac{a}{(a-x)}$$

According to question

$$8 = \frac{2.303}{k} \log \frac{100}{100-75}$$



$$8 = \frac{2.303}{k} \log \frac{100}{25}$$

$$8 = \frac{2.303}{k} \log 4 \quad \dots(i)$$

And

$$t = \frac{2.303}{k} \log \frac{100}{100-50}$$

$$= \frac{2.303}{k} \log 2 \quad \dots(ii)$$

By dividing Eq. (ii) by Eq. (i), we get

$$\frac{t}{8} = \frac{\frac{2.303}{k} \log 2}{\frac{2.303}{k} \times 2 \log 2}$$

$$\frac{t}{8} = \frac{\log 2}{2 \log 2}$$

$$\frac{t}{8} = \frac{1}{2}$$

$$2t = 8$$

$$t = \frac{8}{2} = 4 \text{ min}$$

## Mathematics

1. We have,
- $\alpha^2 = 5\alpha - 3$

$$\Rightarrow \alpha^2 - 5\alpha + 3 = 0$$

$$\Rightarrow \alpha = \frac{5 \pm \sqrt{13}}{2}$$

Similarly,  $\beta^2 = 5\beta - 3$ 

$$\Rightarrow \beta = \frac{5 \pm \sqrt{13}}{2}$$

Since,

$$\alpha \neq \beta$$

$$\therefore \alpha = \frac{5 + \sqrt{13}}{2} \text{ and } \beta = \frac{5 - \sqrt{13}}{2}$$

or

$$\alpha = \frac{5 - \sqrt{13}}{2}$$

and

$$\beta = \frac{5 + \sqrt{13}}{2}$$

$$\therefore \alpha^2 + \beta^2 = \frac{50 + 26}{4} = 19$$

and

$$\alpha\beta = \frac{1}{4} (25 - 13) = 3$$

Thus, the equation having  $\frac{\alpha}{\beta}$  and  $\frac{\beta}{\alpha}$  as its roots is

$$x^2 - x \left( \frac{\alpha}{\beta} + \frac{\beta}{\alpha} \right) + \frac{\alpha\beta}{\alpha\beta} = 0$$

$$\Rightarrow x^2 - x \left( \frac{\alpha^2 + \beta^2}{\alpha\beta} \right) + 1 = 0$$

$$\Rightarrow 3x^2 - 19x + 3 = 0$$

2. We have,
- $y = (x + \sqrt{1+x^2})^n \quad \dots(i)$

$$\text{Let } \frac{d^2y}{dx^2} = y_2$$

$$\text{and } \frac{dy}{dx} = y_1$$

On differentiating Eq. (i), we get

$$\frac{dy}{dx} = n(x + \sqrt{1+x^2})^{n-1} \left( 1 + \frac{x}{\sqrt{x^2+1}} \right)$$

$$= \frac{n(x + \sqrt{1+x^2})^n}{\sqrt{1+x^2}}$$

$$\Rightarrow \frac{dy}{dx} = \frac{ny}{\sqrt{1+x^2}}$$

$$\Rightarrow y_1^2 (1+x^2) = n^2 y^2$$

Again differentiating, we get

$$2y_1 y_2 (1+x^2) + 2xy_1^2 = 2n^2 y y_1$$

On dividing by  $2y_1$ , we get

$$y_2 (1+x^2) + xy_1 = n^2 y$$

$$\Rightarrow \frac{d^2y}{dx^2} (1+x^2) + x \frac{dy}{dx} = n^2 y$$

3. 1,
- $\log_3 \sqrt{3^{1-x} + 2}$
- ,
- $\log_3 (4 \cdot 3^x - 1)$
- are in AP

$$\therefore 2 \log_3 (3^{1-x} + 2)^{1/2} = 1 + \log_3 (4 \cdot 3^x - 1)$$

$$\log_3 (3^{1-x} + 2) = \log_3 3 + \log_3 (4 \cdot 3^x - 1)$$

$$\log_3 (3^{1-x} + 2) = \log_3 [3(4 \cdot 3^x - 1)]$$

$$3^{1-x} + 2 = 3(4 \cdot 3^x - 1)$$

$$3 \cdot 3^{-x} + 2 = 12 \cdot 3^x - 3$$

Let  $3^x = t$ 

$$\therefore \frac{3}{t} + 2 = 12t - 3$$

$$3 + 2t = 12t^2 - 3t$$

$$12t^2 - 5t - 3 = 0$$

$$(12t^2 - 9t + 4t - 3) = 0$$

$$3t(4t - 3) + 1(4t - 3) = 0$$

$$\therefore t = -\frac{1}{3}, \frac{3}{4}$$

$$3^x = \frac{3}{4}$$

$$\Rightarrow x = \log_3 \left( \frac{3}{4} \right) = \log_3 3 - \log_3 4$$

$$\Rightarrow x = 1 - \log_3 4$$

$$4. P(E_1) = \frac{1}{2}, P(E_2) = \frac{1}{3} \text{ and } P(E_3) = \frac{1}{4}$$

$$P(E_1 \cup E_2 \cup E_3) = 1 - P(\bar{E}_1)P(\bar{E}_2)P(\bar{E}_3)$$

$$= 1 - \left(1 - \frac{1}{2}\right)\left(1 - \frac{1}{3}\right)\left(1 - \frac{1}{4}\right)$$

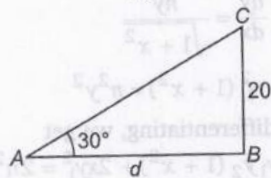
$$= 1 - \frac{1}{2} \times \frac{2}{3} \times \frac{3}{4} = \frac{3}{4}$$

**Note :** If the events  $E_1, E_2$  and  $E_3$  are independent, then  $\bar{E}_1, \bar{E}_2$  and  $\bar{E}_3$  are also independent.

5. In  $\Delta ABC$ ,

$$\tan 30^\circ = \frac{BC}{AB}$$

$$\Rightarrow \frac{1}{\sqrt{3}} = \frac{20}{d}$$



$$\Rightarrow d = 20\sqrt{3} \text{ m}$$

6. Let the first term and common ratio of a GP be A and R respectively.

$$l = AR^{p-1} \Rightarrow \log l = \log A + (p-1) \log R$$

$$m = AR^{q-1} \Rightarrow \log m = \log A + (q-1) \log R$$

$$n = AR^{r-1} \Rightarrow \log n = \log A + (r-1) \log R$$

Now,

$$\begin{vmatrix} \log l & p & 1 \\ \log m & q & 1 \\ \log n & r & 1 \end{vmatrix}$$

$$= \begin{vmatrix} \log A + (p-1) \log R & p & 1 \\ \log A + (q-1) \log R & q & 1 \\ \log A + (r-1) \log R & r & 1 \end{vmatrix} = 0$$

$$7. \lim_{x \rightarrow 0} \frac{\sqrt{1 - \cos 2x}}{\sqrt{2x}}$$

$$= \lim_{x \rightarrow 0} \frac{\sqrt{1 - (1 - 2 \sin^2 x)}}{\sqrt{2x}}$$

$$= \lim_{x \rightarrow 0} \frac{\sqrt{2 \sin^2 x}}{\sqrt{2x}} = \lim_{x \rightarrow 0} \frac{|\sin x|}{x}$$

$$\text{Let } f(x) = \frac{|\sin x|}{x}$$

$$\text{Now, } f(0+0) = \lim_{h \rightarrow 0} \frac{|\sin(0+h)|}{0+h}$$

$$= \lim_{h \rightarrow 0} \frac{\sin h}{h} = 1$$

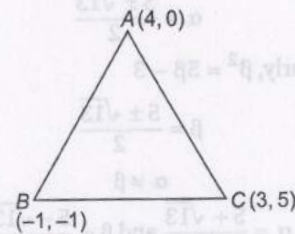
$$\text{and } f(0-0) = \lim_{h \rightarrow 0} \frac{|\sin(0-h)|}{-h} = -1$$

$\therefore$  The limit of function does not exist.

$$8. AB = \sqrt{(4+1)^2 + (0+1)^2} = \sqrt{26}$$

$$BC = \sqrt{(3+1)^2 + (5+1)^2} = \sqrt{52}$$

$$CA = \sqrt{(4-3)^2 + (0-5)^2} = \sqrt{26}$$



So, in isosceles triangles side  $AB = CA$

For right angled triangle

$$BC^2 = AB^2 + AC^2$$

$$\text{So, here } BC = \sqrt{52} = BC^2 = 52$$

$$\text{or } (\sqrt{26})^2 + (\sqrt{26})^2 = 52$$

So, given vertices is right angled and also is isosceles triangle.

$$9. \cot^{-1}(\sqrt{\cos \alpha}) - \tan^{-1}(\sqrt{\cos \alpha}) = x$$

$$\tan^{-1}\left(\frac{1}{\sqrt{\cos \alpha}}\right) - \tan^{-1}(\sqrt{\cos \alpha}) = x$$

$$\Rightarrow \tan^{-1} \frac{1}{\sqrt{\cos \alpha}} - \sqrt{\cos \alpha} = x$$

$$1 + \frac{1}{\sqrt{\cos \alpha}} \sqrt{\cos \alpha} = x$$

$$\Rightarrow \tan^{-1} \frac{1 - \cos \alpha}{2 \sqrt{\cos \alpha}} = x$$

$$\Rightarrow \tan x = \frac{1 - \cos \alpha}{2 \sqrt{\cos \alpha}}$$

$$\Rightarrow \cot x = \frac{2 \sqrt{\cos \alpha}}{1 - \cos \alpha}$$



$$\Rightarrow \operatorname{cosec} x = \frac{1 + \cos \alpha}{1 - \cos \alpha}$$

$$\therefore \sin x = \frac{1 - \cos \alpha}{1 + \cos \alpha}$$

$$= \frac{1 - \left(1 - 2 \sin^2 \frac{\alpha}{2}\right)}{1 + 2 \cos^2 \frac{\alpha}{2} - 1}$$

$$\Rightarrow \sin x = \tan^2 \frac{\alpha}{2}$$

**10. Key Idea :** If a plane contains a line, then the normal to the plane is perpendicular to the line.

Any plane passing through (3, 2, 0) is

$$a(x - 3) + b(y - 2) + c(z - 0) = 0 \quad \dots(i)$$

It passes through (3, 6, 4)

$$\therefore 0 \cdot a + 4b + 4c = 0 \quad \dots(ii)$$

Normal to plane (i) is perpendicular to given line

$$\therefore a + 5b + 4c = 0 \quad \dots(ii)$$

On solving Eqs. (i) and (ii), we get

$$\frac{a}{1} = \frac{b}{-1} = \frac{c}{1} = k$$

So,  $a = k, b = -k, c = k$

Putting the value of  $a, b, c$  in Eq. (i), we get

$$x - y + z = 1$$

**Note :** In any line of a plane it has only one direction cosines and direction ratios may be more than one.

$$11. \frac{d^2y}{dx^2} = e^{-2x}$$

On integrating, we get

$$\frac{dy}{dx} = \frac{e^{-2x}}{-2} + c$$

Again integrating, we get

$$y = \frac{e^{-2x}}{4} + cx + d$$

$$12. \lim_{x \rightarrow \infty} \left[ 1 + \frac{4x+1}{x^2+x+2} \right]^x = \lim_{x \rightarrow \infty} (1 + \alpha)^{1/\alpha} \alpha^{\alpha x}$$

$$\text{where } \alpha = \frac{4x+1}{x^2+x+2}$$

$$= \frac{4 + \frac{1}{x}}{x \left( 1 + \frac{1}{x} + \frac{2}{x^2} \right)} \rightarrow 0 \text{ as } x \rightarrow \infty$$

$$\text{and } \alpha x = \frac{4 + \frac{1}{x}}{1 + \frac{1}{x} + \frac{2}{x^2}} \rightarrow 4 \text{ as } x \rightarrow \infty$$

Given limit =  $e^4$

**13. Key Idea :** The domain of  $\sin^{-1} x$  is  $[-1, 1]$ .

$$f(x) = \sin^{-1} \left( \log_3 \left( \frac{x}{3} \right) \right) \text{ exists, if}$$

$$-1 \leq \log_3 \left( \frac{x}{3} \right) \leq 1$$

$$\Rightarrow 3^{-1} \leq \frac{x}{3} \leq 3^1$$

$$\Rightarrow 1 \leq x \leq 9 \Rightarrow \text{domain is } [1, 9].$$

$$14. 2^{1/4} \cdot 4^{1/8} \cdot 8^{1/16} \dots \infty$$

$$= 2^{1/4} + \frac{2}{8} + \frac{3}{16} + \dots \infty = \frac{1}{2^{2^2}} \left( 1 + \frac{2}{2^2} + \frac{3}{2^3} + \dots \right)$$

$$= \frac{1}{2^2} \left( \frac{1}{1 - \frac{1}{2}} + \frac{1 \cdot \frac{1}{2}}{\left( 1 - \frac{1}{2} \right)^2} \right) = \frac{1}{2^{2^2}} |2+2| = 2^1 = 2$$

$$15. ar^4 = 2 \quad \dots(i)$$

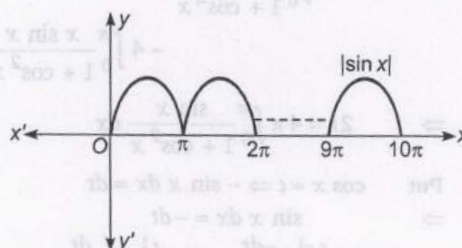
Let  $a$  be the first term and  $r$  be the common ratio, then according to the given condition.

$$\begin{aligned} \text{Since, } a \times ar \times ar^2 \times ar^3 \times ar^4 \times ar^5 \\ \times ar^6 \times ar^7 \times ar^8 \\ = a^9 r^{36} = (ar^4)^9 = 2^9 = 512 \end{aligned}$$

**16. Key Idea :** The period of  $|\sin x|$  is  $\pi$ .

$$\begin{aligned} \int_0^{10\pi} |\sin x| dx \\ = 10 \left[ \int_0^{\pi/2} \sin x dx + \int_{\pi/2}^{\pi} \sin x dx \right] \\ = 10 [-\cos x]_0^{\pi/2} + [-\cos x]_{\pi/2}^{\pi} \\ = 10 [1 + 1] = 10 \times 2 = 20 \end{aligned}$$

**Alternative Solution :**



$\therefore$  Required area

$$\begin{aligned} &= 10 \int_0^{\pi} \sin x dx \\ &= 10 [-\cos x]_0^{\pi} = -10 (\cos \pi - \cos 0) = 20 \end{aligned}$$

$$\begin{aligned}
 17. I_n + I_{n+2} &= \int_0^{\pi/4} \tan^n x (1 + \tan^2 x) dx \\
 &= \int_0^{\pi/4} \tan^n x \sec^2 x dx \\
 &= \int_0^1 t^n dt \quad \text{where } t = \tan x \\
 \Rightarrow I_n + I_{n+2} &= \frac{1}{n+1} \\
 \therefore \lim_{n \rightarrow \infty} n[I_n + I_{n+2}] &= \lim_{n \rightarrow \infty} n \cdot \frac{1}{n+1} = \lim_{n \rightarrow \infty} \frac{n}{n+1} = 1
 \end{aligned}$$

18. **Key Idea :**  $\int_{-\pi}^{\pi} f(x) dx = 2 \int_0^{\pi} f(x) dx$ , if  $f(x)$  is an even function.

$= 0$ , if  $f(x)$  is an odd function.

$$\begin{aligned}
 \text{Let } I &= \int_{-\pi}^{\pi} \frac{2x(1 + \sin x)}{1 + \cos^2 x} dx \\
 \Rightarrow I &= \int_{-\pi}^{\pi} \frac{2x}{1 + \cos^2 x} dx + 2 \int_{-\pi}^{\pi} \frac{x \sin x}{1 + \cos^2 x} dx
 \end{aligned}$$

$$\Rightarrow I = 0 + 4 \int_0^{\pi} \frac{x \sin x}{1 + \cos^2 x} dx$$

( $\because \frac{2x}{1 + \cos^2 x} dx$  is an odd function)

$$\Rightarrow I = 4 \int_0^{\pi} \frac{(\pi - x) \sin(\pi - x)}{1 + \cos^2(\pi - x)} dx$$

$$\Rightarrow I = 4 \int_0^{\pi} \frac{\pi(\pi - x) \sin x}{1 + \cos^2 x} dx$$

$$\begin{aligned}
 \Rightarrow I &= 4\pi \int_0^{\pi} \frac{\sin x}{1 + \cos^2 x} dx \\
 &\quad - 4 \int_0^{\pi} \frac{x \sin x}{1 + \cos^2 x} dx
 \end{aligned}$$

$$\Rightarrow 2I = 4\pi \int_0^{\pi} \frac{\sin x}{1 + \cos^2 x} dx$$

$$\text{Put } \cos x = t \Rightarrow -\sin x dx = dt$$

$$\Rightarrow \sin x dx = -dt$$

$$\therefore I = 2\pi \int_1^{-1} \frac{-dt}{1+t^2} = 2\pi \int_{-1}^1 \frac{dt}{1+t^2}$$

$$= 2\pi [\tan^{-1} t]_{-1}^1$$

$$= 2\pi [\tan^{-1}(1) - \tan^{-1}(-1)]$$

$$= 2\pi \left[ \frac{\pi}{4} + \frac{\pi}{4} \right] = 2\pi \times \frac{\pi}{2} = \pi^2$$

19. **Key Idea :** The period of  $\sin^4 x$  or  $\cos^4 x$  is  $\frac{\pi}{2}$ .

$$\begin{aligned}
 f(x) &= \sin^4 x + \cos^4 x \\
 &= (\sin^2 x + \cos^2 x)^2 - 2\sin^2 x \cos^2 x \\
 &= 1 - \frac{1}{2}(\sin 2x)^2 \\
 &= 1 - \frac{1}{2} \left[ \frac{1 - \cos 4x}{2} \right] \\
 &= \frac{3}{4} + \frac{1}{4} \cos 4x
 \end{aligned}$$

Since,  $\cos x$  is periodic with period  $2\pi$ .

$\therefore$  The period of  $f(x) = \frac{2\pi}{4} = \frac{\pi}{2}$ .

**Alternative Solution :**

Since the period of  $\sin^4 x$  and  $\cos^4 x$  is  $\frac{\pi}{2}$ .

$\therefore$  Period of  $f(x) = \sin^4 x + \cos^4 x$  is the LCM of the period of  $\sin^4 x$  and  $\cos^4 x$ .

$\therefore$  Required period is  $\frac{\pi}{2}$ .

$$20. x^y = e^{x-y}$$

Taking log on both sides, we get

$$y \log x = x - y$$

$$\Rightarrow y(\log x + 1) = x$$

$$\Rightarrow y = \frac{x}{1 + \log x}$$

On differentiating w.r.t  $x$ , we get

$$\Rightarrow \frac{dy}{dx} = \frac{(1 + \log x) - x \times \frac{1}{x}}{(1 + \log x)^2}$$

$$\Rightarrow \frac{dy}{dx} = \frac{\log x}{(1 + \log x)^2}$$

21. **Key Idea :** The angle between the curves is the angle between the tangents to the curve.

$$\text{We have, } x^3 - 3xy^2 + 2 = 0 \quad \dots(i)$$

$$\text{and } 3x^2y - y^3 - 2 = 0 \quad \dots(ii)$$

On differentiating Eqs. (i) and (ii) with respect to  $x$ , we get

$$\left( \frac{dy}{dx} \right)_{c_1} = m_1 = \frac{x^2 - y^2}{2xy}$$

$$\text{and } \left( \frac{dy}{dx} \right)_{c_2} = m_2 = \frac{-2xy}{x^2 - y^2}$$

$$\text{Since } m_1 \times m_2 = -1$$

Hence, the two curves cut at right angles.



- 22. Key Idea :** A function is said to be increasing, if  $f'(x) > 0$  and it said be decreasing if  $f'(x) < 0$ .

Given,  $f(x) = \cot^{-1} x + x$

$$\Rightarrow f'(x) = -\frac{1}{1+x^2} + 1$$

$$= \frac{x^2}{1+x^2}, \text{ clearly, } f'(x) > 0 \text{ for all } x.$$

So,  $f(x)$  increases in  $(-\infty, \infty)$ .

- 23.** We have,

$$f(x) = (x+1)^{1/3} - (x-1)^{1/3}$$

$$\therefore f'(x) = \frac{1}{3} \left[ \frac{1}{(x+1)^{2/3}} - \frac{1}{(x-1)^{2/3}} \right]$$

$$= \frac{(x-1)^{2/3} - (x+1)^{2/3}}{3(x^2-1)^{2/3}}$$

Clearly,  $f'(x)$  does not exists at  $x = \pm 1$  for maxima or minima, put

$$f'(x) = 0, \text{ then } (x-1)^{2/3} = (x+1)^{2/3}$$

$$\Rightarrow x = 0$$

Clearly,  $f'(x) \neq 0$  for any other value of  $x \in [0, 1]$ . The value of  $f(x)$  at  $x = 0$  is 2.

Hence, the greatest value of  $f(x)$  is 2.

**Note :** If any function has no critical point in the given interval  $[a, b]$ , then we check the maximum value of  $x = a, b$ .

**24.** Let  $I = \int \frac{dx}{x(x^n+1)}$

Putting  $x^n + 1 = t$

$$\Rightarrow nx^{n-1}dx = dt$$

$$\therefore I = \frac{1}{n} \int \frac{dt}{t(t-1)}$$

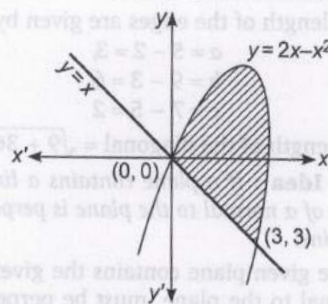
$$= \frac{1}{n} \int \left( \frac{1}{t-1} - \frac{1}{t} \right) dt$$

$$= \frac{1}{n} \log \left( \frac{t-1}{t} \right) + c$$

$$= \frac{1}{n} \log \left( \frac{x^n}{x^n+1} \right) + c$$

**25.** Required area  $= \int_0^3 (y_1 - y_2) dx$

$$= \int_0^3 (2x - x^2) - (-x) dx$$



$$= \int_0^3 (3x - x^2) dx$$

$$= \left[ \frac{3x^2}{2} - \frac{x^3}{3} \right]_0^3$$

$$= \frac{27}{2} - 9$$

$$= \frac{9}{2} \text{ sq unit}$$

- 26.** The general equation of all non-vertical lines in a plane is  $ax + hy = 1$ , where  $h \neq 0$

On differentiating w.r.t  $x$ , we get

$$a + h \frac{dy}{dx} = 0$$

Again differentiating, we get

$$h \frac{d^2y}{dx^2} = 0$$

$$\Rightarrow \frac{d^2y}{dx^2} = 0$$

- 27.** The required vector is along the vector

$$\vec{a} \times (\vec{a} \times \vec{b}) = (\vec{a} \cdot \vec{b}) \vec{a} - (\vec{a} \cdot \vec{a}) \vec{b}$$

$$= -(\hat{i} - \hat{j}) - 2(\hat{i} + 2\hat{j})$$

$$= -3\hat{i} - 3\hat{j}$$

Hence, required vector are given by

$$\pm \frac{(-3\hat{i} - 3\hat{j})}{\sqrt{9+9}} = \pm \frac{1}{\sqrt{2}} (\hat{i} + \hat{j})$$

- 28.** We have,  $2|\hat{i} + x\hat{j} + 3\hat{k}|$

$$= |4\hat{i} + (4x-2)\hat{j} + 2\hat{k}|$$

$$\Rightarrow 4(x^2 + 10) = 20 + (4x-2)^2$$

$$\Rightarrow 3x^2 - 4x - 4 = 0$$

$$\Rightarrow x = 2, -\frac{2}{3}$$

29. The length of the edges are given by

$$a = 5 - 2 = 3,$$

$$b = 9 - 3 = 6$$

and

$$c = 7 - 5 = 2$$

So, length of the diagonal  $= \sqrt{9 + 36 + 4} = 7$

30. **Key Idea :** If a plane contains a line, then the DR's of a normal to the plane is perpendicular to the line.

If the given plane contains the given line, then normal to the plane, must be perpendicular to the line and the condition for the same is

$$al + bm + cn = 0.$$

31. The number of ways in which either player can choose a number from 1 to 25 is 25, so the total number of ways a choosing numbers is  $25 \times 25 = 625$ . So the probability that they will not win a prize in a single trial

$$= 1 - \frac{1}{25} = \frac{24}{25}$$

32. The equation of parabola is

$$(y + 2)^2 = -4 \left( x - \frac{1}{2} \right)$$

Shifting the origin  $\left( \frac{1}{2}, -2 \right)$ , the equation of

parabola becomes  $Y^2 = -4X$

where  $X = x - \frac{1}{2}$ ,  $Y = y + 2$ .

An equation of its directrix is  $X = 1$ .

$\therefore$  Required directrix,  $x = \frac{3}{2}$ .

33. **Key Idea :** For making a triangle, we needs a three vertices of a triangle.

$$\Rightarrow T_{n+1} - T_n = 21$$

$$\Rightarrow {}^{n+1}C_3 - {}^nC_3 = 21 \quad (\because T_n = {}^nC_3)$$

$$\Rightarrow {}^nC_2 + {}^nC_3 - {}^nC_3 = 21 \Rightarrow {}^nC_2 = 21$$

$$\Rightarrow \frac{n(n-1)}{2} = 21$$

$$\Rightarrow n^2 - n - 42 = 0$$

$$\therefore n = 7$$

34. We have,  $A + C = \pi - B$  and

$$\frac{A - B + C}{2} = \frac{\pi}{2} - B$$

$$\therefore 2ca \sin \left( \frac{\pi}{2} - B \right) =$$

$$2ca \cos B = 2ca \frac{c^2 + a^2 - b^2}{2ca} = a^2 + c^2 - b^2$$

$$\begin{aligned} 35. \lim_{x \rightarrow \infty} \left( \frac{x-3}{x+2} \right)^x &= \lim_{x \rightarrow \infty} \left( \frac{1 - \frac{3}{x}}{1 + \frac{2}{x}} \right)^x \\ &= \lim_{x \rightarrow \infty} \frac{\left( 1 - \frac{3}{x} \right)^x}{\left( 1 + \frac{2}{x} \right)^x} \\ &= \frac{e^{-3}}{e^2} = e^{-5} \end{aligned}$$

36. **Key Idea :** If triangle is an equilateral, then orthocentre, incentre and centroid will be same.

Let  $A(1, \sqrt{3})$ ,  $B(0, 0)$ ,  $C(2, 0)$  be the given points.

$$\therefore a = BC$$

$$= \sqrt{(2-0)^2 + (0-0)^2} = 2$$

$$\therefore b = CA$$

$$= \sqrt{(2-1)^2 + (0-\sqrt{3})^2} = 2$$

$$c = AB$$

$$= \sqrt{(0-1)^2 + (0-\sqrt{3})^2} = 2$$

Triangle is equilateral.

$\therefore$  Incentre is the same as centroid of the triangle.

$$\therefore \text{Incentre is } \left( \frac{1+0+2}{3}, \frac{\sqrt{3}+0+0}{3} \right)$$

$$\text{i.e., } \left( 1, \frac{1}{\sqrt{3}} \right)$$

37. We have,  $\vec{a} + \vec{b} + \vec{c} = \vec{0}$

$$\therefore \vec{a} + \vec{b} = -\vec{c}$$

$$\Rightarrow \vec{a} \times \vec{c} + \vec{b} \times \vec{c} = 0$$

$$\Rightarrow \vec{b} \times \vec{c} = \vec{c} \times \vec{a}$$

$$\text{Similarly, } \vec{a} \times \vec{b} = \vec{b} \times \vec{c} = \vec{c} \times \vec{a}$$

38. **Key Idea :** The cubic roots of unity are 1,  $\omega$ ,  $\omega^2$  and  $1 + \omega + \omega^2 = 0$ .

$$\begin{aligned} \text{We have, } (1 + \omega - \omega^2)^7 &= (-\omega^2 - \omega^2)^7 \\ &= (-2)^7 (\omega^2)^7 \\ &= -128 \omega^2 \end{aligned}$$



**39. Key Idea :**  $\sin^2 \theta$  is lies between 0 to 1.

We know that  $\sin^2 \theta \geq 1$

$$\Rightarrow \frac{4xy}{(x+y)^2} \geq 1 \Rightarrow 4xy \geq (x+y)^2$$

$$\Rightarrow (x-y)^2 \leq 0$$

$$\Rightarrow x-y=0$$

$$\Rightarrow y=x \text{ and } x \neq 0, y \neq 0$$

**40.** We have,  $a^2 = 16$

$$\text{and } 9 = b^2 = a^2(1 - e^2)$$

$$\Rightarrow 9 = 16(1 - e^2)$$

$$\therefore e = \frac{\sqrt{7}}{4}$$

Thus, the foci are  $(\pm \sqrt{7}, 0)$ .

The radius of required circle

$$= \sqrt{(\sqrt{7} - 0)^2 + 3^2}$$

$$= \sqrt{7+9} = 4$$

**41.** Applying  $R_1 \rightarrow R_1 + R_3$ , we obtain

$$\begin{vmatrix} 1-i & \omega^2 + \omega & \omega^2 - 1 \\ 1-i & -1 & \omega^2 - 1 \\ -i & -1 + \omega - i & -1 \end{vmatrix} = 0$$

$\therefore \omega^2 + \omega = -1$  which  $R_1$  and  $R_2$  become identical.

$$\text{42. } \lim_{x \rightarrow 2} \frac{x f(2) - 2f(x)}{x-2}$$

$$= \lim_{x \rightarrow 2} \frac{x f(2) - 2f(2) + 2f(2) - 2f(x)}{x-2}$$

$$= \lim_{x \rightarrow 2} \frac{(x-2)f(2)}{x-2} - 2 \lim_{x \rightarrow 2} \frac{f(x) - f(2)}{x-2}$$

$$= f(2) - 2f'(2)$$

$$= 4 - 2 \times 4 = -4$$

**43.** For the two lines  $24x + 7y - 20 = 0$  and  $4x - 3y - 2 = 0$ , the angle bisectors are given by

$$\frac{24x + 7y - 20}{25} = \pm \frac{4x - 3y - 2}{5}$$

Taking positive sign, we get

$$2x + 11y - 5 = 0$$

$\therefore$  The given three lines are concurrent with on line bisecting the angle between the other two.

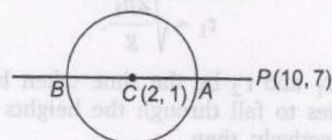
**44.**  $\sqrt{3}x + y = 0$  makes an angle of  $120^\circ$  with  $OX$  and  $\sqrt{3}x - y = 0$  makes an angle  $60^\circ$  with  $OX$ .

So, the required line is  $y - 2 = 0$

**45. Key Idea :** If a point is outside the circle, then the greatest distance is equal to the length of diameter and lowest distance between the circle and a point.

$$\text{Since, } S_1 = 10^2 + 7^2 - 4 \times 10 - 2 \times 7 - 20 > 0.$$

So,  $P$  lies outside the circle. Join  $P$  with the centre  $C(2, 1)$  of the given circle. Suppose  $PC$  cuts the circle at  $A$  and  $B$ , then  $PB$  is the greatest distance of  $P$  from the circle.



$$\text{Now, } PC = \sqrt{(10-2)^2 + (7-1)^2} = 10$$

$$BC = \sqrt{4+1+20} = 5$$

$$\therefore PB = PC + CB$$

$$= 10 + 5 = 15$$

**46.** Let the ellipse be  $\frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$ .

It is given that  $e = \frac{1}{2}$  and  $ae = 2$

Therefore,  $a = 4$

$$\text{Now, } b^2 = a^2(1 - e^2)$$

$$b^2 = 12$$

Thus, the required ellipse is  $\frac{x^2}{16} + \frac{y^2}{12} = 1$

**47.** The mid point of the chord is  $\left(\frac{x_1 + x_2}{2}, \frac{y_1 + y_2}{2}\right)$ . The equation of the chord

in terms of its mid point is  $T = S$ .

$$\text{or } x \left(\frac{y_1 + y_2}{2}\right) + y \left(\frac{x_1 + x_2}{2}\right)$$

$$= 2 \left(\frac{x_1 + x_2}{2}\right) \left(\frac{y_1 + y_2}{2}\right)$$

$$x(y_1 + y_2) + y(x_1 + x_2) = (x_1 + x_2)(y_1 + y_2)$$

$$\frac{x}{x_1 + x_2} + \frac{y}{y_1 + y_2} = 1$$

**48.** We have,

$$\frac{P}{3} = \frac{Q}{7} = \frac{R}{5} = \lambda \quad (\text{say})$$

$$\Rightarrow P = 3\lambda, Q = 7\lambda, R = 5\lambda$$

$$\therefore R^2 = P^2 + Q^2 + 2PQ \cos \theta$$

$$\Rightarrow 25\lambda^2 = 9\lambda^2 + 49\lambda^2 + 42\lambda^2 \cos \theta$$

$$\Rightarrow -33 = 42 \cos \theta$$

$$\Rightarrow \cos \theta = -\frac{11}{14}$$

$$\Rightarrow \theta = \cos^{-1} \left( -\frac{11}{14} \right)$$

**49. Key Idea :** Time taken by the body to fall through the height  $h_1$  is

$$t_1 = \sqrt{\frac{2h_1}{g}}$$

Let  $t_1$  and  $t_2$  be the time taken by the two bodies to fall through the heights  $h_1$  and  $h_2$  respectively, then

$$h_1 = \frac{1}{2}gt_1^2 \text{ and } h_2 = \frac{1}{2}gt_2^2$$

$$\Rightarrow t_1 = \sqrt{\frac{2h_1}{g}}, t_2 = \sqrt{\frac{2h_2}{g}}$$

$$t_1 : t_2 = \sqrt{\frac{2h_1}{g}} : \sqrt{\frac{2h_2}{g}}$$

$$t_1 : t_2 = \sqrt{h_1} : \sqrt{h_2}$$

**50. Key Idea :** If the variance of  $x$  and  $y$  is  $\text{var}(x)$  and  $\text{var}(y)$  and covariance of  $x$  and  $y$  is  $\text{cov}(x, y)$ , then the correlation coefficient is

$$r_{xy} = \frac{\text{cov}(x, y)}{\sqrt{\text{var}(x) \cdot \text{var}(y)}}$$

We know,

$$r_{xy} = \frac{\text{cov}(x, y)}{\sqrt{\text{var}(x) \cdot \text{var}(y)}} = \frac{10.2}{\sqrt{(8.25) \cdot (33.96)}} = \frac{10.2}{16.74} = 0.61$$

**Note :** If two variables are independent, then correlation coefficient is zero.