

# KISHORE VAIGYANIK PROTSAHAN YOJANA - 2012

Date : 04-11-2012

Duration : 3 Hours

Max. Marks : 100

## STREAM - SA

### GENERAL INSTRUCTIONS

- The Test Booklet consists of **80** questions.
- There are Two parts in the question paper. The distribution of marks subjectwise in each part is as under for each correct response.

### MARKING SCHEME :

#### PART-I :

##### MATHEMATICS

Question No. **1 to 15** consist of **ONE (1)** mark for each correct response.

##### PHYSICS

Question No. **16 to 30** consist of **ONE (1)** mark for each correct response.

##### CHEMISTRY

Question No. **31 to 45** consist of **ONE (1)** mark for each correct response.

##### BIOLOGY

Question No. **46 to 60** consist of **ONE (1)** mark for each correct response.

#### PART-II :

##### MATHEMATICS

Question No. **61 to 65** consist of **TWO (2)** marks for each correct response.

##### PHYSICS

Question No. **66 to 70** consist of **TWO (2)** marks for each correct response.

##### CHEMISTRY

Question No. **71 to 75** consist of **TWO (2)** marks for each correct response.

##### BIOLOGY

Question No. **76 to 80** consist of **TWO (2)** marks for each correct response.

## PART I

### One-Mark Questions

### MATHEMATICS

- 1 Let  $f(x)$  be a quadratic polynomial with  $f(2) = 10$  and  $f(-2) = -2$ . Then the coefficient of  $x$  in  $f(x)$  is  
A. 1                      B. 2                      C. 3                      D. 4
- 2 The square-root of  $\frac{(0.75)^3}{1-(0.75)} + (0.75 + (0.75)^2 + 1)$  is  
A. 1                      B. 2                      C. 3                      D. 4
- 3 The sides of a triangle are distinct positive integers in an arithmetic progression. If the smallest side is 10, the number of such triangles is  
A. 8                      B. 9                      C. 10                      D. infinitely many
- 4 If  $a, b, c, d$  are positive real numbers such that  $\frac{a}{3} = \frac{a+b}{4} = \frac{a+b+c}{5} = \frac{a+b+c+d}{6}$ , then  $\frac{a}{b+2c+3d}$  is  
A.  $\frac{1}{2}$                       B. 1                      C. 2                      D. not determinable

- 5 For  $\frac{2^2 + 4^2 + 6^2 + \dots + (2n)^2}{1^2 + 3^2 + 5^2 + \dots + (2n-1)^2}$  to exceed 1.01, the maximum value of  $n$  is

A. 99      B. 100      C. 101      D. 150

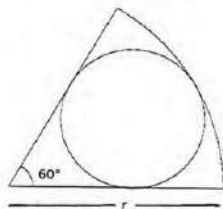
- 6 In triangle  $ABC$ , let  $AD$ ,  $BE$  and  $CF$  be the internal angle bisectors with  $D$ ,  $E$  and  $F$  on the sides  $BC$ ,  $CA$  and  $AB$  respectively. Suppose  $AD$ ,  $BE$  and  $CF$  concur at  $I$  and  $B$ ,  $D$ ,  $I$ ,  $F$  are concyclic, then  $\angle IFD$  has measure

A.  $15^\circ$       B.  $30^\circ$       C.  $45^\circ$       D. any value  $\leq 90^\circ$

- 7 A regular octagon is formed by cutting congruent isosceles right-angled triangles from the corners of a square. If the square has side-length 1, the side-length of the octagon is

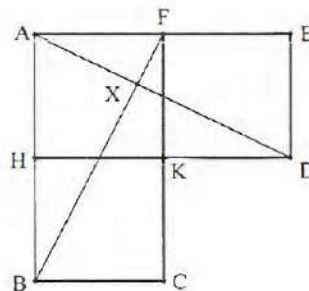
A.  $\frac{\sqrt{2}-1}{2}$       B.  $\sqrt{2}-1$       C.  $\frac{\sqrt{5}-1}{4}$       D.  $\frac{\sqrt{5}-1}{3}$

- 8 A circle is drawn in a sector of a larger circle of radius  $r$ , as shown in the adjacent figure. The smaller circle is tangent to the two bounding radii and the arc of the sector. The radius of the small circle is



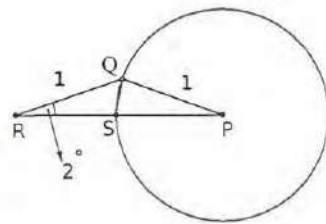
A.  $\frac{r}{2}$       B.  $\frac{r}{3}$       C.  $\frac{2\sqrt{3}r}{5}$       D.  $\frac{r}{\sqrt{2}}$

- 9 In the figure,  $AHKF$ ,  $FKDE$  and  $HBCK$  are unit squares;  $AD$  and  $BF$  intersect in  $X$ . Then the ratio of the areas of triangles  $AXF$  and  $ABF$  is



A.  $\frac{1}{4}$       B.  $\frac{1}{5}$       C.  $\frac{1}{6}$       D.  $\frac{1}{8}$

- 10 Suppose  $Q$  is a point on the circle with centre  $P$  and radius 1, as shown in the figure;  $R$  is a point outside the circle such that  $QR = 1$  and  $\angle QRP = 2^\circ$ . Let  $S$  be the point where the segment  $RP$  intersects the given circle. Then measure of  $\angle RQS$  equals



A.  $86^\circ$       B.  $87^\circ$       C.  $88^\circ$       D.  $89^\circ$

- 11 Observe that, at any instant, the minute and hour hands of a clock make two angles between them whose sum is  $360^\circ$ . At 6:15 the difference between these two angles is

A.  $165^\circ$       B.  $170^\circ$       C.  $175^\circ$       D.  $180^\circ$

- 12 Two workers  $A$  and  $B$  are engaged to do a piece of work. Working alone,  $A$  takes 8 hours more to complete the work than if both worked together. On the other hand, working alone,  $B$  would need  $4\frac{1}{2}$  hours more to complete the work than if both worked together. How much time would they take to complete the job working together?
- A. 4 hours    B. 5 hours    C. 6 hours    D. 7 hours
- 13 When a bucket is half full, the weight of the bucket and the water is 10 kg. When the bucket is two-thirds full, the total weight is 11 kg. What is the total weight, in kg, when the bucket is completely full?
- A. 12    B.  $12\frac{1}{2}$     C.  $12\frac{2}{3}$     D. 13
- 14 How many ordered pairs of  $(m, n)$  integers satisfy  $\frac{m}{12} = \frac{12}{n}$ ?
- A. 30    B. 15    C. 12    D. 10
- 15 Let  $S = \{1, 2, 3, \dots, 40\}$  and let  $A$  be a subset of  $S$  such that no two elements in  $A$  have their sum divisible by 5. What is the maximum number of elements possible in  $A$ ?
- A. 10    B. 13    C. 17    D. 20

## PHYSICS

- 16 A clay ball of mass  $m$  and speed  $v$  strikes another metal ball of same mass  $m$ , which is at rest. They stick together after collision. The kinetic energy of the system after collision is
- A.  $mv^2/2$     B.  $mv^2/4$     C.  $2mv^2$     D.  $mv^2$
- 17 A ball falls vertically downward and bounces off a horizontal floor. The speed of the ball just before reaching the floor ( $u_1$ ) is equal to the speed just after leaving contact with the floor ( $u_2$ );  $u_1 = u_2$ . The corresponding magnitudes of accelerations are denoted respectively by  $a_1$  and  $a_2$ . The air resistance during motion is proportional to speed and is not negligible. If  $g$  is acceleration due to gravity, then
- A.  $a_1 < a_2$     C.  $a_1 > a_2$   
 B.  $a_1 = a_2 \neq g$     D.  $a_1 = a_2 = g$
- 18 Which of the following statements is true about the flow of electrons in an electric circuit?
- A. Electrons always flow from lower to higher potential  
 B. Electrons always flow from higher to lower potential  
 C. Electrons flow from lower to higher potential except through power sources  
 D. Electrons flow from higher to lower potential, except through power sources

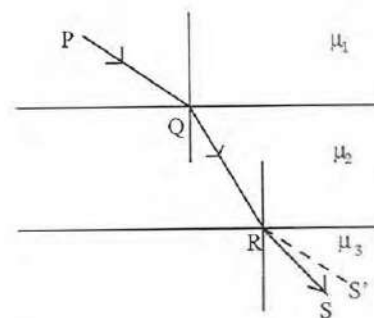
- 19 A boat crossing a river moves with a velocity  $v$  relative to still water. The river is flowing with a velocity  $w/2$  with respect to the bank. The angle with respect to the flow direction with which the boat should move to minimize the drift is

A.  $30^\circ$       B.  $60^\circ$       C.  $150^\circ$       D.  $120^\circ$

- 20 In the Arctic region hemispherical houses called Igloos are made of ice. It is possible to maintain a temperature inside an Igloo as high as  $20^\circ\text{C}$  because

A. ice has high thermal conductivity  
 B. ice has low thermal conductivity  
 C. ice has high specific heat  
 D. ice has higher density than water

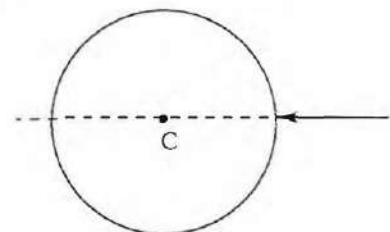
- 21 In the figure below,  $PQRS$  denotes the path followed by a ray of light as it travels through three media in succession. The absolute refractive indices of the media are  $\mu_1$ ,  $\mu_2$  and  $\mu_3$  respectively. (The line segment  $RS'$  in the figure is parallel to  $PQ$ ).



Then

A.  $\mu_1 > \mu_2 > \mu_3$       C.  $\mu_1 = \mu_3 < \mu_2$   
 B.  $\mu_1 < \mu_2 < \mu_3$       D.  $\mu_1 < \mu_3 < \mu_2$

- 22 A ray of white light is incident on a spherical water drop whose center is C as shown below. When observed from the opposite side, the emergent light





A. will be white and will emerge without deviating

B. will be internally reflected

C. will split into different colors such that the angles of deviation will be different for different colors

D. will split into different colors such that the angles of deviation will be the same for all colors

- 23 A convex lens of focal length 15 cm is placed in front of a plane mirror at a distance 25 cm from the mirror. Where on the optical axis and from the centre of the lens should a small object be placed such that the final image coincides with the object?

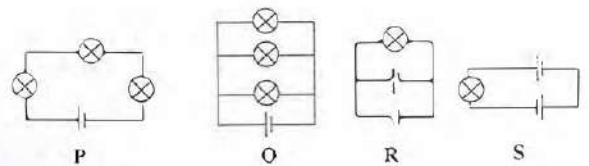
A. 15 cm and on the opposite side of the mirror

B. 15 cm and between the mirror and the lens

C. 7.5 cm and on the opposite side of the mirror

D. 7.5 cm and between the mirror and the lens

- 24 Following figures show different combinations of identical bulb(s) connected to identical battery(ies). Which option is correct regarding the total power dissipated in the circuit?



A.  $P < Q < R < S$

C.  $P < Q < R = S$

B.  $R < Q < P < S$

D.  $P < R < Q < S$

- 25 A circular metallic ring of radius  $R$  has a small gap of width  $d$ . The coefficient of thermal expansion of the metal is  $\alpha$  in appropriate units. If we increase the temperature of the ring by an amount  $\Delta T$ , then width of the gap

A. will increase by an amount  $d\alpha\Delta T$

B. will not change

C. will increase by an amount  $(2\pi R - d)\alpha\Delta T$

D. will decrease by an amount  $d\alpha\Delta T$

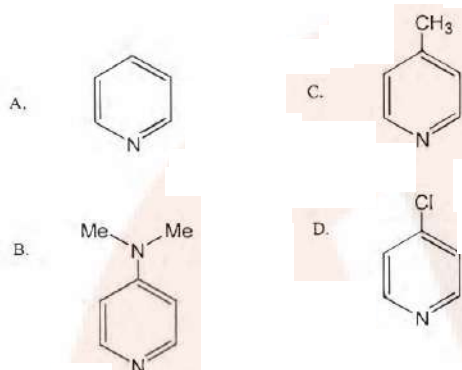
- 26 A girl holds a book of mass  $m$  against a vertical wall with a horizontal force  $F$  using her finger so that the book does not move. The frictional force on the book by the wall is

- A.  $F$  and along the finger but pointing towards the girl
- B.  $\mu F$  upwards where  $\mu$  is the coefficient of static friction
- C.  $mg$  and upwards
- D. equal and opposite to the resultant of  $F$  and  $mg$
- 27 A solid cube and a solid sphere both made of same material are completely submerged in water but to different depths. The sphere and the cube have same surface area. The buoyant force is
- A. greater for the cube than the sphere
- B. greater for the sphere than the cube
- C. same for the sphere and the cube
- D. greater for the object that is submerged deeper
- 28  $^{238}_{92}\text{U}$  atom disintegrates to  $^{214}_{84}\text{Po}$  with a half life of  $4.5 \times 10^9$  years by emitting six alpha particles and  $n$  electrons. Here  $n$  is
- A. 6                      B. 4                      C. 10                      D. 7
- 29 Which statement about the Rutherford model of the atom is NOT true?
- A. There is a positively charged center in an atom called the nucleus
- B. Nearly all the mass of an atom resides in the nucleus
- C. Size of the nucleus is comparable to the atom
- D. Electrons occupy the space surrounding the nucleus
- 30 A girl brings a positively charged rod near a thin neutral stream of water from a tap. She observes that the water stream bends towards her. Instead, if she were to bring a negatively charged rod near to the stream, it will
- A. bend in the same direction
- B. bend in the opposite direction
- C. not bend at all
- D. bend in the opposite direction above and below the rod

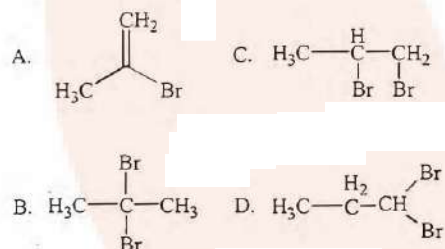
## CHEMISTRY

- 31 The weight of calcium oxide formed by burning 20 g of calcium in excess oxygen is  
 A. 36 g      B. 56 g      C. 28 g      D. 72 g
- 32 The major products in the reaction  $\text{Br}_2\text{CCHO} \xrightarrow{\text{NaOH}}$  are  
 A.  $\text{CHBr}_3 + \text{H}-\text{C}(=\text{O})\text{ONa}$       C.  $\text{NaOBr} + \text{H}-\text{C}(=\text{O})\text{H}$   
 B.  $\text{NaBr} + \text{H}-\text{C}(\text{Br})_2-\text{CH}(\text{OH})-\text{C}(=\text{O})\text{ONa}$       D.  $\text{H}-\text{C}(\text{Br})_2-\text{CH}(\text{OH})-\text{C}(=\text{O})\text{ONa}$
- 33 The number of electrons plus neutrons in  $^{40}_{19}\text{K}^+$  is  
 A. 38      B. 59      C. 39      D. 40
- 34 Among the following, the most basic oxide is  
 A.  $\text{Al}_2\text{O}_3$       C.  $\text{SiO}_2$   
 B.  $\text{P}_2\text{O}_5$       D.  $\text{Na}_2\text{O}$
- 35 By dissolving 0.35 mole of sodium chloride in water, 1.30 L of salt solution is obtained. The molarity of the resulting solution should be reported as  
 A. 0.3      B. 0.269      C. 0.27      D. 0.2692
- 36 Among the quantities, density ( $\rho$ ), temperature ( $T$ ), enthalpy ( $H$ ), heat capacity ( $C_p$ ), volume ( $V$ ) and pressure ( $P$ ), a set of intensive variables are  
 A. ( $\rho$ ,  $T$ ,  $H$ )      C. ( $V$ ,  $T$ ,  $C_p$ )  
 B. ( $H$ ,  $T$ ,  $V$ )      D. ( $\rho$ ,  $T$ ,  $P$ )
- 37 The value of 'x' in  $\text{KAl}(\text{SO}_4)_x \cdot 12\text{H}_2\text{O}$  is  
 A. 1      B. 2      C. 3      D. 4
- 38 Among the following substituted pyridines, the most basic compound is





- 39 The major product in the following reaction is  
 $\text{H}_3\text{C}-\text{C}\equiv\text{C}-\text{H} + \text{HBr (excess)}$



- 40 The major product in the following reaction at 25 °C is



- A.  $\text{CH}_3\text{CONHCH}_2\text{CH}_3$  C.  $\text{NH}_3^+\text{CH}_2\text{CH}_3\cdot\text{CH}_3\text{COO}^-$   
 B.  $\text{CH}_3\text{CH}=\text{NCH}_2\text{CH}_3$  D.  $\text{CH}_3\text{CON}=\text{CHCH}_3$

- 41 A reaction with reaction quotient  $Q_c$  and equilibrium constant  $K_c$ , will proceed in the direction of the products when

- A.  $Q_c = K_c$  C.  $Q_c > K_c$   
 B.  $Q_c < K_c$  D.  $Q_c = 0$

- 42 Acetylsalicylic acid is a pain killer and is commonly known as

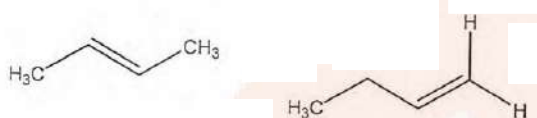
- A. paracetamol C. ibuprofen  
 B. aspirin D. penicillin

- 43 The molecule which does **not** exhibit strong hydrogen bonding is

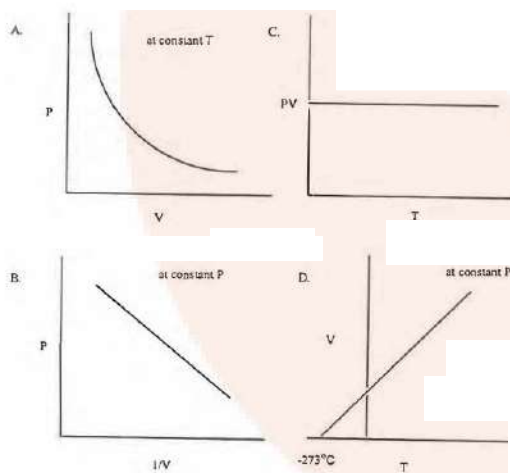
- A. methyl amine C. diethyl ether  
 B. acetic acid D. glucose

## KVPY

- 44 The following two compounds are



- A. geometrical isomers  
B. positional isomers  
C. functional group isomers  
D. optical isomers
- 45 The graph that does **not** represent the behaviour of an ideal gas is



## BIOLOGY

- 46 A smear of blood from a healthy individual is stained with a nuclear stain called hematoxylin and then observed under a light microscope. Which of the following cell type would be highest in number?
- A. neutrophils  
B. lymphocytes  
C. eosinophils  
D. monocytes
- 47 Which of the following biological phenomenon involves a bacteriophage?
- A. transformation  
B. conjugation  
C. translocation  
D. transduction
- 48 In which compartment of a cell does the process of glycolysis takes place?
- A. Golgi Complex  
B. cytoplasm  
C. mitochondria  
D. ribosomes
- 49 Huntington's disease is a disease of the
- A. nervous System  
B. circulatory system  
C. respiratory system  
D. excretory system

## KVPY

- 50 A cell will experience the highest level of endosmosis when it is kept in
- A. distilled water
  - B. sugar solution
  - C. salt solution
  - D. protein solution
- 51 When the leaf of the 'touch-me-not' (chui-mui, *Mimosa pudica*) plant is touched, the leaf droops because
- A. a nerve signal passes through the plant
  - B. the temperature of the plant increases
  - C. water is lost from the cells at the base of the leaf
  - D. the plant dies
- 52 If you are seeing mangroves around you, which part of India are you visiting?
- A. Western Ghats
  - B. Thar desert
  - C. Sunderbans
  - D. Himlayas
- 53 Myeloid tissue is a type of
- A. haematopoietic tissue
  - B. cartilage tissue
  - C. muscular tissue
  - D. areolar tissue
- 54 The heart of an amphibian is usually
- A. two chambered
  - B. three chambered
  - C. four Chambered
  - D. three and half chambered
- 55 Gigantism and acromegaly are due to defects in the function of the following gland:
- A. adrenals
  - B. thyroid
  - C. pancreas
  - D. pituitary
- 56 The pH of  $10^{-8}$  M HCl solution is,
- A. 8
  - B. close to 7
  - C. 1
  - D. 0
- 57 which one of the following organelles can synthesize some of its own proteins?
- A. lysosome
  - B. Golgi apparatus
  - C. vacuole
  - D. mitochondrion

- 58 Maltose is a polymer of
- one glucose and one fructose molecule
  - one glucose and one galactose molecule
  - two glucose molecules
  - two fructose molecules
- 59 The roots of some higher plants get associated with a fungal partner. The roots provide food to the fungus while the fungus supplies water to the roots. The structure so formed is known as
- lichen
  - anabaena
  - mycorrhiza
  - rhizobium
- 60 Prehistoric forms of life are found in rocks. The probability of finding fossils of more complex organisms
- increases from lower to upper strata
  - decreases from lower to upper strata
  - remains constant in each stratum
  - uncertain

## PART II

## Two-Mark Questions

## MATHEMATICS

- 61 Let  $a, b, c$  be positive integers such that  $\frac{a\sqrt{2}+b}{b\sqrt{2}+c}$  is a rational number, then which of the following is always an integer?
- $\frac{2a^2+b^2}{2b^2+c^2}$
  - $\frac{a^2+b^2-c^2}{a+b-c}$
  - $\frac{a^2+2b^2}{b^2+2c^2}$
  - $\frac{a^2+b^2+c^2}{a+c-b}$
- 62 The number of solutions  $(x, y, z)$  to the system of equations
- $$\begin{aligned}x + 2y + 4z &= 9, \\ 4yz + 2xz + xy &= 13, \\ xyz &= 3,\end{aligned}$$
- such that at least two of  $x, y, z$  are integers is
- 3
  - 5
  - 6
  - 4

## KVPY

- 63 In a triangle  $ABC$ , it is known that  $AB = AC$ . Suppose  $D$  is the mid-point of  $AC$  and  $BD = BC = 2$ . Then the area of the triangle  $ABC$  is

A. 2      B.  $2\sqrt{2}$       C.  $\sqrt{7}$       D.  $2\sqrt{7}$

- 64 A train leaves Pune at 7:30 am and reaches Mumbai at 11:30 am. Another train leaves Mumbai at 9:30 am and reaches Pune at 1:00 pm. Assuming that the two trains travel at constant speeds, at what time do the two trains cross each other?

A. 10:20 am      C. 11:30 am  
B. 10:26 am      D. data not sufficient

- 65 In the adjacent figures, which has the shortest path?

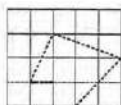


Fig 1.

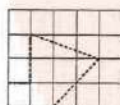


Fig 2

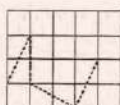


Fig 3

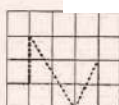
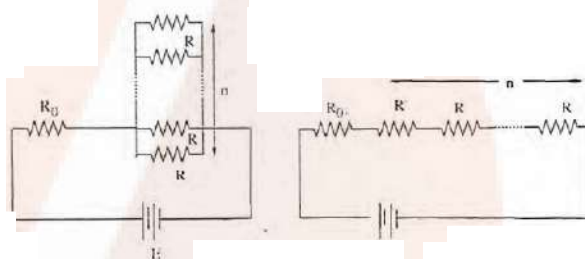


Fig 4

A. Fig 1      B. Fig 2      C. Fig 3      D. Fig. 4

## PHYSICS

- 66 In the circuit shown,  $n$  identical resistors  $R$  are connected in parallel ( $n > 1$ ) and the combination is connected in series to another resistor  $R_0$ . In the adjoining circuit  $n$  resistors of resistance  $R$  are all connected in series along with  $R_0$ .



The batteries in both circuits are identical and net power dissipated in the  $n$  resistors in both circuits is same. The ratio  $R_0/R$  is

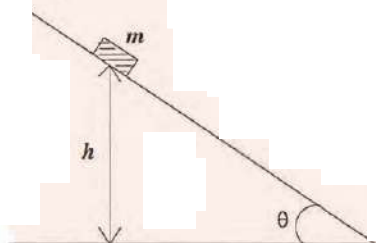
A. 1      B.  $n$       C.  $n^2$       D.  $1/n$

- 67 A firecracker is thrown with velocity of  $50 \text{ m.s}^{-1}$  in a direction which makes an angle of  $75^\circ$  with the vertical axis. At some point on its trajectory, the firecracker splits into two identical pieces in such a way that one piece falls 27 m far

from the shooting point. Assuming that all trajectories are contained in the same plane, how far will the other piece fall from the shooting point? (Take  $g = 10 \text{ m.s}^{-2}$  and neglect air resistance)

- A. 63 m or 144 m      C. 72 m or 99 m  
B. 28 m or 72 m      D. 63 m or 117 m

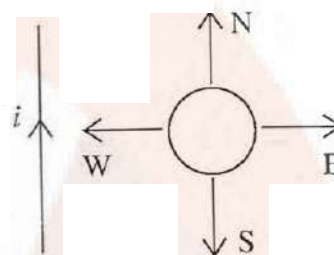
- 68 A block of mass  $m$  is sliding down an inclined plane with constant speed. At a certain instant  $t_0$ , its height above the ground is  $h$ . The coefficient of kinetic friction between the block and the plane is  $\mu$ . If the block reaches the ground at a later instant  $t_g$ , then the energy dissipated by friction in the time interval  $(t_g - t_0)$  is



- A.  $\mu mgh$       C.  $\mu mgh/\sin\theta$   
B.  $mgh$       D.  $\mu mgh/\cos\theta$

- 69 A circular loop of wire is in the same plane as an infinitely

long wire carrying a constant current  $i$ . Four possible motions of the loop are marked by N, E, W, and S as shown.



A clockwise current is induced in the loop when loop is pulled towards

- A. N      B. E      C. W      D. S

- 70 150 g of ice is mixed with 100 g of water at temperature  $80^\circ\text{C}$ . The latent heat of ice is  $80 \text{ cal/g}$  and the specific heat of water is  $1 \text{ cal/g}^\circ\text{C}$ . Assuming no heat loss to the environment, the amount of ice which does **not** melt is

- A. 100 g      B. 0 g      C. 150 g      D. 50 g



## CHEMISTRY

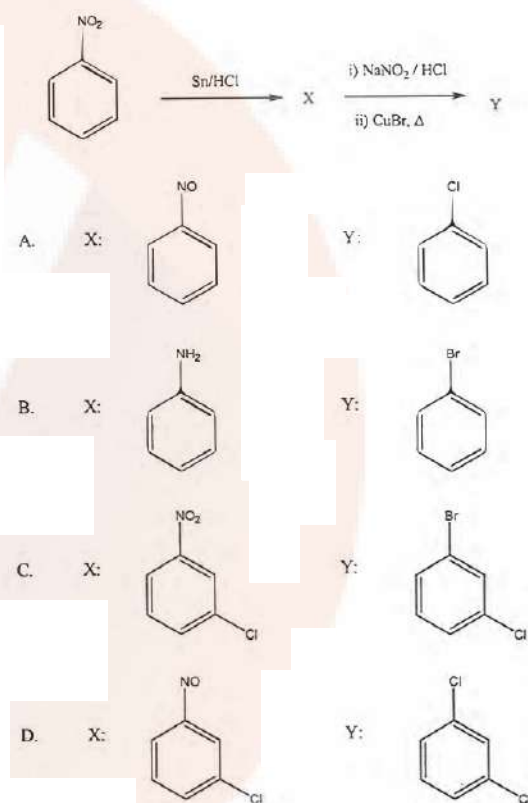
- 71 Upon fully dissolving 2.0 g of a metal in sulfuric acid, 6.8 g of the metal sulfate is formed. The equivalent weight of the metal is

A. 13.6 g      B. 20.0 g      C. 4.0 g      D. 10.0 g

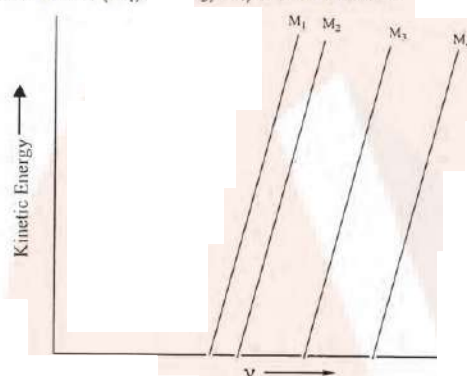
- 72 Upon mixing equal volumes of aqueous solutions of 0.1 M HCl and 0.2 M  $\text{H}_2\text{SO}_4$ , the concentration of  $\text{H}^+$  in the resulting solution is

A. 0.30 mol/L      C. 0.15 mol/L  
B. 0.25 mol/L      D. 0.10 mol/L

- 73 The products X and Y in the following reaction sequence are



- 74 A plot of the kinetic energy ( $\frac{1}{2}mv^2$ ) of ejected electrons as a function of the frequency ( $\nu$ ) of incident radiation for four alkali metals ( $M_1, M_2, M_3, M_4$ ) is shown below.



The alkali metals  $M_1, M_2, M_3$ , and  $M_4$  are, respectively

- A. Li, Na, K, and Rb  
 B. Rb, K, Na, and Li  
 C. Na, K, Li, and Rb  
 D. Rb, Li, Na, and K
- 75 The number of moles of  $Br_2$  produced when two moles of potassium permanganate are treated with excess potassium bromide in aqueous acid medium is
- A. 1  
 B. 3  
 C. 2  
 D. 4

## BIOLOGY

- 76 A baby is born with the normal number and distribution of rods, but no cones in his eyes. We would expect that the baby would be
- A. color blind  
 B. night blind  
 C. blind in both eyes  
 D. blind in one eye
- 77 In mammals, pleural membranes cover the lungs as well as insides of the rib cage. The pleural fluid in between the two membranes
- A. dissolves oxygen for transfer to the alveoli  
 B. dissolves  $CO_2$  for transfer to the blood  
 C. provides partial pressure  
 D. reduces the friction between the ribs and the lungs
- 78 At which phase of the cell cycle, DNA polymerase activity is at its highest?
- A. Gap 1 (G1)  
 B. Mitotic (M)  
 C. Synthetic (S)  
 D. Gap 2 (G2)
- 79 Usain Bolt, an Olympic runner, at the end of a 100 meter sprint, will have more of which of the following in his

muscles?

A. ATP

C. Lactic acid

B. Pyruvic acid

D. Carbon dioxide

- 80 Desert temperature often varies between 0 to 50 °C. The DNA polymerase isolated from a camel living in the desert will be able to synthesize DNA most efficiently at

A. 0 °C

B. 37 °C

C. 50 °C

D. 25 °C

# KVPY

# Answer Key

Answer Key for Stream - SA	
Question_No	Answer_Key
01	C
02	B
03	B
04	A
05	D
06	B
07	B
08	B
09	B
10	B
11	A
12	C
13	D
14	A
15	C
16	B
17	A
18	C
19	D
20	B
21	D
22	A
23	A
24	D
25	A
26	C
27	B
28	B
29	C
30	A
31	C
32	A
33	C
34	D
35	C
36	D
37	B
38	B
39	B
40	C
41	B
42	B
43	C
44	B
45	B
46	A
47	D
48	B
49	A
50	A

Answer Key for Stream - SA	
Question_No	Answer_Key
51	C
52	C
53	A
54	B
55	D
56	C
57	D
58	C
59	C
60	A
61	D
62	B
63	C
64	B
65	C
66	A
67	D
68	B
69	B
70	D
71	B
72	B
73	B
74	B
75	B
76	A
77	D
78	C
79	C
80	B